

Subject card

| Field of study Date of commencement of studies Education level first-cycle studies Subject group October 2025 Academic year of realisation of subject Education level first-cycle studies Subject group Optional subject group Subject group research in the field of study Mode of study Full-time studies Mode of delivery at the university Year of study Semester of study Semester of study ECTS credits Semester of study Department of Corrosion and Electrochemistry -> Faculty of Chemistry -> Faculties of Gdańsk University of Technology Name and surname of lecturer (lecturers) Lesson types Lesson types Lesson type Lesture Lesson types Learning activity and number of study hours Number of study Number of stud | Subject name and code | Introduction to Corrosion Protection, PG_00060887 | | | | | | | |
|--|---------------------------------|--|-------------------------------|--|----------------|-------------------------------|-------------------------------------|-------------|-----|
| Date of commencement of studies | | | | | | | | | |
| Mode of study | Date of commencement of | October 2025 | | | | | 2027/2028 | | |
| Year of study 3 | Education level | first-cycle studies | | Subject group | | | Subject group related to scientific | | |
| Semester of study | Mode of study | Full-time studies | | Mode of delivery | | | • | | |
| Learning profile general academic profile Assessment form assessment Conducting unit Department of Corrosion and Electrochemistry > Faculty of Chemistry > Faculties of Gdańsk University of Technology Name and surname of lecturer (lecturers) Lesson types Lesson types Lecture Tutorial Laboratory Project Seminar SUM Number of study hours E-learning hours included: 0.0 eNauczanie source address: https://enauczanie.pg.edu.pl/moodle/course/view.php?id=46562 Learning activity and number of study hours E-learning activity classes included in study plan Number of study hours Subject objectives The aim of this course is to introduce students to various types of corrosion and to identify methods for preventing them. Students will also become familiar with the general principles that should guide engineers with the peneral principles and technologies in the materials and technologies in the material protection. R6 K03 Understands the need for continuous improvement of protecting | Year of study | 3 | | · | | | Polish | | |
| Conducting unit Conducting unit Department of Corrosion and Electrochemistry -> Faculties of Gdańsk University of Technology Subject currer (lecturers) Lesson types Learning hours included: 0.0 | Semester of study | 5 | | | | | 3.0 | | |
| Technology Subject supervisor dr hab. in2. Stefan Krakowiak | Learning profile | general academic profile | | | | | assessment | | |
| Lesson types | Conducting unit | Department of Corrosion and Electrochemistry -> Faculty of Chemistry -> Faculties of Gdańsk University | | | | | University of | | |
| Lesson type | | Subject supervisor | dr hab. inż. Stefan Krakowiak | | | | | | |
| Number of study hours 15.0 0.0 15.0 0.0 0.0 0.0 30 | of lecturer (lecturers) | Teachers | | | | | | | |
| E-learning hours E-learning hours included: 0.0 eNauczanie source address: https://enauczanie.pg.edu.pl/moodle/course/view.php?id=46562 | Lesson types | | | | | | | | |
| eNauczanie source address: https://enauczanie.pg.edu.pl/moodle/course/view.php?id=46562 Learning activity and number of study hours Learning activity I carricipation in didactic classes included in study plan Number of study and number of study hours Subject objectives The aim of this course is to introduce students to various types of corrosion and to identify methods for preventing them. Students will also become familiar with the general principles that should guide engineers involved in protecting steel structures against corrosion. Course outcome Subject outcome [K6_W07] Has knowledge of raw materials and technologies in the chemicaland polymer industries, also covering issues of corrosion of construction materials and learning and knows the oportunities to improve professional, personal and social competences, and is able to think and act in an entrepreneurial manner. Subject contents Course content – lecture A practical introduction to corrosion processes. Protection of steel structures with organic coatings. Environmental modification as a method of reducing corrosion. Electrochemical protection. Galvanic coatings. Course content – lectoure as method of reducing corrosion. Electrochemical protection. Galvanic coatings. The action of corrosion cells. The action of ocrosion coatings. The action of corrosion coatings. Environmental modification or conversion coatings. Environmental modification or corrosion coatings. | | | 15.0 | 0.0 | 15.0 | 0.0 | | 0.0 | 30 |
| Learning activity and number of study hours | | E-learning hours inclu | ıded: 0.0 | | | 1 | | | |
| and number of study hours Number of study hours Number of study hours | | eNauczanie source a | ddress: https:// | enauczanie.pg | .edu.pl/moodle | course | /view.pl | hp?id=46562 | |
