

## 。 GDAŃSK UNIVERSITY OF TECHNOLOGY

## Subject card

Subject name and code	Thermodynamics, PG_00062723								
Field of study	Technologies for Industry 5.0								
Date of commencement of studies	October 2025		Academic year of realisation of subject			2025/2026			
Education level	first-cycle studies		Subject group			Obligatory subject group in the field of study Subject group related to scientific research in the field of study			
Mode of study	Full-time studies		Mode of delivery			at the	at the university		
Year of study	1		Language of instruction			Polish			
Semester of study	2		ECTS credits			5.0	5.0		
Learning profile	general academic profile		Assessmer			assessment			
Conducting unit	Institute Of Nanotechnology And Materials Engineering -> Faculty Of Applied Physics And Mathematics -> Wydziały Politechniki Gdańskiej							athematics ->	
Name and surname of lecturer (lecturers)	Subject supervisor	dr hab. inż. Jakub Karczewski							
	Teachers		Daniel Jaworski						
			Michał Dominów						
			Kinga Kujawska						
			dr hab. inż. Jakub Karczewski						
Lesson types and methods of instruction	Lesson type	Lecture	Tutorial	Laboratory	Projec	t	Seminar	SUM	
	Number of study hours	30.0	15.0	15.0	0.0		0.0	60	
	E-learning hours included: 0.0								
Learning activity and number of study hours	Learning activity	Participation i classes incluc plan		Participation in consultation hours		Self-study		SUM	
	Number of study hours	60		5.0		60.0		125	
Subject objectives	Understanding the basic principles of thermodynamics, Enabling students to use theoretical knowledge in the context of industrial applications, including the design and optimization of technological processes. Developing skills in the analysis and evaluation of the performance of energy systems, including thermodynamic cycles, heat machines and heat transfer processes.								
Learning outcomes	Course outcome		Subject outcome			Method of verification			
	[K6_U01] applies knowledge of mathematics, physics, chemistry, IT tools and other engineering disciplines to solve theoretical, engineering and technological problems		The student understands the			[SU3] Assessment of ability to use knowledge gained from the subject [SU2] Assessment of ability to analyse information			
	[K6_K01] is aware of the need to constantly update and enrich knowledge and practical skills, and improve professional, personal and social competences					[SK5] Assessment of ability to solve problems that arise in practice			
	[K6_W01] demonstrates knowledge and understanding of mathematics, physics, chemistry and IT tools at the level necessary to formulate and solve typical engineering and technological problems		Student has basic knowledge of technical thermodynamics			[SW1] Assessment of factual knowledge [SW3] Assessment of knowledge contained in written work and projects			

Subject contents Prerequisites	<ul> <li>classical mechanics</li> <li>fluid mechanics</li> <li>first law of thermodynamics</li> <li>ideal and real gases</li> <li>second law of thermodynamic medium</li> <li>compressible flows of a thermodynamic medium</li> <li>combustion and heat exchange</li> <li>compressors, fans and refrigerators</li> <li>steam and combustion engines</li> <li>steam-gas power plants and combined heat and power plants</li> <li>Basic knowledge of physics at the secondary school level</li> </ul>						
and co-requisites	 						
Assessment methods	Subject passing criteria	Passing threshold	Percentage of the final grade				
and criteria		50.0%	40.0%				
		50.0%	20.0%				
		50.0%	40.0%				
Recommended reading	Basic literature	<ul> <li>Jan Szargut "Termodynamika" wydawnictwo naukowe PWN Warszawa 2000</li> <li>Andrzej Teodorczyk "Termodynamika Techniczna" Wydawnictwo Szkolne i Pedagogiczne Warszawa 1995</li> </ul>					
	Supplementary literature	-					
	eResources addresses	Adresy na platformie eNauczanie:					
Example issues/ example questions/ tasks being completed	<ul> <li>explain the principle of operation of the Carnot engine</li> <li>calculate the efficiency of the refrigerator during one cycle</li> <li>calculate the change in entropy of gases under the assumed conditions</li> </ul>						
Work placement	Not applicable						

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