

GDAŃSK UNIVERSITY

Subject card

Subject name and code	Materials Science II, PG_00039868								
Field of study	Mechanical Engineering, Mechanical Engineering								
Date of commencement of studies	October 2020		Academic year of realisation of subject			2020/2021			
Education level	first-cycle studies		Subject group		Obligatory subject group in the field of study Subject group related to scientific research in the field of study				
Mode of study	Full-time studies		Mode of de	Mode of delivery			at the university		
Year of study	1		Language of instruction			Polish			
Semester of study	2		ECTS credits			2.0			
Learning profile	general academic profile		Assessmer	ssment form			assessment		
Conducting unit	Department of Materials Engineering and Bonding -> Faculty of Mechanical Engineering and Ship Technology								
Name and surname of lecturer (lecturers)	Subject supervisor	prof. dr hab. inż. Dionizy Czekaj							
	Teachers		prof. dr hab. inż. Dionizy Czekaj						
			dr inż. Magda Rościszewska						
			Dorota Rogala-Wielgus						
			mgr inż. Ewa Kozłowska						
		dr inż. Łukasz Pawłowski							
Lesson types and methods of instruction	Lesson type	Lecture	Tutorial	Laboratory	Projec	t	Seminar	SUM	
	Number of study hours	15.0	0.0	15.0	0.0		0.0	30	
	E-learning hours included: 0.0								
	Address on the e-learning platform: https://enauczanie.pg.edu.pl/moodle/course/view.php?id=12405								
	Adresy na platformie eNauczanie: Materiałoznawstwo II (M:31531W1) - Moodle ID: 12405								
	https://enauczanie.pg.edu.pl/moodle/course/view.php?id=12405								
	Additional information: Laboratory classes:								
	https://enauczanie.pg.edu.pl/moodle/course/view.php?id=12406								
Learning activity and number of study hours	Learning activity Participation ir classes include plan				Self-study		SUM		
	Number of study hours	30		5.0		15.0		50	
Subject objectives	The aim of the lecture	e is to introduce	e the students v	with selected is	ssues of	moder	n materials ei	ngineering	

