

Subject card

Subject name and code	Molecular physics, PG_00064443							
Field of study	Biomedical Engineering, Biomedical Engineering, Biomedical Engineering							
Date of commencement of studies	February 2025		Academic year of realisation of subject			2024/2025		
Education level	second-cycle studies		Subject group			Optional subject group Specialty subject group Subject group related to scientific research in the field of study		
Mode of study	Full-time studies		Mode of delivery			at the university		
Year of study	1		Language of instruction			Polish		
Semester of study	1		ECTS credits			3.0		
Learning profile	general academic profile		Assessment form			assessment		
Conducting unit	Department of Physics of Electronic Phenomena -> Faculty of Applied Physics and Mathematics						cs	
Name and surname of lecturer (lecturers)	Subject supervisor	dr hab. Jan Franz						
	Teachers	dr hab. Jan Franz						
Lesson types and methods of instruction	Lesson type	Lecture	Tutorial	Laboratory	Projec			SUM
	Number of study hours	30.0	15.0	0.0	0.0		0.0	45
	E-learning hours inclu	i		i				+
Learning activity and number of study hours	Learning activity Participation in classes include plan				Self-study SUM		SUM	
	Number of study hours	45 5.0			25.0		75	
Subject objectives	The aim of the course	e is to familiariz	e the student v	vith selected is	sues of	molecu	lar physics.	
Learning outcomes	Course out	Subject outcome			Method of verification			
	[K7_W02] knows and understands, to an increased extent, selected laws of physics and physical phenomena, as well as methods and theories explaining the complex relationships between them, constituting advanced general knowledge in the field of technical sciences related to the field of study		The student knows and understands the basics of molecular physics.			[SW1] Assessment of factual knowledge		
			The student is able to use the acquired knowledge in the applications of molecular spectroscopy.			[SU3] Assessment of ability to use knowledge gained from the subject		
	develop models of proper behaviour in the work and life		The student is able to prepare a presentation on molecular physics in relation to modern trends in biomedical engineering.			[SK4] Assessment of communication skills, including language correctness		

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Subject contents	1. Introduction to the lecture: photons and waves, atomic structure. 2. Fundamentals of quantum mechanics: wave-particle duality, wave function, Schrödinger equation, Heisenberg uncertainty principle, free particle motion, particle in a square-well, tunnelling phenomenon, hydrogen atom, atomic orbital, hydrogen ion. 3. Electric and magnetic properties of molecules. 4. Molecular solid: types of crystal bonds and their characteristics. 5. Interaction of electromagnetic radiation with particles: forms of particle energy, quantization of energy, energy distribution in the state of thermal equilibrium, probability of absorption and emission of radiation, types of spectroscopy. 6. The rotational energy of molecules. 7. The vibrational energy of molecules. 8. The interaction of electromagnetic radiation with vibrating molecules: Raman spectrum. 9. Molecular electronic transitions: electronic states and the energy of electronic states.						
Prerequisites and co-requisites							
Assessment methods and criteria	Subject passing criteria	Passing threshold	Percentage of the final grade				
	presentation	50.0%	20.0%				
	passing the lecture	50.0%	80.0%				
Recommended reading	Basic literature	 Z. Kęcki, "Podstawy spektroskopii molekularnej", Wydawnictwo Naukowe PWN, Warszawa 2013 G. Ślósarek, Biofizyka molekularna, Wydawnictwo Naukowe PWN Warszawa 2011. H. Haken, H. Ch. Wolf, Fizyka molekularna z elementami chemii kwantowej, Wydawnictwo Naukowe PWN Warszawa 1998. 					
	Supplementary literature	ature P. W. Atkins, R. S. Friedman, "Molecular quantum University Press, 1997.					
	eResources addresses	Adresy na platformie eNauczanie:					
Example issues/ example questions/ tasks being completed	The examples of exam questions:						
	Represent and describe the forms of internal energy of molecules. Represent and describe the physical quantities which characterizing the magnetic properties of molecules.						
Work placement	Not applicable						
Work placement							

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