



## Subject card

Subject name and code	Corrosion measurements, PG_00060320						
Field of study	Materials Engineering						
Date of commencement of studies	October 2023		Academic year of realisation of subject		2025/2026		
Education level	first-cycle studies		Subject group		Optional subject group Subject group related to scientific research in the field of study		
Mode of study	Full-time studies		Mode of delivery		at the university		
Year of study	3		Language of instruction		Polish		
Semester of study	5		ECTS credits		4.0		
Learning profile	general academic profile		Assessment form		exam		
Conducting unit	Department of Corrosion and Electrochemistry -> Faculty of Chemistry -> Faculties of Gdańsk University of Technology						
Name and surname of lecturer (lecturers)	Subject supervisor		dr hab. inż. Stefan Krakowiak				
	Teachers		dr hab. inż. Stefan Krakowiak  dr hab. inż. Michał Szociński  dr inż. Łukasz Gawel  dr hab. inż. Andrzej Miszczyk				
Lesson types	Lesson type	Lecture	Tutorial	Laboratory	Project	Seminar	SUM
	Number of study hours	15.0	0.0	30.0	0.0	0.0	45
	E-learning hours included: 0.0						
	eNauczanie source address: <a href="https://enauczanie.pg.edu.pl/moodle/course/view.php?id=46173">https://enauczanie.pg.edu.pl/moodle/course/view.php?id=46173</a> Moodle ID: 46173 Pomiar korozyjne <a href="https://enauczanie.pg.edu.pl/moodle/course/view.php?id=46173">https://enauczanie.pg.edu.pl/moodle/course/view.php?id=46173</a>						
Learning activity and number of study hours	Learning activity	Participation in didactic classes included in study plan		Participation in consultation hours		Self-study	SUM
	Number of study hours	45		5.0		50.0	100
Subject objectives	To familiarize students with basic corrosion processes and methods of their study.						

Learning outcomes	Course outcome	Subject outcome	Method of verification
	[K6_U01] Can properly use selected analytical, simulation and experimental methods, as well as devices for measuring the fundamental properties of materials and technological processes.	The student is able to plan the time and subsequent steps necessary to assess the causes, mechanism, and rate of corrosion of structural metals and alloys.	[SU4] Assessment of ability to use methods and tools
	[K6_U02] Can operate typical laboratory equipment and analyze material tests	Able to prepare and perform corrosion measurements using modern measurement equipment for corrosion and analytical testing.	[SU1] Assessment of task fulfilment [SU3] Assessment of ability to use knowledge gained from the subject [SU4] Assessment of ability to use methods and tools
	[K6_W07] Has detailed knowledge of selected problems of materials science.	The student is able to determine the resistance of a material under given operating conditions and propose a material solution for defined corrosion hazards.	[SW1] Assessment of factual knowledge
	[K6_K01] Understands the need to improve professional and personal competencies; is conscious of own limitations and knows when to turn to experts, properly establishes priorities helping to accomplish tasks defined by oneself or others.	The student is aware of the need to supplement his/her information in the field of development of measurement methods in corrosion.	[SK5] Assessment of ability to solve problems that arise in practice
Subject contents	Course content – lecture Lecture: -Chemical thermodynamics: corrosion cells, E/pH diagrams, thermodynamic stability of water and its solutions. -Corrosion processes kinetics: E=f(I) diagrams, corrosion processes control. -Types of corrosion: general, pitting, selective, intergranular, crevice, stress corrosion and stress corrosion cracking, corrosion-erosion, cavitation. -Corrosion occurring conditions (practical examples). -Atlas of corrosion fatigue: description and visualization of fatigues. Laboratory: 1.Introduction and safety. 2.Temperature cell. 3.Oxygen concentration cell. 4.Galvanic cell. 5.Crevise corrosion. 6.Intergranular corrosion. 7.Selective corrosion of brass. 8.Pitting corrosion of steel. 9.Water 10.Reserved.		
Prerequisites and co-requisites	Chemical thermodynamics		
Assessment methods and criteria	Subject passing criteria	Passing threshold	Percentage of the final grade
	Passing the lectures	60.0%	50.0%
	Passing the labs	100.0%	50.0%
Recommended reading	Basic literature	Philip. A. Schweitzer, Corrosion-of-Linings-Coatings-Cathodic-and-Inhibitor-Protection-and-Corrosion-Monitoring, CRC Press  F P Ijsseling-General Guidelines for Corrosion Testing of Materials for Marine Applications (European Federation of Corrosion Series)-Maney Publishing (1989)  Yurii I. Kuznetsov, A. D. Mercer, J. G. N. Thomas (auth.)-Organic Inhibitors of Corrosion of Metals-Springer US (1996)	
	Supplementary literature	No requirements	
	eResources addresses	Basic <a href="https://enauczanie.pg.edu.pl/moodle/course/view.php?id=46173">https://enauczanie.pg.edu.pl/moodle/course/view.php?id=46173</a> - The platform provides materials necessary for students to prepare for laboratory exercises and lectures.	
Example issues/ example questions/ tasks being completed	Corrosion cells. Types of corrosion. Potential / pH diagrams		
Practical activites within the subject	Not applicable		

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