

## Subject card

Subject name and code	, PG_00057778							
Field of study	Green Technologies							
Date of commencement of studies	October 2023		Academic year of realisation of subject			2024/2025		
Education level	first-cycle studies		Subject group			Optional subject group Subject group related to scientific research in the field of study		
Mode of study	Full-time studies		Mode of delivery			at the university		
Year of study	2		Language of instruction			English		
Semester of study	3		ECTS credits			5.0		
Learning profile	general academic profile		Assessment form		exam			
Conducting unit	Department of Analytical Chemistry -> Faculty of Chemistry							
Name and surname	Subject supervisor		prof. dr hab. inż. Andrzej Wasik					
of lecturer (lecturers)	Teachers							
Lesson types and methods of instruction	Lesson type	Lecture	Tutorial	Laboratory	Projec	roject Semir		SUM
	Number of study hours	30.0	0.0	30.0	0.0		0.0	60
	E-learning hours included: 0.0							
Learning activity and number of study hours	Learning activity	Participation in didactic classes included in study plan		Participation in consultation hours		Self-study		SUM
	Number of study hours	60		10.0		55.0		125
Subject objectives	Familiarize students with the basics of chemical processes occurring in the natural environment, physical chemistry of the atmosphere, water and soil. Presentation of geochemical cycles of the most important elements in the environment. Familiarization with the most important environmental pollutants, their sources and methods of detection.							

Data wygenerowania: 22.11.2024 01:59 Strona 1 z 2

Learning outcomes	Course outcome	Subject outcome	Method of verification				
	[K6_W02] has a basic knowledge of chemistry including general chemistry, inorganic, organic, physical, analytical, including the knowledge necessary to describe and understand the phenomena and chemical processes occurring in the environment; measurement and the determination of the parameters of these processes.	The student has basic knowledge in the field of chemistry necessary to describe and understand phenomena and chemical processes occurring in the natural environment. Knows the basics of the methods used for measuring the level of environmental pollution.	[SW1] Assessment of factual knowledge				
	[K6_W03] has a basic knowledge of soil, air and water pollutants, design and supervision of environmentally friendly technologies and technologies which do not produce waste, knows technology of cleaning and neutralization of industrial waste and wastewater management, has a basic understanding of the theoretical basis of methods and types of apparatus used in chemical analysis of environmental pollutants	The student has basic knowledge in the field of soil, air and water protection against pollution and the theoretical basis of methods and types of apparatus used in the analysis of environmental pollution.	[SW1] Assessment of factual knowledge				
	[K6_U04] capable of formulating and solving design tasks in the field of environmental technology to recognize their non-technical aspects, including environmental, economic and legal. Is capable of applying the principles of occupational health and safety. Is able to make initial assessment of engineering solutions and actions	The student notices non-technical, including environmental, aspects of technologies used in environmental protection. Applies the principles of occupational health and safety.	[SU4] Assessment of ability to use methods and tools [SU2] Assessment of ability to analyse information				
Subject contents	Atmospheric chemistry. Aquatic chemistry. Soil chemistry. Persistent organic pollutants in the environment. Carbon cycle. Nitrogen cycle. Phosphorus cycle. Oxygen and sulfur cycle. The role of the chemical elements in living organisms. Heavy metals and micronutrients. Environmental analytics. Methods of measuring the degree of pollution. Remote pollution measurement methods.						
Prerequisites and co-requisites	Passed course of Inorganic Chemis	stry					
Assessment methods	Subject passing criteria	Passing threshold	Percentage of the final grade				
and criteria	Laboratory reports	60.0%	30.0%				
	Written exam	60.0%	70.0%				
Recommended reading	Basic literature	Gary W vanLoon and Stephen J Duffy, Environmental Chemistry, Oxford University Press					
	Supplementary literature	1. S. Manahan, Environmental Chemistry, CRC Press, 2009					
	eResources addresses	Adresy na platformie eNauczanie:					
Example issues/ example questions/ tasks being completed	Whats the difference between a macro elements and a micro elements? Give five examples of each group of elements.						
	2. Why do temperature variations occur with various altitudes of the atmosphere? Give a detailed diagram or graph, justifying the changes.						
	3. Explain the concepts of nitrification and biological nitrogen fixation.  4. What is the difference between tandem mass spectrometry and regular mass spectrometry? Draw a						
	4. What is the difference between tandem mass spectrometry and regular mass spectrometry? Draw a diagram of single and tandem mass spectrometer.						
Work placement	Not applicable						

Document generated electronically. Does not require a seal or signature.

Data wygenerowania: 22.11.2024 01:59 Strona 2 z 2