



Subject card

Subject name and code	Technical Thermodynamics, PG_00048913						
Field of study	Chemistry in Construction Engineering						
Date of commencement of studies	October 2023		Academic year of realisation of subject		2023/2024		
Education level	first-cycle studies		Subject group		Obligatory subject group in the field of study Subject group related to scientific research in the field of study		
Mode of study	Full-time studies		Mode of delivery		at the university		
Year of study	1		Language of instruction		Polish		
Semester of study	2		ECTS credits		2.0		
Learning profile	general academic profile		Assessment form		exam		
Conducting unit	Department of Energy Conversion and Storage -> Faculty of Chemistry						
Name and surname of lecturer (lecturers)	Subject supervisor		dr inż. Anna Kuczyńska-Łażewska				
	Teachers						
Lesson types and methods of instruction	Lesson type	Lecture	Tutorial	Laboratory	Project	Seminar	SUM
	Number of study hours	15.0	15.0	0.0	0.0	0.0	30
	E-learning hours included: 0.0						
Learning activity and number of study hours	Learning activity	Participation in didactic classes included in study plan		Participation in consultation hours		Self-study	SUM
	Number of study hours	30		3.0		17.0	50
Subject objectives	To familiarize students with the basics of technical thermodynamics, linking concepts and relationships of the thermodynamic properties of substances and technical applications.						
Learning outcomes	Course outcome		Subject outcome		Method of verification		
	K6_W02		Student knows how to define the basic concepts of thermodynamics. Student knows the basis of technical thermodynamics. Student is able to link concepts and relationships of the thermodynamic properties of substances and technical applications.		[SU3] Assessment of ability to use knowledge gained from the subject		
	K6_W07		Student can solve the practical problems associated with the heat exchange. Student knows how to apply thermodynamics to the modeling and optimization of technological processes.		[SW1] Assessment of factual knowledge		
	K6_U08		The student knows how to distinguish between basic devices for measuring temperatures and pressures. He knows the types of these devices and knows how to measure basic quantities.		[SU4] Assessment of ability to use methods and tools		
Subject contents	1. Basic concepts and the ability to use: Basic concepts of general thermodynamics: internal energy, state of thermodynamic state function, the function of the process, thermodynamic potentials, pressure, temperature, volume, heat, specific heat, enthalpy, entropy, egzergy, thermodynamic system, the system Isolated. Laws of thermodynamics: Zeroth law of thermodynamics, first law of thermodynamics in isolated and open systems, second law of thermodynamics. Classification of thermodynamic processes: reversible, irreversible, spontaneous, quasi-static. 2. Scales and measuring temperature. 3. Properties of gases: Gas Models Equation of state Avogadro's law Warm molar ideal gas Mixtures of gases. 4. Thermodynamics of reversible processes. 5. Basics of thermodynamics of irreversible processes. 6. Properties of real gases, the characteristic changes. Thermodynamic cycles. 7. Heat transfer by convection, conduction and radiation. 8. Technical thermodynamic calculations.						

Prerequisites and co-requisites	mathematics, physics		
Assessment methods and criteria	Subject passing criteria	Passing threshold	Percentage of the final grade
	midterm colloquium	60.0%	50.0%
	egzam	60.0%	50.0%
Recommended reading	Basic literature	1. Klugmann-Radziemska E., Termodynamika Techniczna, Wyd. PG 2009 2. Wiśniewski S: Termodynamika techniczna, Warszawa WNT 1987 3. Staniszewski B.: Termodynamika, PWN 1982 4. Pudlik W.: Termodynamika, Wydawnictwo PG 1998 5. Gumiński K: Termodynamika, PWN 1982	
	Supplementary literature	1. Kalinowski E.: Termodynamika, PW Wrocław, 1994 2. Szarawara J.: Termodynamika Chemiczna, WNT 1985 3. Michałowski S, Wańkowicz K.: Termodynamika procesowa, WNT 1993	
	eResources addresses	Adresy na platformie eNauczanie:	
Example issues/ example questions/ tasks being completed	1. The temperature scales, temperature measurement. 2. The ideal gas model, the semi perfect, real gases 3. Specific heat of ideal gases and półdoskonałych 4. The kinetic energy and the absolute temperature 5. The pressure of an ideal gas 6. Mixtures of ideal gases 7. Heat molar gases and gas mixtures excellent 8. The first law of thermodynamics in a closed system 9. The first law of thermodynamics in the flow system 1. The second law of thermodynamics 2. Work and technical work 3. Entropy for ideal gases and T-s diagram 4. Isothermal transformation of an ideal gas 5. Isobaric transformation of an ideal gas 6. Isochoric process of an ideal gas 7. The adiabatic transformation of an ideal gas 8.The polytropic transformation of an ideal gas 9. Thermodynamic cycles		
Work placement	Not applicable		