

Subject card

Subject name and code	Chemistry, PG_00041989								
Field of study	Power Engineering								
Date of commencement of studies	October 2024		Academic year of realisation of subject			2024/2025			
Education level	first-cycle studies		Subject group			Obligatory subject group in the field of study			
Mode of study	Full-time studies		Mode of delivery			at the university			
Year of study	1		Language of instruction			English			
Semester of study	1		ECTS credits			3.0			
Learning profile	general academic profile		Assessment form			assessment			
Conducting unit	Department of Chemi	stry and Techn	ology of Functional Materials -> Faculty of Cher				Chemistry		
Name and surname	Subject supervisor	oject supervisor dr hab. inż. Ewa Wagner-Wysie			siecka	a			
of lecturer (lecturers)	Teachers								
Lesson types and methods	Lesson type	Lecture	Tutorial	Laboratory	Project		Seminar	SUM	
of instruction	Number of study hours	30.0	0.0	15.0	0.0		0.0	45	
	E-learning hours included: 0.0								
Learning activity and number of study hours	Learning activity	Participation in classes include plan		Participation in consultation hours		Self-study		SUM	
	Number of study hours	45		7.0		23.0		75	
Subject objectives	The aim of the subject is to learn students the concepts of chemistry and the chemical basis of the processes useful in learning problems of widely understood energetics.								
Learning outcomes	Course outcome		Subject outcome			Method of verification			
	[K6_K01] is aware of the need for training and self-improvement in the profession of energy and the possibility of further education; can think and act in a creative and entrepreneurial manner; can define priorities for the implementation of an individual or group task		Student realizes that the use of knowledge related to chemical transformations is important in improving professional qualifications.			[SK2] Assessment of progress of work [SK3] Assessment of ability to organize work			
	[K6_U10] can use correctly selected methods and measuring devices for determination of basic parameters during the water treatment process and wastewater treatment control; can perform basic laboratory tests leading to the assessment of water quality, pollutant load in wastewater [K6_W03] knows the basics of automation and automatic regulation, knows the principles of the selection of electrical devices, drive systems and their control								
Subject contents Prerequisites and co-requisites	Lecture:The structure of matter. Periodic table, chemical elements. Chemical bonds. Types of the chemical compounds. Chemical reactions. Elements of the thermodynamics and chemical kinetics. Gases, liquids and solids - properties, structure. Solutions. Corrosion. Combustion processes. Laboratory: Reaction kinetics. Qualitative analysis of cation and anions. Water hardness. Conductivity of solutions and electrolysis. Corrosion. Electrochemical series and galvanic cells.								
and co-requisites									

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Association methods	Outsiant manada a mitaria	Deseite a three deals	Demonstrate of the final and a		
Assessment methods	Subject passing criteria	Passing threshold	Percentage of the final grade		
and criteria	Written test on the ground of the lecture	50.0%	50.0%		
	Six tests refering to the laboratory excercises	50.0%	50.0%		
Recommended reading	Basic literature	1. Bielański A.: "Chemia ogólna i nieorganiczna", PWN 2002 2. Cotton F.A., Wilkinson G., Gaus P.L.: "Chemia nieorganiczna. Podstawy.", PWN 2002 3. Sienko M.J., Plane R.A.: "Chemia. Podstawy i zastosowania", WNT 2002 4. Pajdowski L.:" Chemia ogólna", PWN 1999 5. McMurray J.: "Chemia organiczna" PWN 2005 6. Atkins P.W.:" Podstawy chemii fizycznej" PWN 1999 7. Bortel E., Koneczny H.:"Zarys technologii chemicznej", PWN 1992 8. red. Luboch E., Bocheńska M., Biernat J.F. "Chemia ogólna. Ćwiczenia laboratoryjne" Wyd. PG 2003 9. Brown T. E., Eugene LeMay H., Bursten B. E., Murphy C., Woodward P.: Chemistry: The Central Science, 12th Ed. 2011, 10. Pauling L.: General Chemistry, 3rd Ed. 11. S. S. Zumdahl, S. A. Zumdahl: Chemistry 7th Ed.003			
	Supplementary literature	1. Kołos W., Sadlej J.:"Atom i cząsteczka", WNT 2007 2. Atkins P.W.:" Przewodnik po chemii fizycznej", PWN 1997 3. Mastalerz P.:"Chemia organiczna", Wyd. Chemiczne 2002 4. Bogoczek R., Kociołek-Balawejder E.:"Technologia chemiczna organiczna. Surowce i półprodukty." Wyd. AE Wrocław 1992 5. 12. Jess A., Wasserscheid P.: Chemical Technology: An Integral Textbook, 2013			
	eResources addresses Adresy na platformie eNauczanie:				
Example issues/ example questions/ tasks being completed	-				
Work placement	Not applicable				

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