

## Subject card

Subject name and code	Physics, PG_00054684									
Field of study	Biotechnology									
Date of commencement of studies	October 2024		Academic year of realisation of subject			2024/2025				
Education level	first-cycle studies		Subject group			Obligatory subject group in the field of study				
Mode of study	Full-time studies		Mode of delivery			at the university				
Year of study	1		Language of instruction			Polish				
Semester of study	2		ECTS credits			5.0				
Learning profile	general academic profile		Assessment form			exam				
Conducting unit	Zakład Spektroskopii Układów Złożonych -> Instytut Fizyki i Informatyki Stosowanej -> Faculty of Applied Physics and Mathematics									
Name and surname	Subject supervisor	sz Wąsowicz								
of lecturer (lecturers)	Teachers	·								
Lesson types and methods of instruction	Lesson type	Lecture	Tutorial	Laboratory			SUM			
	Number of study hours	15.0	15.0	30.0	0.0	0.0		60		
	E-learning hours included: 0.0									
Learning activity and number of study hours	Learning activity Participation in classes include plan			Participation in consultation hours		Self-study		SUM		
	Number of study hours	60		10.0		55.0		125		
Subject objectives	The main objective of the course is: acquire a certain amount of knowledge of general physics, teach thinking in terms of cause-and-effect relationships and to understand the limitations imposed by the fundamental laws of physics, acquire problem-solving skills encountered in engineering work									
Learning outcomes	Course outcome Subject outcome Method of verification					rification				
	K6_W01		Student can identify and understand physical phenomena and can describe them by the use of theoretical models			[SW1] Assessment of factual knowledge				
	K6_U01					[SU4] Assessment of ability to use methods and tools [SU3] Assessment of ability to use knowledge gained from the subject [SU1] Assessment of task fulfilment				
Subject contents	LECTURES OPTICS. Spectrum of electromagnetic waves. Geometric optics: the laws of light reflection and refraction, prism. Wave optics: polarization, diffraction and interference, diffraction grating. Quantum properties of radiation: thermal radiation, photoelectric effect, photons. ATOMIC PHYSICS. Bohr model of the hydrogen atom. Vector model of the atom, quantum numbers, spin-orbit coupling, spin magnetic resonance. X-rays. De Broglie waves. NUCLEAR PHYSICS. Constituents of the nucleus. Nuclear forces and binding energy. Spin and magnetic moment of nucleus. Nuclear magnetic resonance. Exponential decay law. Applications of radioactive isotopes. Fission and fusion reactions.  TUTORIALS 1. Geometric optics. 2. Wave optics. 3. Thermal radiation. 4. Photoelectric effect. Bohrs model of hydrogen atom. 5. X-rays. Braggs law. De Broglie waves. 6. Nuclear forces and binding energy. Exponential decay law.  LABORATORY 1. MECHANICS: mechanics of particles and rigid bodies, elastic collisions, hydrostatics. 2. GRAVITATIONAL FIELD: acceleration due to gravity on the Earth. 3. MECHANICAL WAVES: mechanical resonance, standing waves. 4. ELECTRIC FIELD: electric field distribution, dielectric constant, capacitors,									
Prerequisites and co-requisites	resistors. 5. MAGNETIC FIELD: magnetic field of the Earth, magnetic force on a current-carrying conductor .  6. OPTICS: refractive index, interference, polarization. 7. ATOMIC PHYSICS: atomic emission spectra.  Knowledge from Physics semester I									
and co-requisites										

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Assessment methods	Subject passing criteria	Passing threshold	Percentage of the final grade			
and criteria	Midterm colloquium	50.0%	20.0%			
	Written exam	50.0%	50.0%			
	Laboratory	100.0%	30.0%			
Recommended reading	Basic literature	Halliday, Resnick, Walker, "Fundamentals of Physics", John Wiley & Sons, Inc. 2001				
	Supplementary literature	2.V.Acosta, C.L.Cowan, B.J.Graham. Essentials of Modern Physics, Harper & Row 1973.				
	eResources addresses	Adresy na platformie eNauczanie:				
Example issues/ example questions/ tasks being completed	Bohr's model of the atom of hydrogen. Bohr orbits. Rydberg formula. Bohr magneton. Calculate the wavelength of the red line of the Balmer series Quantum numbers. Orbital, spin and total angular momentum. Spatial quantization of angular moments					
Work placement	Not applicable					

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