

Subject card

Subject name and code	Corrosion Processes, PG_00048916							
Field of study	Chemistry in Construction Engineering							
Date of commencement of studies	October 2020		Academic year of realisation of subject			2021/2022		
Education level	first-cycle studies		Subject group			Obligatory subject group in the field of study		
Mode of study	Full-time studies		Mode of delivery			at the university		
Year of study	2		Language of instruction			Polish		
Semester of study	4		ECTS credits			3.0		
Learning profile	general academic profile		Assessment form			assessment		
Conducting unit	Department of Electrochemistry, Corrosion and Materials Engineering -> Faculty of Chemistry							
Name and surname of lecturer (lecturers)	Subject supervisor		prof. dr hab. inż. Kazimierz Darowicki					
	Teachers		prof. dr hab. inż. Kazimierz Darowicki					
Lesson types and methods of instruction	Lesson type	Lecture	Tutorial	Laboratory	Projec	t	Seminar	SUM
	Number of study	15.0	0.0	15.0	0.0	0.0		30
	hours E-learning hours included: 0.0							
	Adresy na platformie eNauczanie:							
Learning activity	Learning activity Participation in		n didactic Participation in		Self-study SUM		SUM	
and number of study hours		classes includ		consultation hours		·		
	Number of study hours	30		5.0		40.0		75
Subject objectives	To acquaint students with the basic corrosion processes and types of corrosion							
Learning outcomes	Course outcome		Subject outcome			Method of verification		
	K6_K03		the student is able to solve problems related to corrosion of materials			[SK2] Assessment of progress of work		
	K6_W05		the student has a basic knowledge of corrosion processes			[SW1] Assessment of factual knowledge		
	K6_U09		the student is able to choose the type of protection for a given material			[SU4] Assessment of ability to use methods and tools		
Subject contents	Lecture: -Chemical thermodynamics: corrosion cells, E/pH diagrams, thermodynamic stability of water and its solutionsCorrosion processes kinetics: E=f(I) diagrams, corrosion processes controlTypes of corrosion: general, pitting, selective, intergranular, crevice, stress corrosion and stress corrosion cracking, corrosion-erosion, cavitationCorrosion occuring conditions (practical examples)Atlas of corrosion fatigue: description and visualization of fatigues. Laboratory: 1.Introduction and safety. 2.Temperature cell. 3.Oxygen concentration cell. 4.Galvanic cell. 5.Crevice corrosion. 6.Intergranular corrosion. 7.Selective corrosion of brass. 8.Pitting corrosion of steel. 9.Water 10.Reserved.							
Prerequisites and co-requisites	Chemical thermodynamics							
Assessment methods and criteria	Subject passing criteria		Passing threshold		Percentage of the final grade			
	Written exam		60.0%		50.0%			
	Laboratory		60.0%			50.0%		
Recommended reading	Basic literature		http://www.korozja.pl					
	Supplementary literature eResources addresses		No requirements					
Example issues/ example questions/ tasks being completed	Describe the work of a corrosion cell. Characterize the types of corrosion.							
Work placement	Not applicable							

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