



## Subject card

Subject name and code	Basic of biochemistry, PG_00057693						
Field of study	Green Technologies						
Date of commencement of studies	October 2022	Academic year of realisation of subject			2024/2025		
Education level	first-cycle studies	Subject group			Obligatory subject group in the field of study Subject group related to scientific research in the field of study		
Mode of study	Full-time studies	Mode of delivery			at the university		
Year of study	3	Language of instruction			Polish		
Semester of study	6	ECTS credits			3.0		
Learning profile	general academic profile	Assessment form			assessment		
Conducting unit	Department of Chemistry, Technology and Biotechnology of Food -> Faculty of Chemistry						
Name and surname of lecturer (lecturers)	Subject supervisor	prof. dr hab. inż. Agnieszka Bartoszek-Pączkowska					
	Teachers	prof. dr hab. inż. Agnieszka Bartoszek-Pączkowska dr inż. Izabela Koss-Mikołajczyk					
Lesson types and methods of instruction	Lesson type	Lecture	Tutorial	Laboratory	Project	Seminar	SUM
	Number of study hours	30.0	0.0	15.0	0.0	0.0	45
	E-learning hours included: 0.0						
Learning activity and number of study hours	Learning activity	Participation in didactic classes included in study plan	Participation in consultation hours		Self-study	SUM	
	Number of study hours	45	2.0		28.0	75	
Subject objectives	Lectures in Biochemistry aim at making the students familiar with basic biochemical concepts and functioning of organic matter. The particular emphasis is placed on chemical aspects of processes underlying function of living things. The interrelations between toxicity of some environmental factors resulting from human activities and their impact on living things are indicated. The lectures are supplemented by laboratory exercises aimed at explaining the students the ways of observing phenomena that have been raised during lectures.						

Learning outcomes	Course outcome	Subject outcome	Method of verification
	[K6_U03] is able to use information and communication technologies relevant to the common tasks of engineering, is able to use known methods and mathematical-physical models to describe and explain phenomena and chemical processes	Student measures the kinetics of the enzymatic reaction and calculates enzyme activity based on measurements; Student differentiates toxic from nontoxic compounds in the culture of human cells; Student performs DNA electrophoresis and analyzes changes of DNA integrity induced by exposure to genotoxins; Student carries out spectrophotometric measurements and on this basis estimates the binding of ions to cell wall	[SU4] Assessment of ability to use methods and tools [SU2] Assessment of ability to analyse information
	[K6_U04] capable of formulating and solving design tasks in the field of environmental technology to recognize their non-technical aspects, including environmental, economic and legal. Is capable of applying the principles of occupational health and safety. Is able to make initial assessment of engineering solutions and actions	Student exhibits the basic knowledge of biochemistry needed to understand biochemical phenomena and processes utilized in environmental technologies.	[SU3] Assessment of ability to use knowledge gained from the subject [SU2] Assessment of ability to analyse information
	[K6_W04] is aware of the importance of environmental protection and has a basic knowledge of chemical and biological threats to the environment, with particular emphasis on anthropogenic factors, has a basic knowledge of knowledge of the principles of sustainable development as well as national and European environmental management conditions.	Student lists cellular structures and explains their functions; Student describes all levels of basic protein and nucleic acids structures; Students tells the role of lipids and saccharides; Student describes processes of gene expression; Student explains cellular processes of energy production	[SW1] Assessment of factual knowledge
Subject contents	Lectures in Biochemistry aim at making the students familiar with basic biochemical concepts and functioning of organic matter. The particular emphasis is placed on chemical aspects of processes underlying function of living things. The interrelations between toxicity of some environmental factors resulting from human activities and their impact on living things are indicated. The lectures include the topics: 1. Unity of matter in the light of Solar system evolution. Structure and function of prokaryotic and eukaryotic cells. Organelles, including their function in protection against environmental challenges. 2. Proteins. Structure and biological functions. Selected examples: oxygen transporting proteins, physiological role and environmental risks; antibodies, physiological role and application in analytical chemistry. 3. Enzymes as biocatalysts. Examples of catalytic mechanisms. Regulation of enzymatic activity. Inhibitors and poisons. 4. Lipids. Structure and functions. Structure of biological membranes and its protective role. Environmental factors that may damage membranes. 5. Polysaccharides. Occurrence and function. 6. The structure of nucleic acids. DNA structure determines its function. Transfer and use of genetic information. DNA damage by genotoxic factors. 7. Metabolism. Basic features. Organisation and interrelations of metabolic pathways. Biodegradation of saccharides and lipids as a source of metabolic energy and precursors for the reactions. The lectures are supplemented by laboratory exercises aimed at explaining the students the ways of observing phenomena that have been raised during lectures as well as methods used for the assessments of environmental risks for living things.		
Prerequisites and co-requisites	The acquaintance with basics of organic chemistry, physical chemistry and laboratory techniques.		
Assessment methods and criteria	Subject passing criteria	Passing threshold	Percentage of the final grade
	30% laboratory	80.0%	30.0%
	70% lecture	50.0%	70.0%
Recommended reading	Basic literature	Biochemia, E. Bańkowski, PWN, Warszawa, 2009 Podstawy biologii komórki. Wprowadzenie do biologii molekularnej B. Alberts, D. Bray, A. Johnson, J. Lewis, M. Raff, K. Roberts, P. Walter	
	Supplementary literature	Zarys Ekotoksykologii, red. Namieśnik J., Jaśkowski J., EKO- Pharma, Gdańsk 1994 Podstawy ekotoksykologii, Z. Zakrzewski, 1996	
	eResources addresses	Adresy na platformie eNauczanie:	
Example issues/ example questions/ tasks being completed	1. Tell the difference between competitive and noncompetitive enzymes and enzymatic poisons. 2. What role in the cell play DNA and RNA? 3. Describe the differences in the structure and function of starch, glycogen and cellulose. 4. Explain the basic assumptions of chemiosmosis that integrates flow of electrons and ATP synthesis during oxidative phosphorylation. 5. Describe briefly the replication of the leading DNA strand (synthesized in a continuous way) in a prokaryotic cell. 6. Write with full formulas the indicated fragment of Krebs cycle. Add the enzymes involved and all necessary reactants. Is any of the enzymes regulated? What is the mechanism behind this regulation.		
Work placement	Not applicable		

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