

Subject card

Subject name and code	Analytical Chemistry, PG_00053525								
Field of study	Biomedical Engineering								
Date of commencement of studies	October 2023		Academic year of realisation of subject			2025/2026			
Education level	first-cycle studies		Subject group			Optional subject group Subject group related to scientific research in the field of study			
Mode of study	Full-time studies		Mode of delivery			at the university			
Year of study	3		Language of instruction			Polish			
Semester of study	5		ECTS credits			6.0			
Learning profile	general academic profile		Assessment form			exam			
Conducting unit	Department of Chemistry and Technology of Functional Materials -> Faculty of Chemistry								
Name and surname	Subject supervisor		dr hab. inż. Ewa Wagner-Wysiecka						
of lecturer (lecturers)	Teachers								
Lesson types and methods of instruction	Lesson type	Lecture	Tutorial	Laboratory	Projec	:t	Seminar	SUM	
	Number of study hours	30.0	15.0	30.0	0.0	0.0		75	
	E-learning hours included: 0.0								
Learning activity and number of study hours	Learning activity Participation ir classes include plan					Self-study SUM		SUM	
	Number of study hours	75		15.0		60.0		150	
Subject objectives	The aim of the course is to familiarize students with the issues of modern analytical chemistry and analytical problem-solving methodology.								
Learning outcomes	Course out	come	Subject outcome			Method of verification			
	[K6_U51] can conduct laboratory work connected with chemistry and biochemistry, specific to biomedical engineering		The student explains the chemical basics, describes the types of apparatus used in a given analytical method and explains the principle of its operation. Student defines an analytical problem. I plan to conduct an experiment and collect data. The student measures using selected equipment. Analyzes the obtained data and assesses the reliability of the obtained result. Justifies the use of quality assurance systems. Explains the basic issues related to the problems of environmental and process analytics. Understands the essence of the use of modern analytical methods in biomedical engineering.			[SU2] Assessment of ability to analyse information [SU1] Assessment of task fulfilment			
			The student has knowledge of the procedures related to the basic methods of quantitative analysis. Is able to carry out basic determinations, collect and correctly analyze the results obtained. Is able to determine the areas of application of chemical analytical methods in biomedical engineering.			[SW3] Assessment of knowledge contained in written work and projects [SW1] Assessment of factual knowledge			

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Subject contents	Lecture:Definition and role analytical chemistry. The basic terms in analytical chemistry. The division of the analytical methods. The choice of analytical method. The elaboration of the new analytical method. Statistical methods in analytical chemistry. The types of the samples and sampling methods. Separation and preconcentration methods. Trace analysis. Methods for gases determination. Classical analysis: gravimetric and volumetric methods. Titrimetric methods: acid-base titration, complexometry, redoxymetry, precipitation titration. Spectroscopic methods of analysis: UV-Vis, IR, luminescence, emission, atomic absorption, spectroscopies, turbidimetry, naphelometry, magnetic resonance spectroscopy, mas spectrometry, X-ray spectroscopy. Thermoanalytical methods. Electroanalytical methods: potentiometry, electrogravimetry, coulometry, polarography, voltamperometry, conductometry. Chromatographic methods: gas chromatography, high performance liquid chromatography. Electrophoresis. Kinetic methods of analysis. Miniaturization in analytical chemistry. Elements of environmental analysis. Elements of process analytical chemistry. Quality assurance systems. Tutorial: Statistical analysis of data. Solutions. Units for expressing concentrations and calculating concentrations. Acid-base reactions. Buffers. Acid-base titration, titration curves, titration error. Complexometry: complex stability constants. Complexometric titration. Reactions of precipitation of solids. Precipitation titration. Redox reactions. Equilibria in redox systems. Redox titration. Gas laws. Analysis of gases. The analysis of the composed material. Evaluation of the results. Laboratory: Safety in laboratory. Calibration, standards preparation, sampling. Volumetric analysis – precipitation titration. Gravimetric analysis. UV-Vis spectroscopy. Spectrofluorimetry IR spectroscopy. High performance chromatography. Potentiometry. Optimizing in analytical chemistry.						
Prerequisites and co-requisites	Matters realized during the subject "Chemistry"						
Assessment methods and criteria	Subject passing criteria	Passing threshold	Percentage of the final grade				
	Written exam	51.0%	40.0%				
	Test: material covering problems disscused during excercises	51.0%	30.0%				
	Lab problems tests, correctly done excercises, repors	51.0%	30.0%				
Recommended reading	Basic literature	1. J. Minczewski, Z. Marczenko " Chemia analityczna" t.1 i t.2 . PWN, W-wa, 2007 2. W. Szczepaniak "Metody instrumentalne w analizie chemicznej". PWN, W-wa, 2007 3. D. Kealey, P.J. Haines "Chemia analityczna". PWN, W-wa, 2005; 4. T. Lipiec, Z. Szmal "Chemia analityczna z elementami analizy instrumentalnej". PZWL, W-wa,1997 5. D.A. Skoog, D.M. West, F.J. Holler, S.R. Crouch "Podstawy chemii analitycznej". PWN, W-wa, 2006 6. A. Cygański, B. Ptaszyński, J. Krystek "Obliczenia w chemii analitycznej". WNT, W-wa, 2000 7. A. Cygański " chemiczne metody analizy ilościowej". WNT, W-wa, 1999 8. "Ćwiczenia rachunkowe z chemii analitycznej". Praca zbiorowa pod redakcją Z. Galusa, PWN, W-wa, 1993					
	Supplementary literature	"Miniaturyzacja w chemii analitycznej" praca zbiorowa pod red. Z. Brzózki. Oficyna Wydawnicza Politechniki Warszawskiej , W-wa 2005 2. A. Cygański "Metody spektroskopowe w chemii analitycznej" . WNT, W-wa, 2002 3. A. Cygański " Podstawy metod elektroanalitycznych". WNT, W-wa, 1999 4. A. Hulanicki "Reakcje kwasów i zasad w chemii analitycznej". PWN, W-wa, 1992.					
	eResources addresses	addresses Adresy na platformie eNauczanie:					
Example issues/ example questions/ tasks being completed							
Work placement	Not applicable						

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