

## Subject card

| Subject name and code  | CORROSION, PG_00036512   |   |   |                                     |                               |  |           |     |
|--|--|---|---|-------------------------------------|-------------------------------|--|-----------|-----|
| Field of study   | Chemistry  |   |   |                                     |                               |  |           |     |
| Date of commencement of studies                                | October 2022   |   | Academic year of realisation of subject   |                                     |                               | 2024/2025                                      |           |     |
| Education level  | first-cycle studies  |   | Subject group   |                                     |                               | Obligatory subject group in the field of study |           |     |
| Mode of study  | Full-time studies  |   | Mode of delivery  |                                     |                               | at the university                              |           |     |
| Year of study  | 3  |   | Language of instruction   |                                     |                               | Polish   |           |     |
| Semester of study  | 5  |   | ECTS credits  |                                     |                               | 2.0  |           |     |
| Learning profile   | general academic profile   |   | Assessment form   |                                     |                               | assessment                                     |           |     |
| Conducting unit  | Department of Electro  | ochemistry, Co                              | rrosion and Ma  | ring -> I                           | Faculty of Chemistry          |  |           |     |
| Name and surname   | Subject supervisor   |   | prof. dr hab. inż. Kazimierz Darowicki  |                                     |                               |  |           |     |
| of lecturer (lecturers)  | Teachers   |   |   |                                     |                               |  |           |     |
| Lesson types and methods of instruction                        | Lesson type  | Lecture                                     | Tutorial  | Laboratory                          | Project                       | ct Seminar                                     |           | SUM |
|  | Number of study hours  | 15.0  | 0.0   | 0.0                                 | 0.0                           | 0.0  |           | 15  |
|  | E-learning hours included: 0.0   |   |   |                                     |                               |  |           |     |
| Learning activity and number of study hours                    | Learning activity  | Participation in<br>classes include<br>plan |   | Participation in consultation hours |                               | Self-study                                     |           | SUM |
|  | Number of study hours  | 15  |   | 5.0                                 |                               | 30.0   |           | 50  |
| Subject objectives   | bases of corrosion and types of corrosion damage   |   |   |                                     |                               |  |           |     |
| Learning outcomes  | Course outcome Subject outcome Method of verification  |   |   |                                     |                               |  | ification |     |
| Subject contents   | -Energy band theory of metals, semiconductors and dielectricsElectric, magnetics and thermal properties of metalsTypes of crystal lattice of solidsSolid solutionsAlloys and phase transitions, heat treatment Iron-carbon phase diagramClassifications of steels and cast ironsBasics of thermodynamics and chemical kineticsTypes of corrosion failuresCorrosion: general, selective, intergranular, pitting, crevice Stress corrosion cracking and corrosion fatigue. |   |   |                                     |                               |  |           |     |
| Prerequisites and co-requisites                                | Chemical bonds, theory of solutions, chemical thermodynamics, basics of quantum chemistry.   |   |   |                                     |                               |  |           |     |
| Assessment methods and criteria                                | Subject passing criteria   |   | Passing threshold   |                                     | Percentage of the final grade |  |           |     |
|  | test   |   | 60.0%   |                                     |                               | 100.0%   |           |     |
| Recommended reading  | Basic literature   |   | Ch.A.Wert, R.M. Thomson, Fizyka ciała stałego, PWN Warszawa 1974 J. Dereń, J. Chaber, R. Pampuch, Chemia ciała stałego, PWN Warszawa 1977 L.L. Shreier, R.A. Barman, G.T. Burstein, Corrosion, Butterworth, London 1994 P.A. Schweitzer, Fundamentals of Metallic Corrosion, CRC Press, London 2007 |                                     |                               |  |           |     |
|  | Supplementary literature   |   | No requirements   |                                     |                               |  |           |     |
|  | eResources addresses   |   | Adresy na platformie eNauczanie:  |                                     |                               |  |           |     |
| Example issues/<br>example questions/<br>tasks being completed | Make a Pourbaix diagram for zincDerive the Butler-Volmer equationCharacterize the electric double layer model  |   |   |                                     |                               |  |           |     |
| Work placement   | Not applicable   |   |   |                                     |                               |  |           |     |

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