

## GDAŃSK UNIVERSITY

## Subject card

Subject name and code	Thermodynamics, PG_00055881							
Field of study	Power Engineering, Power Engineering, Power Engineering							
Date of commencement of studies	October 2023		Academic year of realisation of subject			2024/2025		
Education level	first-cycle studies		Subject group			Obligatory subject group in the field of study Subject group related to scientific research in the field of study		
Mode of study	Full-time studies		Mode of delivery			at the university		
Year of study	2		Language of instruction			Polish		
Semester of study	3		ECTS credits		9.0			
Learning profile	general academic profile		Assessment form		exam			
Conducting unit	Institute of Energy -> Faculty of Mechanical Engineering and Ship Technology							
Name and surname of lecturer (lecturers)	Subject supervisor Teachers		dr hab. inż. Jan Wajs					
Lesson types and methods of instruction	Lesson type	Lecture	Tutorial	Laboratory	Projec	Project Seminar		SUM
	Number of study hours	45.0	30.0	30.0	0.0		0.0	105
	E-learning hours included: 0.0							
Learning activity and number of study hours	Learning activity	Participation i classes incluc plan		Participation in consultation hours		Self-study		SUM
	Number of study hours	105		9.0		111.0		225
Subject objectives	Student acquire basic knowledge of thermodynamics in the dimension of theory and practice.							

