

Subject card

Subject name and code	Environmental chemistry, PG_00057712								
Field of study	Green Technologies								
Date of commencement of studies	October 2024		Academic year of realisation of subject			2025/2026			
Education level	first-cycle studies		Subject group			Optional subject group Subject group related to scientific research in the field of study			
Mode of study	Full-time studies		Mode of delivery			at the university			
Year of study	2		Language of instruction			Polish			
Semester of study	3		ECTS credits			5.0			
Learning profile	general academic profile		Assessme	Assessment form			exam		
Conducting unit	Department of Inorganic Chemistry -> Faculty of Chemistry								
Name and surname	Subject supervisor		dr hab. Katarzyna Kazimierczuk						
of lecturer (lecturers)	Teachers								
Lesson types and methods of instruction	Lesson type	Lecture	Tutorial	Laboratory	Project		Seminar	SUM	
	Number of study hours	30.0	0.0	30.0	0.0		0.0	60	
	E-learning hours included: 0.0								
Learning activity and number of study hours	Learning activity	Participation in didactic classes included in study plan		Participation in consultation hours		Self-study		SUM	
	Number of study hours	60		10.0		55.0		125	
Subject objectives	Familiarize students with the basics of chemical processes occurring in the natural environment, physical chemistry of the atmosphere, water and soil. Presentation of geochemical cycles of the most important elements in the environment. Familiarization with the most important environmental pollutants, their sources and methods of detection.								

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and so field o to record aspect econol applyi	ognize their non-technical ets, including environmental,	The student notices nontechnical, including environmental, aspects of technologies used in environmental protection. Applies the principles of occupational	[SU2] Assessment of ability to analyse information [SU4] Assessment of ability to use methods and tools				
able to	o make initial assessment of eering solutions and actions	health and safety.	analyse information [SU4] Assessment of ability to				
of soil design enviro techno which knows neutra and w a basi theore types chemi	n and supervision of onmentally friendly ologies and technologies	The student has basic knowledge in the field of soil, air and water protection against pollution and the theoretical basis of methods and types of apparatus used in the analysis of environmental pollution.	[SW1] Assessment of factual knowledge				
of che chemi physic knowle and ut in the and th	W02] has a basic knowledge emistry including general istry, inorganic, organic, cal, analytical, including the ledge necessary to describe inderstand the phenomena hemical processes occurring environment; measurement ne determination of the neters of these processes.	The student has basic knowledge in the field of chemistry necessary to describe and understand phenomena and chemical processes occurring in the natural environment. Knows the basics of the methods used for measuring the level of environmental pollution.	[SW1] Assessment of factual knowledge				
Carbo eleme	Atmospheric chemistry. Aquatic chemistry. Soil chemistry. Persistent organic pollutants in the environment. Carbon cycle. Nitrogen cycle. Phosphorus cycle. Oxygen and sulfur cycle. The role of the chemical elements in living organisms. Heavy metals and micronutrients. Environmental analytics. Methods of measuring the degree of pollution. Remote pollution measurement methods.						
Prerequisites Passed and co-requisites	d course of Inorganic Chemistr	ry					
Assessment methods	Subject passing criteria	Passing threshold	Percentage of the final grade				
and criteria writter	n exam	60.0%	70.0%				
labora	atory reports	60.0%	30.0%				
Recommended reading Basic I	literature	Gary W van Loon and Stephen J Duffy, Environmental Chemistry, Oxford University Press					
Supple	ementary literature	S. Manahan, Environmental Chemistry, CRC Press, 2009					
eReso	eResources addresses Adresy na platformie eNauczanie:						
example questions/	Characterize photochemical smog. List alternative sources of phosphorus. Discuss the carbon cycle in nature.						
Work placement Not ap	Not applicable						

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