| The aim of this course is to introduce students to various types of corrosion and to identify methods for preventing them. Students will also become familiar with the general principles that should guide engineers involved in protecting steel structures against corrosion. Course outcome | | Learning activity | classes includ | | 1 ' | | Self-study | | SUM |
| preventing them. Students will also become familiar with the general principles that should guide engineers involved in protecting steel structures against corrosion. Course outcome | | | 30 | | 5.0 | | 40.0 | | 75 |
| K6_W07] Has knowledge of raw materials and technologies in the chemicaland polymer industries, also covering issues of corrosion of construction materials and is able to propose a method of protecting them against corrosion. K6_K03] Understands the need for continuous learning and knows the opportunities to improve professional, personal and social competences, and is able to think and act in an entrepreneurial manner. Subject contents | Subject objectives | preventing them. Students will also become familiar with the general principles that should guide engineers | | | | | | | |
| materials and technologies in the chemicaland polymer industries, also covering issues of corrosion and material protection. K6_K03 Understands the need for continuous learning and knows the opportunities to improve professional, personal and social competences, and is able to think and act in an entrepreneurial manner. Course content - lecture A practical introduction to corrosion processes. Protection of steel structures with organic coatings. Environmental modification as a method of reducing corrosion. Subject contents Course content - laboratory Structure of metals and alloys - metallography. The action of corrosion coatings. The study and application of organic coatings. The study and application of electroplated coatings. The study and application of electroplated coatings. Surface modification - conversion coatings. Knowledge of the basics of electrochemistry and physical chemistry. Corrosion not control control coatings. Control control coatings. Surface modification - conversion coatings. Control coatings. Control coatings. Control coat | Learning outcomes | Course out | come | Subject outcome | | | Method of verification | | |
| for continuous learning and knows the opportunities to improve professional, personal and social competences, and is able to think and act in an entrepreneurial manner. Course content – lecture A practical introduction to corrosion processes. Protection of steel structures with organic coatings. Environmental modification as a method of reducing corrosion. Electrochemical protection. Galvanic coatings. Course content – laboratory Structure of metals and alloys - metallography. The action of corrosion cells. The study and application of organic coatings. The application of electroplated coatings. The application of electroplated coatings. Surface modification - conversion coatings. Knowledge of the basics of electrochemistry and physical chemistry. | | materials and technologies in the chemicaland polymer industries, also covering issues of corrosion | | corrosion of construction materials and is able to propose a method of protecting them against | | | | | |
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| Prerequisites Knowledge of the basics of electrochemistry and physical chemistry. | Subject contents | A practical introduction to corrosion processes. Protection of steel structures with organic coatings. Environmental modification as a method of reducing corrosion. Electrochemical protection. Galvanic coatings. Course content – laboratory Structure of metals and alloys - metallography. The action of corrosion cells. The study and application of organic coatings. The application of electroplated coatings. | | | | | | | |
| and do regulated | | | | | | | | | |
| | Assessment methods and criteria | Subject passin | Passing threshold | | | Percentage of the final grade | | | |
| | | , , , | | - | | | | | |
| Passing the laboratory 100.0% 50.0% | | I | | | | | | | |

Data wygenerowania: 27.11.2025 13:44 Strona 1 z 2

| Recommended reading | Basic literature | R. Juchniewicz, Anti-corrosion technology parts I and II; P. A. Schweitzer, Corrosion-of-Linings-Coatings-Cathodic-and-Inhibitor-Protection-and-Corrosion-Monitoring, CRC Press, Taylor & Francis Group, Amy Forsgren, Corrosion_Control_Through_Organic_Coatings, CRC Press, Taylor & Francis Group, | | | | |
|--|--|---|--|--|--|--|
| | Supplementary literature | K. Żakowski, K. Darowicki - Cathodic Protection, Wydawnictwo PG A. Miszczyk, M.Szociński, K. Darowicki, Paint coatings for anticorrosion protection. Application guidelines and quality control., Wydawnictwo PG | | | | |
| | eResources addresses | | | | | |
| Example issues/ example questions/ tasks being completed | cathodic protection; coating protection; corrosive environment modifications; structure design with corrosion hazards in mind. | | | | | |
| Practical activites within the subject | Not applicable | | | | | |

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