Learning outcomes	Course outcome	Subject outcome	Method of verification				
	[K6_W08] possesses basic knowledge including the methodology of designing machine parts, mechanical devices, selection of construction materials, manufacturing and operation, with the lifetime cycle	The student can indicate the essential properties of metals, ceramics, polymers and composite structural materials influencing the properties of a device	[SW1] Assessment of factual knowledge				
	[K6_U10] is able to formulate the principles of selecting a material for a construction, ensuring the correct operation of a device	The student can compare the basic properties of materials in terms of their suitability for a specific application.	[SU2] Assessment of ability to analyse information				
	[K6_W03] possesses and is able to practically apply the knowledge on the construction, properties and testing methods of construction materials	The student knows the structure, basic properties and methods of characterization of structural materials.	[SW1] Assessment of factual knowledge				
Subject contents	Engineering materials. Properties of engineering materials: economic, mechanical, non-mechanical volume properties, surface properties, production and aesthetics properties; Prices and availability of materials. Periodic table of elements, structure of the PTE, law of periodicity. Electron configurations of elements, sizes of atoms and ions; ionization potentials; electron affinity; electronegative and electropositive elements; Binding of atoms in a crystal, Classification and characterization of bonds; Comparative characteristics of bonds; Basics of crystallography; Elements of the crystal lattice; Elementary cell; Crystallographic systems, Types of crystal lattices; Density of crystal lattice filling; Atomic structure - dense arrangement of atoms; Sequence of ABC.ABC layers and atomic structure - CCP (cubic close-packed); Sequence of layers AB.AB or AC.AC the densest hexagonal structure HCP (hexagonal close-packing). Material characteristics of construction materials; Definitions of stress, strain and modulus of elasticity; Hooke's law; Static tensile test; Elastic properties of the system of two atoms; Perfect durability; Crystal imperfections - defects in metals and ceramics: Principles of measuring the hardness of metals by Brinell, Vickers and Rockwell methods. Dynamic hardness measurements, Hardness and yield strength. Phase equilibrium systems. Phase balance diagram of the "Fe-C" system. Metals and their alloys. Cast iron. Non-ferrous alloys. Ceramics and glasses. Advanced ceramics. Polymers and composites. Technology processes; Manufacture of metal products, ceramics, glasses, polymers and composites. Electrical properties of materials. Electrical conductivity, semiconductors. Magnetic, optical and thermal properties of materials.						
Prerequisites and co-requisites							
Assessment methods	Subject passing criteria	Passing threshold	Percentage of the final grade				
and criteria	Laboratory classes	100.0%	50.0%				
	Colloquium from the lecture	51.0%	50.0%				
Recommended reading	Basic literature	 Michael Ashby, Hugh Shercliff and David Cebon, <i>Materials</i> <i>Engineering, Science, Processing and Design,</i> Elsevier Ltd, 2007 Michael Ashby, David Jones, <i>Engineering Materials 1, An</i> <i>Introduction to Properties, Applications, and Design,</i> Elsevier Ltd, 2012 Michael Ashby, David Jones, <i>Engineering Materials 2, An</i> <i>Introduction to Microstructures and Processing,</i> Elsevier Ltd, 2013 W. D. Callister, Jr., Materials science and engineering, an introduction, 7th ed., Wiley, 2007, ISBN 0-471-73696-1. 					
		 Michael Ashby, David Jones, E Introduction to Properties, Appl. 2012 Michael Ashby, David Jones, E Introduction to Microstructures W. D. Callister, Jr., Materials so 	ng and Design, Elsevier Ltd, 2007 ngineering Materials 1, An ications, and Design, Elsevier Ltd, ngineering Materials 2, An and Processing, Elsevier Ltd, 2013 ience and engineering, an				
	Supplementary literature	 Michael Ashby, David Jones, E Introduction to Properties, Appl. 2012 Michael Ashby, David Jones, E Introduction to Microstructures W. D. Callister, Jr., Materials so introduction, 7th ed., Wiley, 200 A.J. Moulson, , J.M. Herbert, El and Applications, Chapman and 	ng and Design, Elsevier Ltd, 2007 ngineering Materials 1, An ications, and Design, Elsevier Ltd, ngineering Materials 2, An and Processing, Elsevier Ltd, 2013 ience and engineering, an 17, ISBN 0-471-73696-1. ectroceramics, Materials Properties				
	Supplementary literature eResources addresses	 Michael Ashby, David Jones, E Introduction to Properties, Appl. 2012 Michael Ashby, David Jones, E Introduction to Microstructures W. D. Callister, Jr., Materials so introduction, 7th ed., Wiley, 200 A.J. Moulson, J.M. Herbert, El and Applications, Chapman and R. Pampuch, An Introduction to 	ng and Design, Elsevier Ltd, 2007 ngineering Materials 1, An ications, and Design, Elsevier Ltd, ngineering Materials 2, An and Processing, Elsevier Ltd, 2013 ience and engineering, an 17, ISBN 0-471-73696-1. ectroceramics, Materials Properties d Hall, 1990 Ceramics, Springer International				
Example issues/ example questions/ tasks being completed	eResources addresses 1. Periodic table of elements, law of 2. Binding of atoms in a crystal - th 3. Elements of the crystal lattice	 Michael Ashby, David Jones, E Introduction to Properties, Appl. 2012 Michael Ashby, David Jones, E Introduction to Microstructures W. D. Callister, Jr., Materials so introduction, 7th ed., Wiley, 200 A.J. Moulson, J.M. Herbert, El and Applications, Chapman and R. Pampuch, An Introduction to Publishing Switzerland, 2014 Materiałoznawstwo II (M:31531W1) https://enauczanie.pg.edu.pl/moodle 	ng and Design, Elsevier Ltd, 2007 ngineering Materials 1, An ications, and Design, Elsevier Ltd, ngineering Materials 2, An and Processing, Elsevier Ltd, 2013 ience and engineering, an 7, ISBN 0-471-73696-1. ectroceramics, Materials Properties d Hall, 1990 Ceramics, Springer International - Moodle ID: 12405 e/course/view.php?id=12405 terization of atomic bonds				