



## Subject card

Subject name and code	Technical Thermodynamics 2, PG_00042058						
Field of study	Power Engineering, Power Engineering, Power Engineering, Power Engineering, Power Engineering						
Date of commencement of studies	October 2020		Academic year of realisation of subject		2021/2022		
Education level	first-cycle studies		Subject group		Obligatory subject group in the field of study Subject group related to scientific research in the field of study		
Mode of study	Full-time studies		Mode of delivery		at the university		
Year of study	2		Language of instruction		English		
Semester of study	4		ECTS credits		3.0		
Learning profile	general academic profile		Assessment form		assessment		
Conducting unit	Department of Energy and Industrial Apparatus -> Faculty of Mechanical Engineering and Ship Technology						
Name and surname of lecturer (lecturers)	Subject supervisor		prof. dr hab. inż. Dariusz Mikielewicz				
	Teachers		dr inż. Marcin Jewartowski				
			mgr inż. Stanisław Głuch				
			dr hab. inż. Michał Klugmann				
			dr inż. Waldemar Targański				
			prof. dr hab. inż. Dariusz Mikielewicz				
			dr hab. inż. Jacek Barański				
Lesson types and methods of instruction	Lesson type	Lecture	Tutorial	Laboratory	Project	Seminar	SUM
	Number of study hours	15.0	0.0	15.0	0.0	0.0	30
	E-learning hours included: 0.0						
	Adresy na platformie eNauczanie:						
Learning activity and number of study hours	Learning activity	Participation in didactic classes included in study plan		Participation in consultation hours		Self-study	SUM
	Number of study hours	30		5.0		40.0	75
Subject objectives	Acquaintance of students with selected topics in thermodynamics such as heat transfer (4h), wet air (4h), Joule-Thompson effect (3h) and combustion (4h)						
Learning outcomes	Course outcome		Subject outcome		Method of verification		
	K6_U04		15 h of lectures supplemented with 15h of labs should be sufficient extension to the recommended topics in thermodynamics, which additionally supplemented with student's self work should enable successful pass of the examination.		[SU3] Assessment of ability to use knowledge gained from the subject		
	K6_W02		Suggested extension of the thermodynamics is directed onto particularly important operational issues in power engineering.		[SW1] Assessment of factual knowledge		

Subject contents	1. Heat transfer - fundamentals of mechanism of heat transfer, elementary problems in heat transfer, basics of heat exchangers  2. Joule-Thompson effect  3. Wet air - parameters characterising wet air, basic processes of wet air  4. Combustion - stoichiometry of combustion, fundamentals of combustion kinetics		
Prerequisites and co-requisites	Thermodynamics I, Fluid mechanics I		
Assessment methods and criteria	Subject passing criteria	Passing threshold	Percentage of the final grade
	lab classes	56.0%	0.0%
	written test	56.0%	100.0%
Recommended reading	Basic literature	<b>1. M.J. Moran, H.N. Shapiro, D.D. Boettner, M.B. Bailey, Fundamentals of Engineering Thermodynamics 8<sup>th</sup> Ed., Wiley, 2014</b>  <b>2. Y. Cengel, M. Boles, Thermodynamics An Engineering Approach, 8<sup>th</sup> Edition, Wiley, 2014</b>  <b>3. Incropera F.P., DeWitt D.P., Bergman T.L., Lavine A.S., Fundamentals Heat Mass Transfer, 7<sup>th</sup> Edition, 2011.</b>	
	Supplementary literature	1. Pudlik W.: Termodynamika. Wyd. PG, 2011.  2. Wiśniewski S., Wiśniewski T: Termodynamika techniczna. WNT, 2013.	
	eResources addresses		

Example issues/ example questions/ tasks being completed	<ol style="list-style-type: none"> <li>1. Present and discuss known mechanisms of heat transfer on the example of overall heat transfer through a multilayer wall separating two fluids with different temperatures.</li> <li>2. Define the thermal resistance due to conduction, convection and overall heat transfer.</li> <li>3. Discuss how to include the effect of fouling on overall thermal resistance.</li> <li>4. Definition of logarithmic mean temperature difference and temperature distribution in the parallel and counter-current heat exchangers.</li> <li>5. Define specific humidity and relative humidity. What is a difference?</li> <li>6. What is saturation temperature?</li> <li>7. Construct sample of psychrometric chart. What the lines represent?</li> <li>8. Describe graphically on a psychrometric chart all changes in the properties of air</li> <li>9. The dry-bulb and wet-bulb temperatures in a classroom are 24degC and 16 degC, respectively. Determine (at psychrometric chart) the humidity ratio, relative humidity and dew point at atmospheric pressure.</li> <li>10. Construction of Psychrometric Chart</li> <li>11. Design and operation of Linde-Hampson liquifier with representation of the process on a thermodynamic diagram.</li> <li>12. Definition of inversion point and inversion curve.</li> <li>13. What is the Joule-Thomson effect? The purpose and the coefficient of this effect.</li> <li>14. Definition of combustion process</li> <li>15. The stages of the solid fuel combustion</li> <li>16. The main characteristics of the flames</li> <li>17. Describe what is air excess number and how we can calculate it</li> <li>18. What is the difference between adiabatic flame temperature and real flame temperature</li> </ol>
Work placement	Not applicable