



## Subject card

Subject name and code	Organic Chemistry, PG_00037380						
Field of study	Chemistry						
Date of commencement of studies	October 2023	Academic year of realisation of subject			2024/2025		
Education level	first-cycle studies	Subject group			Obligatory subject group in the field of study Subject group related to scientific research in the field of study		
Mode of study	Full-time studies	Mode of delivery			at the university		
Year of study	2	Language of instruction			Polish		
Semester of study	4	ECTS credits			4.0		
Learning profile	general academic profile	Assessment form			assessment		
Conducting unit	Department of Organic Chemistry -> Faculty of Chemistry						
Name and surname of lecturer (lecturers)	Subject supervisor	dr hab. inż. Teresa Olszewska					
	Teachers	dr hab. inż. Teresa Olszewska					
Lesson types and methods of instruction	Lesson type	Lecture	Tutorial	Laboratory	Project	Seminar	SUM
	Number of study hours	30.0	15.0	0.0	0.0	0.0	45
	E-learning hours included: 0.0						
Learning activity and number of study hours	Learning activity	Participation in didactic classes included in study plan		Participation in consultation hours		Self-study	SUM
	Number of study hours	45		5.0		50.0	100
Subject objectives	The aim of the subject is to familiarize students with the basics of organic chemistry covering the structure, properties, reactions and reaction mechanisms of organic compounds.						
Learning outcomes	Course outcome	Subject outcome			Method of verification		
	[K6_W09] has knowledge on chemical management and the concept of sustainable development necessary to conduct the management of chemicals (including dangerous substances) in the industrial plant, knows health and safety issues and ergonomics.	A student knows the toxicity and danger associated with the use of specific substances from the class of organic compounds in question. He knows typical substitutes for harmful chemical solvents for less toxic to the environment.			[SW1] Assessment of factual knowledge		
	K6_W03	A student knows the basic reaction mechanisms of main organic compounds classes discussed during the lectures, understands the concepts of stereospecific reaction; kinetic and thermodynamic reaction control.			[SW1] Assessment of factual knowledge		
	K6_W02	A student has knowledge of the structure and properties of organic compounds discussed during the lectures and knows the basic reactions they undergo.			[SW1] Assessment of factual knowledge		
[K6_U01] knows how to get information from literature, databases and other sources, can integrate the information obtained, interpret and critically evaluate it, and draw conclusions, and to formulate and justify the opinions	A student is able to propose a method of synthesis of a given organic compound starting from an appropriate substrate. In addition, the student is able to indicate the most and least energy-stable conformation of simple organic molecules and determine the configuration of a chiral molecule with a chiral center.			[SU1] Assessment of task fulfillment			

Subject contents	<p><b>Introductory information</b> - chemical bonds, writing chemical formulas, resonance method, molecular geometry prediction, atomic and molecular orbitals, hybridization, constitutional isomerism, condensed and dashed structural formulas.</p> <p><b>Saturated hydrocarbons</b> - nomenclature, homologous series, physical properties, conformational analysis of alkanes and cycloalkanes, chemical reactions of alkanes, synthesis of alkanes and cycloalkanes.</p> <p><b>Alcohols and alkyl halides</b> - nomenclature, acid and basic properties, transformation of alcohols into alkyl halides, alkanes halogenation reaction and its mechanism, activation energy, radical chain reactions, reaction selectivity, nucleophilic substitution and elimination reactions - mechanisms and stereochemistry, stability of carbocations.</p> <p><b>Alkenes</b> - nomenclature, electronic structure, stereochemistry, synthesis and chemical properties, rearrangement of carbocations, ion and radical addition reactions to multiple bonds, regioselectivity, stereospecificity, hydroxylation and hydroboration of alkenes, polymerization, oxidation and ozonolysis, substitution in the allylic position.</p> <p><b>Stereoisomers</b> - geometric isomers, chiral molecules, enantiomers and diastereomers, configuration, CIP rules, racemic mixtures, reactions leading to diastereomers. <b>Dienes and polyenes</b> - properties of conjugated unsaturated systems, resonance method - further information, reactions of conjugated dienes, kinetic and thermodynamic control, Diels-Alder reaction.</p>											
Prerequisites and co-requisites	<p>The basic knowledge of the structure of elements and their compounds, in particular elements of groups I, II and III of the periodic table.</p> <p>The basic knowledge of the concept of acids, bases and salts; the types of chemical reactions and the geometry of some simple molecules.</p>											
Assessment methods and criteria	<table border="1"> <thead> <tr> <th data-bbox="453 1102 794 1128">Subject passing criteria</th> <th data-bbox="799 1102 1141 1128">Passing threshold</th> <th data-bbox="1145 1102 1484 1128">Percentage of the final grade</th> </tr> </thead> <tbody> <tr> <td data-bbox="453 1135 794 1184">lecture - two tests based on lecture material</td> <td data-bbox="799 1135 1141 1184">60.0%</td> <td data-bbox="1145 1135 1484 1184">50.0%</td> </tr> <tr> <td data-bbox="453 1191 794 1240">tutorials - tests on the material carried out during the classes</td> <td data-bbox="799 1191 1141 1240">60.0%</td> <td data-bbox="1145 1191 1484 1240">50.0%</td> </tr> </tbody> </table>			Subject passing criteria	Passing threshold	Percentage of the final grade	lecture - two tests based on lecture material	60.0%	50.0%	tutorials - tests on the material carried out during the classes	60.0%	50.0%
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	eResources addresses	Adresy na platformie eNauczenie: CHEMIA ORGANICZNA, semestr IV, rok 2024/25 - Moodle ID: 45260 <a href="https://enauczenie.pg.edu.pl/moodle/course/view.php?id=45260">https://enauczenie.pg.edu.pl/moodle/course/view.php?id=45260</a>
Example issues/ example questions/ tasks being completed	<p>1) Acid-catalyzed dehydration of neopentyl alcohol, (CH<sub>3</sub>)<sub>2</sub>CCH<sub>2</sub>OH, yields 2-methyl-2-butene as the major product. Outline a mechanism showing all steps in its formation.</p> <p>2) Outline all steps in a synthesis of propyne from each of the following:</p> <p>a) CH<sub>3</sub>COCH<sub>3</sub></p> <p>b) CH<sub>3</sub>CH<sub>2</sub>CHBr<sub>2</sub></p> <p>c) CH<sub>3</sub>CHBrCH<sub>2</sub>Br</p> <p>d) CH<sub>3</sub>CH=CH<sub>2</sub></p> <p>3) Write structural formulas for the products that form when 1-butene reacts with each of the following reagents:</p> <p>a) HBr in the presence of peroxides</p> <p>b) Br<sub>2</sub> in CCl<sub>4</sub>, then KI in acetone</p> <p>c) cold concentrated H<sub>2</sub>SO<sub>4</sub></p> <p>d) Br<sub>2</sub> in H<sub>2</sub>O</p> <p>e) O<sub>3</sub>, then Zn, H<sub>2</sub>O</p> <p>4) Describe with equations a simple chemical test that you could use to distinguish between the members of following pairs of compounds. (In each case tell what the visible result would be.)</p> <p>a) pentane and 1-pentyne</p> <p>b) pentane and 1-pentanol</p> <p>c) 2-pentyne and 1-pentyne</p>	
Work placement	Not applicable	

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