



Subject card

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|---|--|---|----------------------------|---|---|------------|-----|
| Subject name and code | Basics of chemical reactor engineering, PG_00060871 | | | | | | |
| Field of study | Chemical Technology | | | | | | |
| Date of commencement of studies | October 2024 | Academic year of realisation of subject | | | 2026/2027 | | |
| Education level | first-cycle studies | Subject group | | | Obligatory subject group in the field of study | | |
| Mode of study | Full-time studies | Mode of delivery | | | at the university | | |
| Year of study | 3 | Language of instruction | | | Polish | | |
| Semester of study | 6 | ECTS credits | | | 2.0 | | |
| Learning profile | general academic profile | Assessment form | | | assessment | | |
| Conducting unit | Faculty of Chemistry -> Faculties of Gdańsk University of Technology | | | | | | |
| Name and surname of lecturer (lecturers) | Subject supervisor | | dr hab. inż. Jacek Gębicki | | | | |
| | Teachers | | | | | | |
| Lesson types | Lesson type | Lecture | Tutorial | Laboratory | Project | Seminar | SUM |
| | Number of study hours | 15.0 | 15.0 | 0.0 | 0.0 | 0.0 | 30 |
| | E-learning hours included: 0.0 | | | | | | |
| eNauczenie source address: https://enauczenie.pg.edu.pl/2025/user/index.php?id=4947 | | | | | | | |
| Learning activity and number of study hours | Learning activity | Participation in didactic classes included in study plan | | Participation in consultation hours | | Self-study | SUM |
| | Number of study hours | 30 | | 2.0 | | 18.0 | 50 |
| Subject objectives | To introduce students to concepts related to the technological classification of reactors, particularly the description of ideal reactors. To familiarize students with the relationships used to calculate reactant residence times and the degree of reactant conversion in reactors operating under isothermal and adiabatic conditions. To familiarize students with the design equations and heat balance in ideal reactors. To develop students' skills in basic calculations related to reactor theory. | | | | | | |
| Learning outcomes | Course outcome | Subject outcome | | | Method of verification | | |
| | [K6_W04] understands processes occurring in the life cycle of equipment and facilities and has knowledge of mechanical engineering, chemical apparatus, technical thermodynamics and chemical engineering and chemical reactor engineering necessary to analyse technological processes and correctly design installations and systems in the chemical industry | has knowledge of the theory of chemical reactors and knows the operating principle of basic types of reactors in the chemical industry. | | | [SW3] Assessment of knowledge contained in written work and projects | | |
| | [K6_K01] understands the need for continuing education, and is aware of the opportunities to improve professional, personal and social competences | is ready to use the acquired knowledge to continuously improve their qualifications and develop their engineering skills. | | | [SK5] Assessment of ability to solve problems that arise in practice [SK1] Assessment of group work skills | | |
| [K6_U04] performs basic design calculations of selected processes and unit operations, is able to calculate and select the basic apparatus of chemical industry in a process line | able to perform basic calculations using knowledge of design equations and heat balance for various types of ideal reactors. | | | [SU1] Assessment of task fulfilment [SU3] Assessment of ability to use knowledge gained from the subject | | | |

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| Subject contents | <p>Course content – lecture</p> <ol style="list-style-type: none"> 1. Technological classification of reactors. Batch reactor. Perfectly mixed flow reactor. Plug flow reactor. 2. Isothermal and adiabatic processes. 3. Cascade of perfectly mixed flow reactors. Comparison of a perfectly mixed reactor cascade with a plug flow reactor. 4. Design of isothermal, perfectly mixed, or plug flow reactors. 5. The influence of conversion kinetics on the selection of reactor type for single reactions. | | |
| | <p>Course content – exercises</p> <ol style="list-style-type: none"> 1. Mole and mass fractions. Determining the degree of reaction conversion. Conversion and the stoichiometry of chemical reactions. 2. Kinetics of chemical reactions (reaction order, reaction rate constant). Dependence of reaction rate on temperature (Arrhenius equation). 3. Design equation for a batch reactor. Determining residence time and reactor volume. 4. Thermal effects of reactions, heat balance of a batch reactor. 5. Design of a batch reactor (kinetics, reaction order, residence time, reactor volume). 6. Design of a batch reactor (characterization of the reaction environment, material selection, heat). 7. Design of a batch reactor (selection of stirrer power and dimensions). | | |
| Prerequisites and co-requisites | Knowledge of unit operations and processes, the structure and functioning of basic industrial plant components. Principles of conservation of mass, energy, and momentum. Mass and thermal process balancing. | | |
| Assessment methods and criteria | Subject passing criteria | Passing threshold | Percentage of the final grade |
| | exercises - test | 60.0% | 20.0% |
| | exercises - project implementation | 100.0% | 20.0% |
| | lecture - 2 tests | 60.0% | 60.0% |
| Recommended reading | Basic literature | <ol style="list-style-type: none"> 1. A. Burghardt, Bartelmus G., Inżynieria reaktorów chemicznych, PWN 2001. 2. J. Szarawara, J. Piotrowski: Podstawy teoretyczne technologii chemicznej, WNT 2010. 3. S. Kucharski, J. Głowiński: Podstawy obliczeń projektowych w technologii chemicznej, Oficyna Wydawnicza Politechniki Wrocławskiej, 2005 | |
| | Supplementary literature | <ol style="list-style-type: none"> 1. B. Tabiś, W. Żukowski: Przykłady i zadania z zakresu inżynierii reaktorów chemicznych, Politechnika Krakowska 2000 2. K. Schmidt-Szałowski i in.: Technologia Chemiczna, PWN 2013 | |
| | eResources addresses | | |

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| Example issues/ example questions/ tasks being completed | <p>Task 1. In a tank with a usable volume of 0.5 m³ and a stirred tank, the reaction A+B=C is carried out. The initial concentration of component A is 0.05 kmol/m³ and the concentration of component B is twice as high. The rate constant is 0.09 m³/(kmol*h). Determine the conversion rate of A.</p> <p>Task 2. The conversion rate A=2P is described by a first-order kinetic equation (k=2.5 h). Determine the average residence time of the reaction mixture in a cascade of four identical, perfectly mixed reactors if a conversion rate of a=0.9 is required. What residence time would be necessary to achieve this conversion rate in a single stirred reactor?</p> |
| Practical activities within the subject | Not applicable |

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