



Subject card

Subject name and code	Identification of materials, PG_00048407						
Field of study	Chemistry						
Date of commencement of studies	October 2023		Academic year of realisation of subject		2025/2026		
Education level	first-cycle studies		Subject group		Obligatory subject group in the field of study		
Mode of study	Full-time studies		Mode of delivery		at the university		
Year of study	3		Language of instruction		Polish		
Semester of study	6		ECTS credits		2.0		
Learning profile	general academic profile		Assessment form		assessment		
Conducting unit	Department of Chemistry and Technology of Functional Materials -> Faculty of Chemistry -> Faculties of Gdańsk University of Technology						
Name and surname of lecturer (lecturers)	Subject supervisor		dr inż. Radosław Pomećko				
	Teachers						
Lesson types	Lesson type	Lecture	Tutorial	Laboratory	Project	Seminar	SUM
	Number of study hours	0.0	0.0	15.0	0.0	0.0	15
	E-learning hours included: 0.0						
Learning activity and number of study hours	Learning activity	Participation in didactic classes included in study plan		Participation in consultation hours		Self-study	SUM
	Number of study hours	15		5.0		30.0	50
Subject objectives	The aim of the course is to link the relationship between the structure of materials and their physical and chemical properties, and to familiarize with experimental methods that enable the measurment of these features.						
Learning outcomes	Course outcome		Subject outcome		Method of verification		
	[K6_U06] can analyze the functioning of equipment, apparatus and technology lines used in laboratories and chemical industry, and can recognize and propose methods to solve the simple engineering tasks which he can meet as an Engineer and select and use routine methods, chemical apparatus and tools to solve practical engineering tasks, including also technological processes; can himself/herself read and make technical drawings using CAD software		The student knows how to determine the physical and physicochemical characteristics of the material, important for its function. The student knows and can propose methods to measure these properties		[SU4] Assessment of ability to use methods and tools [SU2] Assessment of ability to analyse information [SU1] Assessment of task fulfilment		

Subject contents	Course content – laboratory Program laboratoriów obejmuje trzy pięciogodzinne bloki zajęć obejmujące: 1) Blok pierwszy: a) Przedstawienie założeń przedmiotu. Wprowadzenie teoretyczne do zajęć. Zasady BHP b) Realizacja ćwiczenia: Synteza i charakterystyka polimerów przewodzących (obejmujące: syntezę polianiliny oraz badania jej elektrchromizmu, oraz właściwości pojemnościowych, syntezę i charakterystykę elektrolitów żelowych) 2) Blok drugi: Realizacja ćwiczenia: Synteza, identyfikacja polimerów (obejmujące: syntezę metakrylanów oraz żywic formaldehydowych oraz badania IR otrzymanych substancji uzupełnione o badania gęstości, palności, zawartości azotu) 3) Blok trzeci: a) Realizacja ćwiczenia: Synteza, identyfikacja układów hydrożelowych (obejmujące: syntezę hydrożeli opartych na PVA lub żel krzemionkowy, charakteryzacja metodami IR, badanie chłonności otrzymanych materiałów. Badania pamięci kształtów hydrożeli PVA) b) Pisemne zaliczenie kursu		
Prerequisites and co-requisites	Basics of organic chemistry, electrochemistry and physical chemistry		
Assessment methods and criteria	Subject passing criteria	Passing threshold	Percentage of the final grade
		70.0%	50.0%
		60.0%	50.0%
Recommended reading	Basic literature	Handbook of Materials Characterization, S.K. Sharma, Springer 2018	
		Chemia polimerów, Jan Pielichowski, Andrzej Puszyński, Fosze, Rzeszów 2015	
		"Polymer electrolytes", F.M. Gray, VCH 1998	
	Supplementary literature	Handbook of Spectroscopy: Second, Enlarged Edition, G. Gauglitz, D.S. Moore, WileyVCH 2014	
		Springer Handbook of Materials Measurement Methods, H. Czichos, T. Saito, L. Smith, Springer, Berlin, Heidelberg, 2006	
	eResources addresses		
Example issues/ example questions/ tasks being completed	- the impact of copolymer structure on its physicochemical properties - structure of organic and inorganic hydrogels - structure of polymers and their electro-conductive properties		
Practical activities within the subject	Not applicable		

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