

## Subject card

Subject name and code	Basics of nanophysics, PG_00036981								
Field of study	Nanotechnology								
Date of commencement of studies	October 2023		Academic year of realisation of subject			2023/2024			
Education level	second-cycle studies		Subject group			Optional subject group Subject group related to scientific research in the field of study			
Mode of study	Full-time studies		Mode of delivery			at the university			
Year of study	1		Language of instruction			English			
Semester of study	1		ECTS credits			1.0			
Learning profile	general academic profile		Assessment form			assessment			
Conducting unit	Zakład ceramiki -> Instytut Nanotechnologii i Inżynierii Materiałowej -> Faculty of Applied Physics and Mathematics							ics and	
Name and surname	Subject supervisor		dr inż. Sebastian Wachowski						
of lecturer (lecturers)	Teachers	eachers dr inż. Sebastian Wachowsk							
Lesson types and methods of instruction	Lesson type	Lecture	Tutorial	Laboratory	Projec	t	Seminar	SUM	
	Number of study hours	15.0	0.0	0.0	0.0		0.0	15	
	E-learning hours included: 0.0								
Learning activity and number of study hours	earning activity Participation in classes including plan			Participation in consultation hours		Self-study SUM		SUM	
	Number of study hours	15		2.0		8.0		25	
Subject objectives	The aim of the course is to provide students with basic knowledge about nanotechnology.								
Learning outcomes	Course outcome		Subject outcome			Method of verification			
	K7_K09		Knowledge of various aspects and effects of engineering activities.			[SK2] Assessment of progress of work			
	K7_W09		Knowledge of English terminology related to nanotechnology			[SW1] Assessment of factual knowledge			
	K7_W02		Knowledge from selected branches of nanotechnology		[SW1] Assessment of factual knowledge				
Subject contents	<ol> <li>Introduction.</li> <li>General concepts related to nanotechnology.</li> <li>Bonding in elemental solids: covalent, metallic and van der Waals bonding.</li> <li>Bonding in multielement crystals: ionic, mixed ionic-covalent and hydrogen bonding.</li> <li>Crystalline structure of solids.</li> <li>Band structure of solids: free electron, nearly free electron and tight binding model.</li> <li>Density of states in 0D, 1D, 2D and 3D materials.</li> <li>Properties of carbon nanotubes and graphen.</li> </ol>								

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Prerequisites							
and co-requisites							
Assessment methods	Subject passing criteria	Passing threshold	Percentage of the final grade				
and criteria	Exam	50.0%	100.0%				
Recommended reading	Basic literature	Takaaki Tsurumi et al. Nanoscale physics for materials science, CRC Press.					
		Michael A. Stroscio Phonons in nanostructures, Cambridge University Press.					
	Thomas Heinzel Mesoscopic electronic in Wiley.		ctronic in solid state nanostructures,				
	John D. Joannopoulos et al. Photonic crystals, molding the flow Princeton University Press.						
	Supplementary literature	and chemistry of materials, Wiley.					
	eResources addresses	e:					
		- Nowy - Moodle ID: 30731 oodle/course/view.php?id=30731					
Example issues/ example questions/ tasks being completed	Critical confinement - examples.						
	Schoedingers equation - infinite potential well.						
	How the band gap depends on the size of the crystal						
Work placement	Not applicable						

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