

GDAŃSK UNIVERSITY OF TECHNOLOGY

Subject card

Subject name and code	Corrosion Processes, PG_00048916								
Field of study	Chemistry in Construction Engineering								
Date of commencement of studies	October 2022		Academic year of realisation of subject			2023/2024			
Education level	first-cycle studies		Subject group			Obligatory subject group in the field of study			
Mode of study	Full-time studies		Mode of delivery			at the university			
Year of study	2		Language of instruction			Polish			
Semester of study	4		ECTS credits			3.0			
Learning profile	general academic profile		Assessment form			assessment			
Conducting unit	Department of Electr	rrosion and Materials Engineering ->			Faculty of Chemistry				
Name and surname of lecturer (lecturers)	Subject supervisor		prof. dr hab. inż. Kazimierz Darowicki						
	Teachers								
Lesson types and methods of instruction	Lesson type	Lecture	Tutorial	Laboratory	Projec	t	Seminar	SUM	
	Number of study hours	15.0	0.0	15.0	0.0		0.0	30	
	E-learning hours included: 0.0								
Learning activity and number of study hours	Learning activity	Participation i classes includ plan		Participation in consultation hours		Self-study		SUM	
	Number of study hours	study 30		5.0		40.0		75	
Subject objectives	To acquaint students with the basic corrosion processes and types of corrosion								
Learning outcomes	Course out	Subject outcome			Method of verification				
	К6_К03		the student is able to solve problems related to corrosion of materials			[SK2] Assessment of progress of work			
	K6_W05					[SW1] Assessment of factual knowledge			
	K6_U09		the student is able to choose the type of protection for a given material			[SU4] Assessment of ability to use methods and tools			
Subject contents	Lecture: -Chemical thermodynamics: corrosion cells, E/pH diagrams, thermodynamic stability of water and its solutionsCorrosion processes kinetics: E=f(I) diagrams, corrosion processes controlTypes of corrosion: general, pitting, selective, intergranular, crevice, stress corrosion and stress corrosion cracking, corrosion-erosion, cavitationCorrosion occuring conditions (practical examples)Atlas of corrosion fatigue: description and visualization of fatigues. Laboratory: 1.Introduction and safety. 2.Temperature cell. 3.Oxygen concentration cell. 4.Galvanic cell. 5.Crevice corrosion. 6.Intergranular corrosion. 7.Selective corrosion of brass. 8.Pitting corrosion of steel. 9.Water 10.Reserved.								
Prerequisites and co-requisites	Chemical thermodyn	amics							
Assessment methods	Subject passing criteria		Passing threshold		Percentage of the final grade				
and criteria	Written exam		60.0%		50.0%				
	Laboratory		60.0%				50.0%		
Recommended reading	Basic literature		http://www.korozja.pl						
	Supplementary literature		No requirements						
	eResources addresses Adresy na platformie eNauczanie:								
Example issues/ example questions/ tasks being completed	Describe the work of a corrosion cell. Characterize the types of corrosion.								
Work placement	Not applicable								