

## 。 GDAŃSK UNIVERSITY OF TECHNOLOGY

## Subject card

Subject name and code	Modelling of physical phenomena, PG_00031936								
Field of study	Technical Physics								
Date of commencement of studies	February 2024		Academic year of realisation of subject			2024/2025			
Education level	second-cycle studies		Subject group			Obligatory subject group in the field of study			
						Subject group related to scientific research in the field of study			
Mode of study	Full-time studies		Mode of delivery			at the university			
Year of study	1		Language of instruction			English			
Semester of study	2		ECTS credits			3.0			
Learning profile	general academic profile		Assessment form			assessment			
Conducting unit	Department of Theore	etical Physics a	s and Quantum Information -> Faculty of Applied Physics and Mathematics						
Name and surname of lecturer (lecturers)	Subject supervisor		prof. dr hab. Julien Guthmuller						
	Teachers	prof. dr hab.	Julien Guthmul	ler					
Lesson types and methods of instruction	Lesson type	Lecture	Tutorial	Laboratory	Projec	t	Seminar	SUM	
	Number of study hours	15.0	0.0	30.0	0.0		0.0	45	
	E-learning hours included: 0.0								
Learning activity and number of study hours	Learning activity	Participation in classes includ plan	I didactic Participation in   ed in study consultation hours		Self-study		SUM		
	Number of study hours	45		5.0		25.0		75	
Subject objectives	Introduce the students to the basic theoretical and computational methods to perform quantum simulations of molecular systems properties. The students will gain knowledge in the quantum chemistry techniques and will apply them in practice to diatomic and polyatomic molecules. The students will learn how to analyze their results and how to assess them by comparison with experimental data.								
Learning outcomes	Course outcome		Subject outcome		Method of verification				
	[K7_W04] Has enhanced knowledge of mathematical, numerical and simulation methods applied in the description and modelling of physical phenomena.		The students will learn the theories, approximations and algorithms required to simulate atomic and molecular phenomena.			[SW1] Assessment of factual knowledge			
	[K7_U05] Can plan and conduct theoretical calculations, experimental research and computer simulations, critically analyze their results, draw conclusions and form reasoned opinions.		The students will learn the use of computer programs to describe molecular properties. The students will learn how to analyze their results and how to assess them by comparison with experimental data.			[SU1] Assessment of task fulfilment [SU2] Assessment of ability to analyse information [SU4] Assessment of ability to use methods and tools			
	[K7_U06] Can apply obtained knowledge of physics to exact sciences, natural and technical sciences.		The knowledge acquired by the students can be applied in the fields of solid state physics, nanotechnology, chemistry and biology.			[SU3] Assessment of ability to use knowledge gained from the subject [SU4] Assessment of ability to use methods and tools			

Subject contents	3orn-Oppenheimer approximation and definition of the potential energy surface. Calculation of potential nergy curves, dipole moments and bond lengths for diatomic molecules.						
	- Hartree-Fock approximation and Roothaan equations. Optimization of molecular geometries, calculation of ionization energies, electron affinities and molecular orbitals.						
	- Post-Hartree-Fock methods and atomic basis sets. Accurate calculation of ionization energies with the Coupled-Cluster methods. Investigation of the basis set convergence.						
	- Vibrational energies in the harmonic approximation. Calculation of vibrational frequencies, normal modes, infrared spectra and Raman spectra for polyatomic molecules.						
	- Density functional theory and time-dependent density functional theory. Calculation of excited state properties, absorption spectra, fluorescence energies and solvent effects.						
Prerequisites and co-requisites	Not applied.						
Assessment methods	Subject passing criteria	Passing threshold	Percentage of the final grade				
and criteria	projects	55.0%	60.0%				
	test	55.0%	40.0%				
Recommended reading	Basic literature	Piela L., Idee chemii kwantowej, PWN 2005					
		Jensen F., Introduction to Computational Chemistry, John Wiley & Sons Ltd. 2011					
		Szabo A. and Ostlund N. S., Modern Quantum Chemistry, Dover Publications, Inc.					
		https://orcaforum.cec.mpg.de/					
	Supplementary literature	Not applied.					
	eResources addresses Adresy na platformie eNauczanie:						
Example issues/ example questions/ tasks being completed	not applied						
Work placement	Not applicable						

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