

## Subject card

Introduction to Biological Modelling Systems, PG_00049378								
Biomedical Engineering  October 2025								
October 2025		Academic year of realisation of subject			2027/2028			
first-cycle studies		Subject group			Optional subject group			
					Subject group related to scientific research in the field of study			
Full-time studies		Mode of delivery			at the university			
3		Language of instruction			Polish			
6		ECTS credits			4.0			
general academic profile		Assessment form			assessment			
Division Of Theoretical Physics And Quantum Information -> Institute Of Physics And Applied Computer Science -> Faculty Of Applied Physics And Mathematics -> Wydziały Politechniki Gdańskiej							Computer	
Subject supervisor d			dr hab. inż. Marta Łabuda					
rs								
type	Lecture	Tutorial	Laboratory	Projec	t	Seminar	SUM	
r of study	15.0	0.0	15.0	15.0		0.0	45	
ng hours inclu	i							
g activity	Participation in classes includ plan			udy	SUM			
r of study	45		4.0				100	
Introduce the students to the basics of computational and theoretical methods necessary to perform simulations to illustrate the properties of molecules. Students will gain the knowledge on the quantum chemistry methods and techniques allowing to use them in practice in simple atomic systems as well as complex biological systems.								
Course outcome			Subject outcome			Method of verification		
[K6_U07] can apply methods of process and function support, specific to the field of study		The student knows the basic concepts and definitions of molecular physics and quantum-mechanical computations. The student knows and distinguishes basic ab initio quantum chemistry methods.			[SU1] Assessment of task fulfilment [SU4] Assessment of ability to use methods and tools			
[K6_W02] knows and understands, to an advanced extent, selected laws of physics and physical phenomena as well as methods and theories explaining the complex relationships between them, constituting the basic general knowledge in the field of technical sciences related to the field of study			The student knows the basics of theory necessary for performing calculations in atomic and molecular systems.			[SW3] Assessment of knowledge contained in written work and projects		
Introduction. Theoretical background of quantum mechanics. Schroedinger equation. An electron and nuclei movements. Adiabatic approximation. Energy of the excitation. Potential energy curves. Introduction to the quantum chemistry methods. The Self-Consistent Field method and LCAO method. Atomic orbital basis sets. Calculation technique. Ab initio methods. Hartree-Fock method, multi-configurational and multi-reference methods. CI, CC and MP2 methods. Application of the methods to diatomic molecules. Quantum chemistry packages: description, structure and characteristics, advantages and disadvantages of the packages. An interpretation and visualisation of the results. Simple dynamics simulation in the chemical reactions. Simulation of the collisions and interactions between atoms in molecules.  Introduction to spectroscopy, basics of informatics.								
	ction. Theoreticents. Adiabation chemistry malculation techoe methods. Cry packages: es. An interpress. Simulation	ction. Theoretical background ents. Adiabatic approximation in chemistry methods. The Sealculation technique. Ab inition the methods. CI, CC and MP2 ry packages: description, stress. An interpretation and visus. Simulation of the collisions.	ction. Theoretical background of quantum ments. Adiabatic approximation. Energy of the nether chemistry methods. The Self-Consistent Falculation technique. Ab initio methods. Hartice methods. CI, CC and MP2 methods. Appry packages: description, structure and chaes. An interpretation and visualisation of the is. Simulation of the collisions and interaction	ction. Theoretical background of quantum mechanics. Schients. Adiabatic approximation. Energy of the excitation. Point chemistry methods. The Self-Consistent Field method an alculation technique. Ab initio methods. Hartree-Fock methods methods. CI, CC and MP2 methods. Application of the pry packages: description, structure and characteristics, ad es. An interpretation and visualisation of the results. Simples. Simulation of the collisions and interactions between at	ction. Theoretical background of quantum mechanics. Schroeding ents. Adiabatic approximation. Energy of the excitation. Potential of the chemistry methods. The Self-Consistent Field method and LCA alculation technique. Ab initio methods. Hartree-Fock method, multiple methods. CI, CC and MP2 methods. Application of the methods ry packages: description, structure and characteristics, advantages. An interpretation and visualisation of the results. Simple dynamis. Simulation of the collisions and interactions between atoms in the collisions.	ction. Theoretical background of quantum mechanics. Schroedinger equal ents. Adiabatic approximation. Energy of the excitation. Potential energy of the excitation protential energy of the excitation characteristics. Adiabatic approximation. Energy of the excitation. Potential energy of the excitation. Potential energy of the excitation. Potential energy of the excitation and LCAO methods alculation technique. Ab initio methods. Hartree-Fock method, multi-configure methods. CI, CC and MP2 methods. Application of the methods to diatory packages: description, structure and characteristics, advantages and cess. An interpretation and visualisation of the results. Simple dynamics simulation of the collisions and interactions between atoms in molecular controls.	ction. Theoretical background of quantum mechanics. Schroedinger equation. An elect cents. Adiabatic approximation. Energy of the excitation. Potential energy curves. Introc in chemistry methods. The Self-Consistent Field method and LCAO method. Atomic or alculation technique. Ab initio methods. Hartree-Fock method, multi-configurational and the methods. CI, CC and MP2 methods. Application of the methods to diatomic moleculary packages: description, structure and characteristics, advantages and disadvantages. An interpretation and visualisation of the results. Simple dynamics simulation in the is. Simulation of the collisions and interactions between atoms in molecules.	

Data wygenerowania: 26.04.2025 11:11 Strona 1 z 2

Assessment methods	Subject passing criteria	Passing threshold	Percentage of the final grade			
and criteria	Lecture-writting exam	50.0%	40.0%			
	Labs-raports 50.0% 60.0%		60.0%			
Recommended reading	Basic literature	Review of the subject in the form of the e-booklet				
		Distance learning materials				
		WN 2005				
		Leach A. ,Molecular Modelling: Principles and applications Longman 1996				
		antum Chemistry McMillan, New				
		Schlick T. ,Molecular Modeling and Simulation Springer 2002				
		Jensen F. ,Introduction to Computational Chemistry, Academic Press 2007				
	Supplementary literature	User manual : www.molpro.net				
		Graphical user interfaces, for example: Gabedit (main page of the developer)				
		Instructions how to use the computational packages				
	eResources addresses	Adresy na platformie eNauczanie:				
Example issues/ example questions/ tasks being completed						
Work placement	Not applicable					

Document generated electronically. Does not require a seal or signature.

Data wygenerowania: 26.04.2025 11:11 Strona 2 z 2