Course outcome	Subject outcome	Method of verification					
[K6_W15] knows and understands the basic quantities characteristic methods for thermodynamics, fluic mechanics and hydraulics, hydrology; knows the calculation		[SW1] Assessment of factual knowledge					
[K6_U06] is able to use the basic knowledge on the operation of energy equipment in the field of thermal power plants, thermal and energy and heating systems, combustion engines, compressors and rotating machines to assess the technical condition of the system	Student possesses the knowledge needed to identify physical phenomena occurring during the operation of simple thermodynamic systems (open and closed). On this basis, he correctly describes the types of energy conversion or transformation occurring in them.	[SU2] Assessment of ability to analyse information					
[K6_W02] has a basic knowledge of physics (including optics, electricity and magnetism), chemistry, technical thermodynamics, fluid mechanics and general mechanics needed to understand and describe the basic phenomena occurring in devices and systems, energy plants and transmission networks and their environment	Student defines basic concepts of thermodynamic, 1st and 2nd Law of Thermodynamic and state equations of gases. Student describes gas/steam cycles. Student uses the theory of moist gases and explains fundamentals of thermodynamic combustion.	[SW1] Assessment of factual knowledge					
[K6_U05] is able to formulate and carry out energy balances in		[SU1] Assessment of task fulfilment [SU4] Assessment of ability to use methods and tools					
Subject contents         LECTURE: Basic concepts. The first law of thermodynamics. Ideal gas mod and real gases. Gas laws, thermal and caloric equation of state. Characteri mixtures. Thermodynamic gas cycles. Entropy. The second law of thermod Isobaric evaporation process. Properties of mono-component saturated ste steam. Characteristic processes of steam. Thermodynamic steam cycles. Of Mollier diagram and the basic moist air processes. Fundamentals of refrige sorption heat pumps. Elements of combustion thermodynamics.							
EXERCISES: Simple conversion of energy, heat, work. The balances of power of open or closed thermodynamics systems. State and functions of state of ideal and semi-ideal gases and gas mixtures. Characteristic processes of gases. Gas thermodynamic cycles. Characteristic changes of steam. Calculations thermodynamic steam cycles.							
LABORATORIES: Measurements of thermodynamic parameters: temperature and pressure. Determination of mass flow rate. Determination of air and water enthalpy. Energy balance of heat pump. Testing of the refrigerating unit. Determination of calorific value of solid and gas fuels. Energy balance of piston engine. Testing of the compressor.							
Knowledge from course of physics and mathematics.							
Subject passing criteria	Passing threshold	Percentage of the final grade					
Written exam	56.0%	50.0%					
Middterm colloquiums	56.0%	30.0%					
Laboratory reports	100.0%	20.0%					
Basic literature	<ul> <li>Y. Cengel, M. Boles, Thermodynamics An Engineering Approach, 8th Edition, Wiley, 2014.</li> <li>M.J. Moran, H.N. Shapiro, D.D. Boettner, M.B. Bailey, Fundamentals of Engineering Thermodynamics 8th Ed., Wiley, 2014.</li> <li>R. Mayhew, Engineering thermodynamics/Work &amp; Heat Transfer. Wiley &amp; Sons Inc. 1993, USA.</li> </ul>						
	[K6_W15] knows and understands the basic quantities characteristic methods for thermodynamics, fluid mechanics and hydraulics, hydrology; knows the calculation methods and IT tools necessary to analyse the results of laboratory and field work         [K6_U06] is able to use the basic knowledge on the operation of energy equipment in the field of thermal power plants, thermal and energy and heating systems, combustion engines, compressors and rotating machines to assess the technical condition of the system         [K6_W02] has a basic knowledge of physics (including optics, electricity and magnetism), chemistry, technical thermodynamics, fluid mechanics and general mechanics needed to understand and describe the basic phenomena occurring in devices and systems, energy plants and transmission networks and their environment         [K6_U05] is able to formulate and carry out energy balances in devices and energy systems, also perform an energy audit of a simple building object, is able to perform an energy audit of a simple building object, is able to perform an energy audit of a simple building object, is able to perform a preliminary profitability analysis of a planned energy investment         LECTURE: Basic concepts. The first and real gases. Gas laws, thermal a mixtures. Thermodynamic gas cycle Isobaric evaporation process. Prope steam. Characteristic processes of gases. C Calculations thermodynamic steam of of mass flow rate. Determination of c refrigerating unit. Determination of c resting of the compressor.         Knowledge from course of physics a       Subject passing criteria         Written exam       Middterm colloquiums         Laboratory reports	[K6_W15] knows and understands the basic quantities characteristic performances and hydraulics, fluid mechanics and hydraulics, fluid methods of necessar to analyse the results of laboratory and field work         Student uses graphs and tables of the physical properties to prepare analyse the results of laboratory and field work           [K6_U06] is able to use the basic knowledge on the operation of energy equipment in the field of thermal parameters. Student is concurring during the operation of simple the technical condition of the energy and heating systems.         Student possesses the knowledge needed to identify physical phenomena occurring during the operation of simple the technical condition of the energy use and the technical and closed). On this basis, he correctly describes the types of the technical condition of the energy use and the technical and general mechanics needed to inderstand and describe the basic and systems, energy plants and transmison networks and their environment         Student is able to write the energy uses and energy systems, also of thermodynamic combustion.           [K6_U05] is able to formulate and cergy outperson or system and general mechanics needed to understand and describe the basic and systems, energy plants and transmission networks and their environment         Student is able to write the energy uses and energy systems, also of thermodynamic combustion.           [K6_U05] is able to formulate and cergy agase. Gas laws, thermal and caloric equation of state. Character mixtures. Thermodynamic gas cycles. Entropy. The second law of thermodynamics systems. State and caloric equation of state. Character mixtures. Thermodynamic gas cycles. Entropy. The second law of thermodynamics systems. State and functions of state of ideal and semi- sobaric varporation process. Properties of mono- component saturated statem cycles.					

	Supplementary literature	no requirements			
	eResources addresses	Adresy na platformie eNauczanie:			
example questions/	Present equations of first law of thermodynamics. Describe Carnot Cycle. Describe Rankine / Otto / Diesel / Brayton cycle. Methods of improving the efficiency of Clausius-Rankine cycle. Present definitions of second law of thermodynamics. Operational principle of compressor heat pumps. Heating and humidification of air. Energy balance of piston engine.				
Work placement	Not applicable				