

Subject card

Subject name and code	Thermodynamics , PG_00055381								
Field of study	Mechanical Engineering								
Date of commencement of studies	October 2021		Academic year of realisation of subject			2022/2023			
Education level	first-cycle studies		Subject group			Obligatory subject group in the field of study			
						Subject group related to scientific research in the field of study			
Mode of study	Full-time studies		Mode of delivery			at the	at the university		
Year of study	2		Language of instruction			Polish	Polish		
Semester of study	3		ECTS credits			7.0	7.0		
Learning profile	general academic profile		Assessme	nt form	exam				
Conducting unit	Department of Energy and Industrial Apparatus -> Faculty of Mechanical Engineering and Ship Technology								
Name and surname	Subject supervisor		dr hab. inż. Jan Wajs						
of lecturer (lecturers)	Teachers		dr inż. Marcin Jewartowski						
			mgr inż. Stanisław Głuch						
				mgr inż. Piotr Jasiukiewicz					
				mgr inż. Kamil Stasiak					
				dr hab. inż. Michał Klugmann					
			dr inż. Waldemar Targański						
			dr inż. Paweł Dąbrowski						
			dr hab. inż. Jan Wajs						
			ur nab. mz. J	an wajs	_				
Lesson types and methods	Lesson type	Lecture	Tutorial	Laboratory	Projec	:t	Seminar	SUM	
of instruction	Number of study hours	45.0	15.0	30.0	0.0		0.0	90	
	E-learning hours included: 0.0								
Learning activity and number of study hours	Learning activity	Participation in didact classes included in st plan		Participation in consultation hours		Self-study		SUM	
	Number of study 90 hours			7.0		78.0		175	
Subject objectives	Students acquire basic knowledge of thermodynamics in the dimension of theory and practice								

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and real gases. Gas laws, thermal and caloric equation of state. Characteristic processes of ideal g mixtures. Thermodynamic gas cycles. Entropy. The second law of thermodynamics and its consequing list of superstanding examples of second law of thermodynamics and its consequing examples. Properties of mono-component saturated steam. Properties of superstanding examples and moist gas mixtures and moist gas consequences.	ty to						
including classic mechanics, acoustics, optics, electricity and magnetism, shows knowledge of the elements of quantum physics of energy conversion or transformation occurring in them. KK6_U06 is able to use mathematical and physical models for analysing the processes and phenomena occurring in mechanical devices within the range of material strength, thermodynamics and fluid mechanics K6_W09 possesses basic knowledge within the range of thermodynamics and fluid mechanics.	ty to						
mathematical and physical models for analysing the processes and phenomena occurring in mechanical devices within the range of material strength, thermodynamics and fluid mechanics [K6 W09] possesses basic knowledge within the range of thermodynamics and fluid mechanics. [K6 W09] possesses basic knowledge within the range of thermodynamics and fluid mechanics or thermodynamics and fluid mechanics, construction and operation of heat generating devices, process equipment, including renewable energy sources, cooling and air conditioning Subject contents Subject contents Calculate Calculate							
knowledge within the range of thermodynamics and fluid mechanics, construction and operation of heat generating devices, process equipment, including renewable energy sources, cooling and air conditioning Thermodynamics, 1st and 2nd Law of Thermodynamics and equations of state of gases. Student analyzes the typical processes of ideal gas and steam, gas or steam cycles and heat transfer mechanisms. Student uses theory of the moist gases and explains air treatment processes for air conditioning. Student uses basic concepts of the thermodynamics of combustion. Student performs the measurements on an experimental setup, makes necessary calculations and presents the results in the form of tables and graphs. Student is able to analyze energy balance of various thermal machines. Subject contents LECTURE: Basic concepts. The first law of thermodynamics. Ideal gas model. Properties of ideal, s and real gases. Gas laws, thermal and caloric equation of state. Characteristic processes of ideal g mixtures. Thermodynamic gas cycles. Entropy. The second law of thermodynamics and its consequence is superficient of the processes. Properties of regireration. Basics of compresses. Fundamentals of refrigeration. Basics of compresses is a processes. Fundamentals of refrigeration. Basics of compresses.	rual						
and real gases. Gas laws, thermal and caloric equation of state. Characteristic processes of ideal g mixtures. Thermodynamic gas cycles. Entropy. The second law of thermodynamics and its consequ Isobaric evaporation process. Properties of mono-component saturated steam. Properties of superl steam. Characteristic processes of steam. Thermodynamic steam cycles. Gas mixtures and moist g Mollier diagram and the basic moist air processes. Fundamentals of refrigeration. Basics of compre							
	LECTURE: Basic concepts. The first law of thermodynamics. Ideal gas model. Properties of ideal, semi-ideal and real gases. Gas laws, thermal and caloric equation of state. Characteristic processes of ideal gas. Gas mixtures. Thermodynamic gas cycles. Entropy. The second law of thermodynamics and its consequences. Isobaric evaporation process. Properties of mono-component saturated steam. Properties of superheated steam. Characteristic processes of steam. Thermodynamic steam cycles. Gas mixtures and moist gases. Mollier diagram and the basic moist air processes. Fundamentals of refrigeration. Basics of compressor and sorption heat pumps. Elements of combustion thermodynamics.						
EXERCISES: Simple conversion of energy, heat, work. The balances of power of open or closed thermodynamics systems. State and functions of state of ideal and semi-ideal gases and gas mixtu Characteristic processes of gases. Gas thermodynamic cycles. Characteristic changes of steam. Calculations thermodynamic steam cycles.	es.						
LABORATORIES: Measurements of thermodynamic parameters: temperature and pressure. Determined of mass flow rate. Determination of air and water enthalpy. Energy balance of heat pump. Testing of refrigerating unit. Determination of calorific value of solid and gas fuels. Energy balance of piston enterting of the compressor.	f the						
Prerequisites And co-requisites Knowledge from course of physics and mathematics.							
Assessment methods Subject passing criteria Passing threshold Percentage of the final							
and criteria Laboratory reports 100.0% 20.0%							
Middterm colloquiums 56.0% 30.0%							
Written exam 56.0% 50.0%							

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Recommended reading	Basic literature	 Y. Cengel, M. Boles, Thermodynamics An Engineering Approach, 8th Edition, Wiley, 2014. M.J. Moran, H.N. Shapiro, D.D. Boettner, M.B. Bailey, Fundamentals of Engineering Thermodynamics 8th Ed., Wiley, 2014. R. Mayhew, Engineering thermodynamics/Work & Heat Transfer. Wiley & Sons Inc. 1993, USA. 			
	Supplementary literature	no requirements			
	eResources addresses	Adresy na platformie eNauczanie:			
Example issues/ example questions/ tasks being completed	Present equations of first law of thermodynamics. Describe Carnot Cycle. Describe Rankine / Otto / Diesel cycle. Present definitions of second law of thermodynamics. Operational principle of compressor heat pumps. Heating and humidification of air.				
Work placement	Not applicable				

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