



Subject card

Subject name and code	Supramolecular chemistry and medicine, PG_00053339						
Field of study	Biomedical Engineering, Biomedical Engineering, Biomedical Engineering						
Date of commencement of studies	October 2024	Academic year of realisation of subject			2024/2025		
Education level	second-cycle studies	Subject group			Optional subject group Specialty subject group Subject group related to scientific research in the field of study		
Mode of study	Full-time studies	Mode of delivery			at the university		
Year of study	1	Language of instruction			Polish		
Semester of study	2	ECTS credits			3.0		
Learning profile	general academic profile	Assessment form			assessment		
Conducting unit	Department of Chemistry and Technology of Functional Materials -> Faculty of Chemistry						
Name and surname of lecturer (lecturers)	Subject supervisor	dr hab. inż. Ewa Wagner-Wysiecka					
	Teachers						
Lesson types and methods of instruction	Lesson type	Lecture	Tutorial	Laboratory	Project	Seminar	SUM
	Number of study hours	15.0	0.0	15.0	0.0	0.0	30
	E-learning hours included: 0.0						
Learning activity and number of study hours	Learning activity	Participation in didactic classes included in study plan	Participation in consultation hours	Self-study	SUM		
	Number of study hours	30	3.0	42.0	75		
Subject objectives	The aim of the course is to familiarize students with the broadly understood aspects of the interdisciplinary field of science - supramolecular chemistry - with particular emphasis on application areas in medicine and related sciences.						
Learning outcomes	Course outcome	Subject outcome			Method of verification		
	[K7_W51] Knows and understands, to an increased extent, selected aspects of chemistry and biochemistry constituting general knowledge in the field of biomedical engineering.	Student defines concepts related to supramolecular chemistry and its applications which is part of the knowledge in the field of biomedical engineering - Student describes techniques and methods used in the field of biomedical engineering - using the achievements of modern supramolecular chemistry			[SW1] Assessment of factual knowledge		
	[K7_U02] can perform tasks related to the field of study as well as formulate and solve problems applying recent knowledge of physics and other areas of science	Student is able to prepare a design concept illustrating the use of modern supramolecular chemistry in medicine			[SU5] Assessment of ability to present the results of task [SU2] Assessment of ability to analyse information [SU1] Assessment of task fulfilment		
	[K7_K02] is ready to provide critical evaluation of received content and to acknowledge the importance of knowledge in solving cognitive and practical problems	The student interprets the results of the obtained research, drawing constructive conclusions in relation to the data presented in the world literature			[SK2] Assessment of progress of work [SK5] Assessment of ability to solve problems that arise in practice		

Subject contents	<p>Lecture: Definition of supramolecular chemistry. Types of complexing molecules; the concepts of ligand, substrate, receptor, host, guest. Types of interactions in supramolecular systems. Examples of synthetic host molecules, complexing compounds (podandss, coronands, cryptands, spherands, calixarenes, hetero- and homo-calixarenes, metalloporphyrins and others). The complexation of cations, anions and molecules. Effects: chelate, macrocyclic and cryptic. Pearson's theory of hard and soft acids and bases (HSAB). Strategy for the synthesis of macrocyclic compounds. Factors promoting cyclization (template effect, preorganization of substrate molecules, dilution method). Self-organization and molecular recognition. Chemistry of macromolecules (polymers), basic properties of macromolecules, types of polymerization reactions. Macrocyclic ligands: cyclodextrines and their complexes, dendrimers, cyclophanes and steroids. The phenomenon of preorganization, systems containing a large number of hydrogen bonds (rosettes, tapes - ribbons, fibers and two-dimensional networks, capsules with hydrogen bonds, gas hydrate clathrates). Supramolecular interactions exemplified with polysaccharide biopolymers (chemistry of chitosan, alginate, starch), protein biopolymers (collagen, fibrinogen). Self-organization of nucleic acids (DNA and RNA chemistry), types of supramolecular interactions occurring in biopolymers and the possible uses of these interactions. Intermediate systems between chemical molecules and cells of living organisms, lipid membranes, microemulsions, micelles, fibers, nanotubes, liquid crystals Examples of the application of supramolecular chemistry in the food and cosmetics industries. Supramolecular systems occurring in nature (biological systems) and their role; examples of natural complexing compounds (antibiotics, siderophores, etc.). Supramolecular biomimetic systems: enzymes, cells, channels. Applications of supramolecular chemistry in medicine: a) diagnostics - optical and fluorescent sensors, logic gates, electrochemical sensors, imaging systems; multifunctional nanoparticles b) therapeutics - chelate therapy, drug delivery and controlled release systems, supramolecular antibiotics. (Nano)Molecular machines.</p> <p>Laboratory:</p> <ol style="list-style-type: none"> 1. Modification of bacterial cellulose for imparting antimicrobial properties. 2. Immobilization techniques of bioactive materials used in medicine - encapsulation polysaccharide-based hydrogel 3. Supramolecular interactions in constructing III generation dressings 4. Supramolecular polymers interactions used in 3D-bioprinting 5. Supramolecular analytical chemistry - applications in biomedical analysis 											
Prerequisites and co-requisites	Knowledge and skills (applies also to the practical part of the subject - laboratory) in the field of organic, analytical, physical chemistry, and biochemistry.											
Assessment methods and criteria	<table border="1"> <thead> <tr> <th data-bbox="456 1227 794 1256">Subject passing criteria</th> <th data-bbox="799 1227 1137 1256">Passing threshold</th> <th data-bbox="1142 1227 1481 1256">Percentage of the final grade</th> </tr> </thead> <tbody> <tr> <td data-bbox="456 1263 794 1314">Performing of the laboratory classes and passing tests</td> <td data-bbox="799 1263 1137 1314">100.0%</td> <td data-bbox="1142 1263 1481 1314">50.0%</td> </tr> <tr> <td data-bbox="456 1321 794 1373">colloquium on the lecture material (written or oral)</td> <td data-bbox="799 1321 1137 1373">51.0%</td> <td data-bbox="1142 1321 1481 1373">50.0%</td> </tr> </tbody> </table>			Subject passing criteria	Passing threshold	Percentage of the final grade	Performing of the laboratory classes and passing tests	100.0%	50.0%	colloquium on the lecture material (written or oral)	51.0%	50.0%
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Example issues/ example questions/ tasks being completed	Describe the basic relationships characteristic for guest-host chemistry. List and illustrate with examples the types of interactions in supramolecular systems (giving examples of systems occurring in nature and synthetic ones). Discuss drug delivery and controlled release systems based on supramolecular interactions.
Work placement	Not applicable

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