

Subject card

Subject name and code	Crystallography, PG_00020925								
Field of study	Nanotechnology								
Date of commencement of studies	October 2020		Academic year of realisation of subject			2021/2022			
Education level	first-cycle studies		Subject group			Obligatory subject group in the field of study			
						Subject group related to scientific research in the field of study			
Mode of study	Full-time studies		Mode of delivery			blended-learning			
Year of study	2		Language of instruction			Polish			
Semester of study	3		ECTS credits			5.0			
Learning profile	general academic profile		Assessment form			exam			
Conducting unit	Department of Solid	tment of Solid State Physics -> Faculty of Applied Physics and Mathematics							
Name and surname of lecturer (lecturers)	Subject supervisor	prof. dr hab. inż. Maria Gazda							
	Teachers		dr inż. Kacper Dzierzgowski						
		nż. Maria Gazo	a						
Lesson types and methods of instruction	Lesson type	Lecture	Tutorial	Laboratory	Projec	t	Seminar	SUM	
	Number of study hours	30.0	0.0	15.0	0.0		0.0	45	
	E-learning hours included: 30.0								
	Adresy na platformie eNauczanie: Krystalografia - Moodle ID: 15183 https://enauczanie.pg.edu.pl/moodle/course/view.php?id=15183								
Learning activity and number of study hours			articipation in didactic asses included in study lan		Participation in consultation hours		udy	SUM	
	Number of study hours	45		18.0		62.0		125	
Subject objectives	Gaining knowledge on the fundamentals of crystallography and relations between the crystal structure and properties of materials.								
Learning outcomes	Course outcome		Subject outcome			Method of verification			
	K6_W05					[SW1] Assessment of factual knowledge			
	K6_W06		· ·			[SW1] Assessment of factual knowledge			
	K6_U01					[SU1] Assessment of task fulfilment			
	K6_U04		Is able to perform some experiments, e.g. XRD, density determination etc			[SU1] Assessment of task fulfilment			
Subject contents	crystals, their charact	teristic features	equations; Symmetry of crystals, symmetry groups. •Examples of s and structural properties. Reciprocal lattice: definitione and ral studies. •Structural defects - their influence on the selected properties.						
	*Chemical bonds. * Crystal growth , Morphology of crystals.								
	Physical properties of crystals. Anisotropy.								
Prerequisites and co-requisites	No requirements								

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Assessment methods	Subject passing criteria	Passing threshold	Percentage of the final grade				
and criteria	Laboratory - average mark	51.0%	30.0%				
	Homework	30.0%	5.0%				
	test	51.0%	65.0%				
Recommended reading	Basic literature	Krystalografia, Bojarski i inni Any textbook on crystallography					
	Supplementary literature	No requirements					
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	eResources addresses	Krystalografia - Moodle ID: 15183 https://enauczanie.pg.edu.pl/moodle/course/view.php?id=15183					
Example issues/ example questions/ tasks being completed	1. How many atoms belong to the cel shown in the figure 1? What is the coordination numer of larger atom						
	2. Define Miller indices. Draw the planes (411), (002) and (100) in an orthorhombic crystal of cel parameters a = 4 Å , b = 2 Å i c = 8 Å . Give indices of the planes equivalent to (100).						
	3. Crystal has two mirror planes: one perpendicular to y and other to z. Determine points equivalent to ½ ¾ ½ What multiplicity has this point?						
	Calculate packing density for bcc structure.						
	5. What information may be obtained on the basis of X-ray diffraction investigation of a monocrystal?						
Work placement	Not applicable						

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