

## 。 GDAŃSK UNIVERSITY OF TECHNOLOGY

## Subject card

Subject name and code	Atomic and molecular physics II, PG_00039517								
Field of study	Technical Physics								
Date of commencement of studies	February 2024		Academic year of realisation of subject			2024/2025			
Education level	second-cycle studies		Subject group			Optional subject group Subject group related to scientific research in the field of study			
Mode of study	Full-time studies		Mode of delivery			at the university			
Year of study	1		Language of instruction			Polish			
Semester of study	2		ECTS credits		4.0				
Learning profile	general academic pro	academic profile		nt form		exam			
Conducting unit	Zakład Fizyki Zderzeń Elektronowych -> Instytut Fizyki i Informatyki Stosowanej -> Faculty of Applied Physics and Mathematics								
Name and surname of lecturer (lecturers)	Subject supervisor		dr hab. Paweł Możejko						
	Teachers		dr hab. Jan Franz						
			prof. dr hab. Julien Guthmuller						
			dr hab. Paweł Możejko						
Lesson types and methods of instruction	Lesson type	Lecture	Tutorial	Laboratory	Projec	ct Seminar		SUM	
	Number of study hours	30.0	15.0	0.0	0.0		15.0	60	
	E-learning hours included: 0.0								
Learning activity and number of study hours	Learning activity	Participation in didactic classes included in study plan		Participation in consultation hours		Self-study		SUM	
	Number of study hours	60		8.0		32.0		100	
Subject objectives	Presentation of selected topics related to the theoretical description of the structure of multi-electron atoms and molecules (lecture, exercises). Presentation of selected computational methods of the atomic and molecular physics (lecture, exercises).								
	seminar	s).							
Learning outcomes	Course outcome		Subject outcome			Method of verification			
	[K7_K03] Can cooperate and work in a group, performing different functions. Can make self- assessment, as well as constructively assess the effects of other persons' work.		The student is able to prepare and conduct a seminar presentation. Can take part in a discussion of another person's seminar presentation.			[SK1] Assessment of group work skills [SK4] Assessment of communication skills, including language correctness			
	[K7_W02] Has enhanced, theoretically-founded, detailed knowledge of selected field of physics, and sufficient knowledge of related fields of science or technology.		The student knows the basics of the theoretical description of the structure of multi-electron atoms and molecules.			[SW1] Assessment of factual knowledge			

Subject contents							
	1) Approximative methods of solving the Schroedinger equation - variation principle						
	T/ Approximative methods of solving the Schloedinger equation - variation principle						
	2) The ground state of the helium atom						
	3) The Ritz method						
	4) Multi-electron systems - Hartree-Fock equations						
	5) Hydrogen-like ion and hydrogen molecule						
	6) Born-Oppenheimer approximation						
	7) Electronic structure of molecules						
	8) Oscillatory structure of molecules						
	9) Rotational structure of molecules						
	10) Basic methods of quantum chemistry ((multi-configuration methods, DFT)						
	Seminars: topics related to modern experimental methods used in the study of atoms and molecules.						
Prerequisites and co-requisites	Knowledge of elementary quantum mechanics.						
Assessment methods	Subject passing criteria	Passing threshold	Percentage of the final grade				
and criteria	Writing exam	50.0%	50.0%				
	Seminar	50.0%	50.0%				
Recommended reading	Basic literature	Pauling, L: Introduction to Quantum Mechanics: With Applications to Chemistry (Dover)					
	L. Piela "Idee Chemii Kwantowej" PWN		WN				
	Demtroder Wolfgang "Atoms, Molecules and Photons" Sprin						
	Supplementary literature	S. Kryszewski "Mechanika kwantow	a" Wyd. UG				
	eResources addresses	Adresy na platformie eNauczanie:					
Example issues/ example questions/ tasks being completed	How to describe the electronic structure of the hydrogen-like ion?						
tasks being completed	How to calculate the ground state of the hydrogen molecule?						
	Please explain the Born-Oppenheimer approximation using the example of a diatomic molecule.						
	What is the DFT method?						
Work placement	Not applicable						

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