

## Subject card

Subject name and code	Materials Science - quantum particle approach, PG_00052037								
Field of study	Nanotechnology								
Date of commencement of studies	October 2022		Academic year of realisation of subject			2023/2024			
Education level	second-cycle studies		Subject group			Optional subject group Subject group related to scientific research in the field of study			
Mode of study	Full-time studies		Mode of delivery			at the university			
Year of study	2		Language of instruction			English English			
Semester of study	3		ECTS credits			6.0			
Learning profile	general academic profile		Assessment form			asses	assessment		
Conducting unit	Zakład fizyki układów nieuporządkowanych -> Instytut Nanotechnologii i Inżynierii Materiałowej -> Faculty of Applied Physics and Mathematics								
Name and surname	Subject supervisor		dr hab. Macie						
of lecturer (lecturers)	Teachers		dr hab. Macie						
Lesson types and methods of instruction	Lesson type	Lecture	Tutorial	Laboratory	Projec	ct Seminar		SUM	
	Number of study hours	30.0	0.0	45.0	0.0	0.0		75	
	E-learning hours included: 0.0								
	Additional information: In-person, online - if necessary.								
Learning activity and number of study hours				Participation in consultation hours		Self-study		SUM	
	Number of study hours	75		5.0		70.0		150	
Subject objectives	Purposes:  1.Pass the knowledge on application of quantum methods for issues of change of electronic structure present in molecules and crystals.  2. Teaching axioms of quantum mechanics and their applications.  3. Teaching of commonly utilized quantum methods based on wave functions and electron densities: HF, CI, MCSCF, CC, MPn, DFT.  4. Teaching of utilization of commonly applied basis sets in quantum calculations								

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Learning outcomes	Course outcome	Subject outcome	Method of verification			
	K7_U03	Student can maintain software for quantum calculations on many-processor computer along with advanced visualization software which is used for displaying partial results and for building the structures.	[SU1] Assessment of task fulfilment [SU2] Assessment of ability to analyse information [SU3] Assessment of ability to use knowledge gained from the subject [SU4] Assessment of ability to use methods and tools			
	K7_U06	Sudent can for himself, solve the problems given by the teacher by means of software for quantum calculations.	[SU5] Assessment of ability to present the results of task [SU4] Assessment of ability to use methods and tools [SU3] Assessment of ability to use knowledge gained from the subject			
	K7_W02	Student has a deep knowledge on quantum methods which can be applied to computations of electronic structure's change of chemical systems which build nanostructures, as well as on capabilities and limitations of such method.	[SW1] Assessment of factual knowledge [SW2] Assessment of knowledge contained in presentation			
	K7_W05	Student has deep knowledge on: operators' definitions in spherical coordinates, matrix representation and diagonalization and orthogonalization methods, normalization of wavefunctions, Slater and Gaussian basis sets, Hartree-Fock-Roothan approach, atomic and molecular orbitals, Cl methods, perturbation methods, methods of finding and characterizing stationary states.	[SW3] Assessment of knowledge contained in written work and projects [SW2] Assessment of knowledge contained in presentation			
Subject contents	application of quantum methods in cases of solving of electronic-structure change for systems of molecules and crystals, axioms of quantum mechanics and their applications, commonly utilized quantum methods based on wave functions and electron densities: HF, CI, MCSCF, MPn, CC, DFT, basis sets.					
Prerequisites and co-requisites	Base knowledge on physics, mathematics and chemistry.					
Assessment methods	Subject passing criteria	Passing threshold	Percentage of the final grade			
and criteria	sprawozdanie	51.0%	50.0%			
	exam	51.0%	50.0%			
Recommended reading	Basic literature	<ol> <li>Frank Jensen, Introduction to Computational Chemistry, Wydawnictwo Wiley, 2007,</li> <li>C. J. Ballhausen, H. B. Gray, Molecular Orbital Theory, Wydawnictwo W. A. Benjamin Inc. 1964,</li> </ol>				
	Supplementary literature	Yung-Kuo Lim, Problems and Solutions on Quantum Mechanics, Wydawnictwo World Scientific, 2005,				
	eResources addresses	Podstawowe https://enauczanie.pg.edu.pl/moodle/course/view.php?id=32861 - Teacher's materials. Adresy na platformie eNauczanie: Materials science. Quantum particle approach. 2023 Moodle ID:				
		32861 https://enauczanie.pg.edu.pl/moodle/course/view.php?id=32861				

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Example issues/ example questions/ tasks being completed	Calculate given commutators in cartesian and spherical coordinates,
	2. Normalize given wavefunctions,
	3. Orthogonalize given basis stes,
	Calculate energies of given electron configurations
	5. What atomic and molecular orbitals should be taken into account in the case of given electron states of given molecules and given spins.
	Calculate CI coefficients for hydrogen molecule for given electron configuration.
Work placement	Not applicable

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