



## Subject card

Subject name and code	Corrosion metallurgy, PG_00039724						
Field of study	Materials Engineering, Materials Engineering, Materials Engineering						
Date of commencement of studies	October 2021	Academic year of realisation of subject			2023/2024		
Education level	first-cycle studies	Subject group			Optional subject group Subject group related to scientific research in the field of study		
Mode of study	Full-time studies	Mode of delivery			at the university		
Year of study	3	Language of instruction			Polish		
Semester of study	6	ECTS credits			3.0		
Learning profile	general academic profile	Assessment form			assessment		
Conducting unit	Department of Corrosion and Electrochemistry -> Faculty of Chemistry						
Name and surname of lecturer (lecturers)	Subject supervisor	dr hab. inż. Michał Szociński					
	Teachers						
Lesson types and methods of instruction	Lesson type	Lecture	Tutorial	Laboratory	Project	Seminar	SUM
	Number of study hours	15.0	0.0	30.0	0.0	0.0	45
	E-learning hours included: 0.0						
	Address on the e-learning platform: <a href="https://enauczanie.pg.edu.pl/moodle/course/view.php?id=1082">https://enauczanie.pg.edu.pl/moodle/course/view.php?id=1082</a>						
Learning activity and number of study hours	Learning activity	Participation in didactic classes included in study plan	Participation in consultation hours	Self-study	SUM		
	Number of study hours	45	5.0	25.0	75		
Subject objectives	The aim of the subject is to familiarize students with: fundamental information concerning structure of metals and alloys, their properties with a special emphasis on susceptibility to corrosion, fundamental procedures connected with preparation of metallographic specimens and their evaluation.						
Learning outcomes	Course outcome	Subject outcome			Method of verification		
	K6_U02	student is able to identify corrosion defects at macro- and microscopic level			[SU4] Assessment of ability to use methods and tools [SU3] Assessment of ability to use knowledge gained from the subject		
	K6_W07	student can relate the information about structure of basic construction materials to their corrosion resistance			[SW1] Assessment of factual knowledge		
	K6_U01	student is able to prepare metallographic specimen and describe the microstructure of basic construction materials			[SU3] Assessment of ability to use knowledge gained from the subject [SU4] Assessment of ability to use methods and tools		
	K6_K01	student is able to identify the limitations connected with application of basic construction materials and to search for suitable material solutions			[SK5] Assessment of ability to solve problems that arise in practice		
Subject contents	Structure and properties of metals and alloys, general classification of metals and alloys, detailed structure of carbon steels, stainless steels, cast irons and non-ferrous alloys (copper and aluminum), preparation of metallographic specimens for microscopic analysis, preparation of samples for standard corrosion tests, procedure of macro- and microexamination of metallographic specimens, metallographic examination of corrosion damages of metals and alloys.						

Prerequisites and co-requisites	<p>Fundamentals of physics and physical chemistry.</p> <p>Fundamentals of materials science.</p> <p>Fundamentals of mechanics of materials.</p>											
Assessment methods and criteria	Subject passing criteria	Passing threshold	Percentage of the final grade									
	Tests and reports from laboratory exercises	60.0%	50.0%									
	Test from lectures	60.0%	50.0%									
Recommended reading	<table border="1"> <tr> <td data-bbox="448 445 794 960">Basic literature</td> <td colspan="2" data-bbox="794 445 1489 960"> <p>Dobrzański L.A. i in.: Metaloznawstwo i obróbka cieplna materiałów narzędziowych WNT. Warszawa 1990.</p> <p>Przybyłowicz K.: Metaloznawstwo (wyd. VIII). WNT. Warszawa 2007.</p> <p>Pr. Zb. [red. M. Głowacka]: Metaloznawstwo. Wyd. Politechniki Gdańskiej. Gdańsk 1996.</p> <p>Pr. Zb. [red. J. Hucińska]: Metaloznawstwo. Materiały do ćwiczeń laboratoryjnych. Pr. Zb. [red. M. Głowacka]: Metaloznawstwo. Wyd. Politechniki Gdańskiej. Gdańsk 1996.</p> <p>Pr. Zb. [red. J. Hucińska]: Metaloznawstwo. Materiały do ćwiczeń laboratoryjnych. Wyd. Politechniki Gdańskiej. Gdańsk 1995.</p> </td> </tr> <tr> <td data-bbox="448 960 794 1196">Supplementary literature</td> <td colspan="2" data-bbox="794 960 1489 1196"> <p>Pr. Zb. [red. M. Głowacka, A. Zieliński]: Podstawy materiałoznawstwa. Pr. Zb. [red. M. Głowacka]: Metaloznawstwo. Wyd. Politechniki Gdańskiej. Gdańsk 2014.</p> <p>Pr. Zb. [red. J. Hucińska]: Metaloznawstwo. Materiały do ćwiczeń laboratoryjnych. Wyd. Politechniki Gdańskiej. Gdańsk 2014.</p> </td> </tr> <tr> <td data-bbox="448 1196 794 1234">eResources addresses</td> <td colspan="2" data-bbox="794 1196 1489 1234">Adresy na platformie eNauczanie:</td> </tr> </table>			Basic literature	<p>Dobrzański L.A. i in.: Metaloznawstwo i obróbka cieplna materiałów narzędziowych WNT. Warszawa 1990.</p> <p>Przybyłowicz K.: Metaloznawstwo (wyd. VIII). WNT. Warszawa 2007.</p> <p>Pr. Zb. [red. M. Głowacka]: Metaloznawstwo. Wyd. Politechniki Gdańskiej. Gdańsk 1996.</p> <p>Pr. Zb. [red. J. Hucińska]: Metaloznawstwo. Materiały do ćwiczeń laboratoryjnych. Pr. Zb. [red. M. Głowacka]: Metaloznawstwo. Wyd. Politechniki Gdańskiej. Gdańsk 1996.</p> <p>Pr. Zb. [red. J. Hucińska]: Metaloznawstwo. Materiały do ćwiczeń laboratoryjnych. Wyd. Politechniki Gdańskiej. Gdańsk 1995.</p>		Supplementary literature	<p>Pr. Zb. [red. M. Głowacka, A. Zieliński]: Podstawy materiałoznawstwa. Pr. Zb. [red. M. Głowacka]: Metaloznawstwo. Wyd. Politechniki Gdańskiej. Gdańsk 2014.</p> <p>Pr. Zb. [red. J. Hucińska]: Metaloznawstwo. Materiały do ćwiczeń laboratoryjnych. Wyd. Politechniki Gdańskiej. Gdańsk 2014.</p>		eResources addresses	Adresy na platformie eNauczanie:	
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Example issues/ example questions/ tasks being completed	<ol style="list-style-type: none"> <li>1. Name and describe the types of unit cells forming the crystalline lattice of metals and alloys.</li> <li>2. Describe the point defects in crystalline lattice.</li> <li>3. How can electric conductivity of metals and alloys be determined?</li> </ol>											
Work placement	Not applicable											

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