



## Subject card

Subject name and code	Laboratory of inorganic technologies, PG_00060874						
Field of study	Laboratorium technologii nieorganicznych						
Date of commencement of studies	October 2025		Academic year of realisation of subject		2027/2028		
Education level	first-cycle studies		Subject group		Obligatory subject group in the field of study		
Mode of study	Full-time studies		Mode of delivery		at the university		
Year of study	3		Language of instruction		Polish		
Semester of study	6		ECTS credits		2.0		
Learning profile	general academic profile		Assessment form		assessment		
Conducting unit	Department of Process Engineering and Chemical Technology -> Faculty of Chemistry -> Faculties of Gdańsk University of Technology						
Name and surname of lecturer (lecturers)	Subject supervisor		dr hab. inż. Marek Lieder				
	Teachers						
Lesson types	Lesson type	Lecture	Tutorial	Laboratory	Project	Seminar	SUM
	Number of study hours	0.0	0.0	30.0	0.0	0.0	30
	E-learning hours included: 0.0						
	eNauczanie source address: <a href="https://enauczenie.pg.edu.pl/2025/course/view.php?id=1102">https://enauczenie.pg.edu.pl/2025/course/view.php?id=1102</a>						
Learning activity and number of study hours	Learning activity	Participation in didactic classes included in study plan		Participation in consultation hours		Self-study	SUM
	Number of study hours	30		2.0		18.0	50
Subject objectives	Students broaden their knowledge of selected inorganic technologies through their own research  in the laboratory and contact with industrial installations.						
Learning outcomes	Course outcome		Subject outcome		Method of verification		
	[K6_U03] Uses chemical knowledge to design compounds, perform physicochemical and analytical measurements, and obtain appropriate sources of information.		Performs physicochemical and analytical measurements in accordance with applicable laboratory procedures. Analyses and interprets measurement results, formulating conclusions in accordance with the principles of chemical methodology. Compiles measurement results, calculates errors and presents results in tabular or graphical form.		[SU1] Ocena realizacji zadania [SU4] Ocena umiejętności korzystania z metod i narzędzi		
	[K6_W05] Has knowledge of electrical engineering, automation and computer science, including the operation of measurement and control systems		Knows the principles of operation of sensors, transducers and data acquisition systems used in the measurement of physicochemical quantities. Has knowledge of the use of IT tools for monitoring, visualising and analysing process data.		[SW1] Ocena wiedzy faktograficznej		
	[K6_K02] is aware of the responsibility for his/her work and is ready to work in a team and share responsibility for common tasks.		Complies with the rules of cooperation and safety in the laboratory team. Responsibly carries out assigned experimental tasks, ensuring reliability and safety at work. Cooperates with other members of the group in analysing results and preparing a joint report.		[SK1] Ocena umiejętności pracy w grupie [SK4] Ocena umiejętności komunikacji, w tym poprawności językowej [SK5] Ocena umiejętności rozwiązywania problemów występujących w praktyce		

Subject contents	Course content – laboratory Inorganic technology in practice: combustion, exhaust gas purification and water management at the  Gdańsk combined heat and power plant field classes  Obtaining superphosphate,  Carbon monoxide capture (IV),  Hydrogen production by alkaline water electrolysis  Inorganic chemistry processes in industrial technology field classes at the Malbork Sugar Factory  Inorganic technology in the paper industry field laboratory classes at Mondi Świecie Sp. z o.o.  Obtaining sodium chlorate (I)		
Prerequisites and co-requisites	Inorganic technology, lecture. Passed exam in technology: soda, sulphuric acid, phosphoric acid  and artificial fertilisers, nitrogen compounds (ammonia, nitric acid, urea, ammonium nitrate), fuel combustion, water treatment, chlorine production.		
Assessment methods and criteria	Subject passing criteria	Passing threshold	Percentage of the final grade
	Knowledge tests	60.0%	50.0%
	Lab reports	0.0%	50.0%
Recommended reading	Basic literature	1. Schmidt-Szawłowski, K; Szafran, M.; Bobryk, E.; Sentek, J: Technologia Chemiczna. Przemysł Nieorganiczny, PWN, Warszawa, 2013. 2. Bretsznajder S., Podstawy ogólne technologii chemicznej, WNT, Warszawa, 1973 3. Kępiński J., Technologia chemiczna nieorganiczna, PWN, Warszawa, 1984. 4. Bortel E., Koneczny H , Zarys technologii chemicznej, PWN, Warszawa, 1992.	
	Supplementary literature	Praca zbiorowa, Soda i produkty towarzyszące, WNT, Warszawa, 1978. Dylewski R., Gnot W., Gonet M., Elektrochemia przemysłowa, Wydawnictwo Politechniki Śląskiej, Gliwice, 1999. Głowiński J. (Red.), Przykłady i zadania do przedmiotu Podstawy Technologii Chemicznej, Wydawnictwo Politechniki Wrocławskiej, Wrocław, 1991.	
	eResources addresses		
Example issues/ example questions/ tasks being completed	1. Draw a schematic diagram of the process of obtaining NPK fertiliser.  2. Describe the advantages and disadvantages of obtaining hydrogen by electrolysis.  3. Present the most important chemical and technological challenges in the production of chlorate(I).		
Practical activities within the subject	Not applicable		

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