



Subject card

Subject name and code	, PG_00053321						
Field of study	Biomedical Engineering, Biomedical Engineering, Biomedical Engineering						
Date of commencement of studies	October 2023	Academic year of realisation of subject			2023/2024		
Education level	second-cycle studies	Subject group			Optional subject group Subject group related to scientific research in the field of study		
Mode of study	Full-time studies	Mode of delivery			at the university		
Year of study	1	Language of instruction			Polish		
Semester of study	2	ECTS credits			3.0		
Learning profile	general academic profile	Assessment form			assessment		
Conducting unit	Instytut Fizyki i Informatyki Stosowanej -> Faculty of Applied Physics and Mathematics						
Name and surname of lecturer (lecturers)	Subject supervisor	dr hab. Jan Franz					
	Teachers	dr hab. Jan Franz dr Małgorzata Franz					
Lesson types and methods of instruction	Lesson type	Lecture	Tutorial	Laboratory	Project	Seminar	SUM
	Number of study hours	30.0	15.0	0.0	0.0	0.0	45
	E-learning hours included: 0.0						
Learning activity and number of study hours	Learning activity	Participation in didactic classes included in study plan	Participation in consultation hours		Self-study		SUM
	Number of study hours	45	5.0		25.0		75
Subject objectives	The aim of the course is to familiarize the student with selected issues of molecular physics.						
Learning outcomes	Course outcome	Subject outcome			Method of verification		
	[K7_W02] Knows and understands, to an increased extent, selected laws of physics and physical phenomena, as well as methods and theories explaining the complex relationships between them, constituting advanced general knowledge in the field of technical sciences related to the field of study	The student knows and understands the basics of molecular physics.			[SW1] Assessment of factual knowledge		
	[K7_U02] can perform tasks related to the field of study as well as formulate and solve problems applying recent knowledge of physics and other areas of science	The student is able to use the acquired knowledge in the applications of molecular spectroscopy.			[SU3] Assessment of ability to use knowledge gained from the subject		
	[K7_K01] is ready to create and develop models of proper behaviour in the work and life environment; undertake initiatives; critically evaluate actions of their own, teams and organisations they are part of; lead a group and take responsibility for its actions; responsibly perform professional roles taking into account changing social needs, including: n - developing the achievements of the profession, n- observing and developing rules of professional ethics and acting to comply to these rules	The student is able to prepare a presentation on molecular physics.			[SK4] Assessment of communication skills, including language correctness		

Subject contents	1. Introduction to the lecture: photons and waves, atomic structure. 2. Fundamentals of quantum mechanics: wave-particle duality, wave function, Schrödinger equation, Heisenberg uncertainty principle, free particle motion, particle in a square-well, tunnelling phenomenon, hydrogen atom, atomic orbital, hydrogen ion. 3. Electric and magnetic properties of molecules. 4. Molecular solid: types of crystal bonds and their characteristics. 5. Interaction of electromagnetic radiation with particles: forms of particle energy, quantization of energy, energy distribution in the state of thermal equilibrium, probability of absorption and emission of radiation, types of spectroscopy. 6. The rotational energy of molecules. 7. The vibrational energy of molecules. 8. The interaction of electromagnetic radiation with vibrating molecules: Raman spectrum. 9. Molecular electronic transitions: electronic states and the energy of electronic states.														
Prerequisites and co-requisites															
Assessment methods and criteria	<table border="1" data-bbox="448 360 1487 499"> <thead> <tr> <th data-bbox="448 360 794 394">Subject passing criteria</th> <th data-bbox="794 360 1141 394">Passing threshold</th> <th data-bbox="1141 360 1487 394">Percentage of the final grade</th> </tr> </thead> <tbody> <tr> <td data-bbox="448 394 794 427">calculational tasks</td> <td data-bbox="794 394 1141 427">50.0%</td> <td data-bbox="1141 394 1487 427">20.0%</td> </tr> <tr> <td data-bbox="448 427 794 461">presentation</td> <td data-bbox="794 427 1141 461">50.0%</td> <td data-bbox="1141 427 1487 461">20.0%</td> </tr> <tr> <td data-bbox="448 461 794 499">passing the lecture</td> <td data-bbox="794 461 1141 499">50.0%</td> <td data-bbox="1141 461 1487 499">60.0%</td> </tr> </tbody> </table>			Subject passing criteria	Passing threshold	Percentage of the final grade	calculational tasks	50.0%	20.0%	presentation	50.0%	20.0%	passing the lecture	50.0%	60.0%
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Example issues/ example questions/ tasks being completed	<p data-bbox="448 1055 1487 1088">The examples of lecture assignments:</p> <p data-bbox="448 1155 1487 1189">Represent and describe the forms of internal energy of molecules.</p> <p data-bbox="448 1234 1487 1267">Represent and describe the physical quantities which characterizing the magnetic properties of molecules.</p>														
Work placement	Not applicable														