



## Subject card

Subject name and code	, PG_00062833						
Field of study	Recycling and Energy Recovery						
Date of commencement of studies	October 2023	Academic year of realisation of subject			2023/2024		
Education level	first-cycle studies	Subject group					
Mode of study	Full-time studies	Mode of delivery			at the university		
Year of study	1	Language of instruction			Polish		
Semester of study	2	ECTS credits			3.0		
Learning profile	general academic profile	Assessment form			assessment		
Conducting unit	Department of Organic Chemistry -> Faculty of Chemistry						
Name and surname of lecturer (lecturers)	Subject supervisor	dr hab. Sławomir Makowiec					
	Teachers	dr hab. Sławomir Makowiec dr inż. Karolina Kucharska dr inż. Agata Sommer dr inż. Jan Alfuth dr inż. Karol Biernacki dr hab. inż. Hanna Staroszczyk					
Lesson types and methods of instruction	Lesson type	Lecture	Tutorial	Laboratory	Project	Seminar	SUM
	Number of study hours	20.0	0.0	20.0	0.0	0.0	40
	E-learning hours included: 0.0						
Learning activity and number of study hours	Learning activity	Participation in didactic classes included in study plan	Participation in consultation hours		Self-study	SUM	
	Number of study hours	40	5.0		30.0	75	
Subject objectives	The student acquires basic knowledge and skills in organic chemistry. Learning the chemical and physical properties of basic groups of organic compounds. Learning examples of practical use of knowledge about the reactivity of organic compounds in the context of the synthesis of fuel additives, biomass processing, and polymer modification.						

Learning outcomes	Course outcome	Subject outcome	Method of verification
	[K6_W01] demonstrates knowledge and understanding of mathematics and other exact sciences and engineering disciplines at the level necessary to solve theoretical, engineering and technological problems and issues.	The student knows what reactions the main groups of organic compounds undergo. The student knows what physicochemical properties have the main groups of organic compounds.	[SW3] Assessment of knowledge contained in written work and projects [SW1] Assessment of factual knowledge
	[K6_W02] analyzes engineering and technological issues and problems in the area of raw materials and energy recovery using appropriate and appropriate analytical, numerical and experimental tools and methods	The student knows what methods of separating mixtures of organic compounds to use depending on their chemical and physical properties.	[SW3] Assessment of knowledge contained in written work and projects [SW1] Assessment of factual knowledge
	[K6_U01] applies knowledge of mathematics and other exact sciences and engineering disciplines to solve theoretical, engineering and technological problems and issues.	The student is able to determine the basic physicochemical properties of a compound based on its structure.	[SU2] Assessment of ability to analyse information [SU3] Assessment of ability to use knowledge gained from the subject
	[K6_U02] solves engineering issues and problems in the area of raw materials and energy recovery through the use of appropriate analytical, numerical and experimental tools and methods.	The student is able to carry out a simple physical or chemical process of separating mixtures of organic compounds into individual components.	[SU3] Assessment of ability to use knowledge gained from the subject [SU2] Assessment of ability to analyse information
Subject contents	<p>Lecture topics</p> <p>1 Introduction to coal chemistry. Structural formulas - ways of writing the structure of molecules. Isomerism of organic compounds. Alkanes, alkenes, alkynes, aromatic hydrocarbons, reactivity, chemical and physical properties, methods of preparation and purification. Acids, bases, electrophiles, nucleophiles, radicals - brief characteristics.</p> <p>2 Alcohols, phenols and ethers. Preparation, reactivity, physical properties. The use of alcohols and ethers as fuel additives. Chemical basis of the process of increasing the octane number of gasoline.</p> <p>3 Carbonyl compounds: ketones, aldehydes, carboxylic acids, esters and amides. Synthetic methods of production and natural sources. Reactivity and physical properties. Chemistry of the biodiesel production process.</p> <p>4 Organic compounds containing nitrogen: amines, nitriles, amides, amino acids, ureas, isocyanates. Reactivity and preparation methods. Heterocyclic compounds structure and physicochemical properties.</p> <p>5 Synthetic polymers. Polymer structure and physical properties. Ways to receive. Monomer synthesis methods. Polymerization and polycondensation. Depolymerization reactions. Polymer degradation</p> <p>6. Natural polymers. Cellulose, starch, glycogen, proteins. Structure and the resulting chemical and physical properties. Methods of modifying natural polymers.</p> <p>7. Decomposition processes of natural polymers. Natural polymers as a component of biomass for energy purposes.</p> <p>8. Biomass conversion processes. Biofuel generations. Liquid and gaseous biofuels from biomass.</p> <p>9. Biochemical conversion of biomass. Photolysis, anaerobic digestion, dark fermentation and photofermentation.</p> <p>10. Thermochemical conversion of biomass. Gasification, hydrothermal liquefaction, pyrolysis, carbonization.</p> <p>11. Biohydrogen and biogas. Composition of liquid and gaseous streams in the processes of producing liquid and gaseous biofuels. Chemical transformations of by-products in the liquid phase. Biorefining.</p> <p>Topics of laboratory classes</p> <p>A Separation of multi-component mixtures</p> <p>1 Separation of a multi-component mixture containing acidic, basic and neutral compounds. Preparation of solutions needed for extraction. Pre-filtration of insoluble ingredients. The use of extraction to separate a mixture into groups of substances: acidic, basic, neutral.</p> <p>2 Carrying out distillation at atmospheric pressure of the components of the neutral fraction previously obtained from the extraction. Crystallization of products obtained from the basic fraction.</p> <p>3 Carrying out vacuum distillation of the components of the acid fraction previously obtained from extraction.</p> <p>B Obtaining biodiesel from used frying fats</p> <p>The aim of this laboratory exercise is to familiarize students with the process of producing biodiesel from frying fats, which are waste from the food industry. Students will gain practical knowledge about the conversion of fats to fatty acid esters and understand the importance of alternative energy sources in the context of sustainable development.</p> <p>C. Conversion of waste biomass to gaseous biofuels.</p> <p>The aim of the laboratory exercise is to familiarize students with unit operations used during pre-treatment, hydrolysis and fermentation of biomass. Students will gain practical knowledge about the conversion of biomass into biohydrogen/biogas/bioethanol and become familiar with statistical and analytical tools for optimizing the processes.</p>		
Prerequisites and co-requisites	Knowledge of basic chemical laws and concepts. Knowledge of basic physical laws and concepts.		

Assessment methods and criteria	Subject passing criteria	Passing threshold	Percentage of the final grade
		60.0%	20.0%
		60.0%	40.0%
		60.0%	20.0%
		60.0%	20.0%
Recommended reading	Basic literature	Robert Thornton Morrison , John Boyd - Chemia Organiczna Tom 1,2,3 PWN 2011	
		John McMurry - Chemia Organiczna Tom 1,2,3 PWN 2000	
	Supplementary literature	Przemysław Mastalerz - Chemia Organiczna PWN 1986	
	eResources addresses	Adresy na platformie eNauczanie: Podstawy chemii organicznej - Moodle ID: 38535 <a href="https://enauczanie.pg.edu.pl/moodle/course/view.php?id=38535">https://enauczanie.pg.edu.pl/moodle/course/view.php?id=38535</a>	
Example issues/ example questions/ tasks being completed			
Work placement	Not applicable		