



Subject card

Subject name and code	Photophysics and introduction to molecular spectroscopy , PG_00061307						
Field of study	Nanotechnology						
Date of commencement of studies	October 2022	Academic year of realisation of subject			2024/2025		
Education level	first-cycle studies	Subject group			Optional subject group Subject group related to scientific research in the field of study		
Mode of study	Full-time studies	Mode of delivery			at the university		
Year of study	3	Language of instruction			Polish		
Semester of study	6	ECTS credits			2.0		
Learning profile	general academic profile	Assessment form			assessment		
Conducting unit	Institute of Nanotechnology and Materials Engineering -> Faculty of Applied Physics and Mathematics						
Name and surname of lecturer (lecturers)	Subject supervisor	dr hab. inż. Agnieszka Witkowska					
	Teachers	dr hab. inż. Agnieszka Witkowska					
Lesson types and methods of instruction	Lesson type	Lecture	Tutorial	Laboratory	Project	Seminar	SUM
	Number of study hours	15.0	0.0	0.0	0.0	0.0	15
	E-learning hours included: 0.0						
Learning activity and number of study hours	Learning activity	Participation in didactic classes included in study plan		Participation in consultation hours		Self-study	SUM
	Number of study hours	15		2.0		18.0	35
Subject objectives	The aim of the course is to present the main issues related to the interaction between electromagnetic radiation and matter, to discuss photophysical processes and the basics of molecular spectroscopy.						
Learning outcomes	Course outcome		Subject outcome		Method of verification		
	K6_W06		The student acquires knowledge of the optical and photophysical properties of materials and nanomaterials and the correlation of these properties with their structure and other non-optical properties.		[SW1] Assessment of factual knowledge		
	K6_U06		The student acquires knowledge that will enable him/her to present in a simple and understandable way technological and scientific issues and problems related to the properties and applications of nanostructures in photophysical processes.		[SU3] Assessment of ability to use knowledge gained from the subject [SU2] Assessment of ability to analyse information		
	K6_W07		The student acquires knowledge in the field of optical properties of materials and nanomaterials, learns spectroscopic methods of studying the structural and physico-chemical properties of materials and nanomaterials.		[SW1] Assessment of factual knowledge		

Subject contents	<p><b>Lecture:</b></p> <p>1) Introduction. The nature of EM radiation (classical electrodynamics and Maxwell's equations, waveparticle duality ); nature and structure of matter (Waveparticle duality, quantum mechanics, atomic and molecular orbitals, energy states in a molecule); interaction of EM radiation with matter (electric dipole approximation, Einstein coefficients, selection rules for optical transitions).</p> <p>2) Jabłoński diagram of photophysical processes: radiative transitions (fluorescence and phosphorescence, Kasha's rule, mirror symmetry rule for absorption and emission spectra, Stokes shift, quantum efficiency and lifetime of fluorescence and phosphorescence decay), non-radiative transitions (vibrational relaxation, internal conversion, intersystem crossing). Types of luminescence.</p> <p>3) Spectroscopy methods: classification of spectroscopy methods; Beer-Lambert law, transmittance, absorbance, absorption coefficient, attenuation length; measurement methods (continuous wave and Fourier Transform methods); absorption and emission spectrometer; spectrum, basic parameters of spectral lines and their physical meaning. Infrared absorption spectroscopy vs. Raman spectroscopy. UV-Vis spectroscopy.</p>											
Prerequisites and co-requisites	Basic knowledge of electromagnetism, modern physics, atomic and molecular physics and solid state physics.											
Assessment methods and criteria	<table border="1"> <thead> <tr> <th data-bbox="456 640 794 669">Subject passing criteria</th> <th data-bbox="799 640 1137 669">Passing threshold</th> <th data-bbox="1142 640 1481 669">Percentage of the final grade</th> </tr> </thead> <tbody> <tr> <td data-bbox="456 676 794 705">Participation in classes</td> <td data-bbox="799 676 1137 705">0.0%</td> <td data-bbox="1142 676 1481 705">10.0%</td> </tr> <tr> <td data-bbox="456 712 794 741">Written test</td> <td data-bbox="799 712 1137 741">51.0%</td> <td data-bbox="1142 712 1481 741">90.0%</td> </tr> </tbody> </table>			Subject passing criteria	Passing threshold	Percentage of the final grade	Participation in classes	0.0%	10.0%	Written test	51.0%	90.0%
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Participation in classes	0.0%	10.0%										
Written test	51.0%	90.0%										
Recommended reading	Basic literature	<p>[1] D.L.Andrews, Molecular Photophysics and Spectroscopy, Morgan &amp; Claypool Publ.</p> <p>[2] J.Sadlej, Spektroskopia molekularna, WNT, Warszawa (in polish)</p>										
	Supplementary literature	<p>[3] H.Haken, H.Ch.Wolf, Molecular Physics and Elements of Quantum Chemistry, Springer</p> <p>[4] D.L.Pavia i in., Introduction to Spectroscopy, Brooks/Cole</p>										
	eResources addresses	<p>Adresy na platformie eNauczenie:</p> <p>Fotofizyka i podstawy spektroskopii molekularnej - 2025 - Moodle ID: 44058</p> <p><a href="https://enauczenie.pg.edu.pl/moodle/course/view.php?id=44058">https://enauczenie.pg.edu.pl/moodle/course/view.php?id=44058</a></p>										
Example issues/ example questions/ tasks being completed	<ol style="list-style-type: none"> <li>Using Maxwell's equations, prove that light is an electro-magnetic wave.</li> <li>List and describe a few types of molecular orbitals.</li> <li>List and briefly discuss mechanisms of light absorption in the molecule.</li> <li>Describe term symbol which characterize atomic states under Russell-Saunders coupling condition. Discuss the excited Singlet and Triplet state.</li> <li>Explain the phenomena of absorption, spontaneous and stimulated emission. What is the probability of individual processes occurring and what parameter allows to assess this probability.</li> <li>Photophysical vs. photochemical processes state the difference between them.</li> <li>Present the Jabłoński diagram and use it to discuss the basic photophysical processes.</li> <li>Explain Kasha's rule</li> <li>Fluorescence: basic rules, laws, quantum yield and the fluorescence decay time.</li> <li>Spectroscopy: definition, types of spectroscopy methods.</li> <li>Specify and describe physical meaning of the parameters that characterize spectral line shape.</li> <li>Define: transmittance, absorbance and absorption coefficient. Describe the relations between them.</li> <li>Raman spectroscopy: discuss the origin and the idea of the phenomenon and describe shape of the Raman spectra.</li> <li>What is the auxochrome and how it can change the UV-Vis spectrum.</li> </ol>											
Work placement	Not applicable											

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