



Subject card

Subject name and code	Organometallic chemistry, PG_00069253						
Field of study	Chemistry						
Date of commencement of studies	February 2025	Academic year of realisation of subject			2025/2026		
Education level	second-cycle studies	Subject group					
Mode of study	Full-time studies	Mode of delivery			at the university		
Year of study	2	Language of instruction			Polish		
Semester of study	3	ECTS credits			3.0		
Learning profile	general academic profile	Assessment form			assessment		
Conducting unit	Department of Inorganic Chemistry -> Faculty of Chemistry -> Faculties of Gdańsk University of Technology						
Name and surname of lecturer (lecturers)	Subject supervisor	dr hab. inż. Rafał Grubba					
	Teachers						
Lesson types	Lesson type	Lecture	Tutorial	Laboratory	Project	Seminar	SUM
	Number of study hours	15.0	0.0	20.0	0.0	10.0	45
	E-learning hours included: 0.0						
Learning activity and number of study hours	Learning activity	Participation in didactic classes included in study plan	Participation in consultation hours		Self-study		SUM
	Number of study hours	45	5.0		25.0		75
Subject objectives	The aim of the course is to familiarize students with the structure, properties, and applications of coordination compounds, with particular emphasis on organometallic compounds. The program focuses on developing the ability to design new compounds and develop innovative strategies for their preparation. The course includes a series of lectures, seminars, and laboratory classes, during which issues related to the importance of organometallic compounds in chemical synthesis, catalysis, materials chemistry, and medicine will be discussed. Both teamwork and individual collaboration with the teacher promote the development of practical technical and analytical skills, as well as the improvement of skills in presenting research results and preparing reports on experiments.						

Learning outcomes	Course outcome	Subject outcome	Method of verification
	[K7_U02] prepares detailed documentation of the results of independently conducted experiments and analyzes the obtained results, uses professional vocabulary with understanding and prepares and communicates information	The student is able to prepare a report on the performed syntheses of organometallic compounds.	[SU5] Assessment of ability to present the results of task [SU4] Assessment of ability to use methods and tools [SU3] Assessment of ability to use knowledge gained from the subject [SU2] Assessment of ability to analyse information [SU1] Assessment of task fulfilment
	[K7_K02] is able to cooperate and work in a group, taking on different roles	The student is aware of the importance of teamwork in carrying out tasks related to the synthesis of organometallic compounds.	[SK5] Assessment of ability to solve problems that arise in practice [SK4] Assessment of communication skills, including language correctness [SK3] Assessment of ability to organize work [SK2] Assessment of progress of work [SK1] Assessment of group work skills
	[K7_W04] indicates methods for the synthesis of chemical compounds with defined properties	The student has the knowledge necessary to design organometallic compounds with a given structure and properties using advanced methods of inorganic and organic synthesis.	[SW3] Assessment of knowledge contained in written work and projects [SW2] Assessment of knowledge contained in presentation [SW1] Assessment of factual knowledge
	[K7_U82] is able to proficiently obtain and process information related to field of study and academic environment in foreign language at B2+ level of the Common European Framework of Reference for Languages (CEFR)	The student is able to use databases and English-language publications on organometallic compounds as a basis for preparing reports and presentations.	[SU5] Assessment of ability to present the results of task [SU2] Assessment of ability to analyse information
	[K7_U05] analyzes the functioning of devices, equipment and technological lines used in laboratories and the chemical industry	The student is able to use the Schlenk technique to synthesize organometallic compounds and operate a vacuum-argon line and a glove box.	[SU4] Assessment of ability to use methods and tools [SU3] Assessment of ability to use knowledge gained from the subject [SU1] Assessment of task fulfilment
	[K7_U03] plans and performs the synthesis of chemical compounds with the required properties	s able to perform multi-step syntheses of organometallic compounds under anhydrous and oxygen-free conditions using the Schlenk technique.	[SU4] Assessment of ability to use methods and tools [SU3] Assessment of ability to use knowledge gained from the subject [SU2] Assessment of ability to analyse information [SU1] Assessment of task fulfilment

Subject contents	<p>Course content – lecture</p> <p>LECTURE:</p> <p>The course covers the basics of coordination and organometallic chemistry. The first part discusses issues related to the structure, nomenclature, isomerism, and physicochemical properties of d-block metal complexes. The second part is devoted to selected classes of ligands and organometallic compounds, along with the basics of their synthesis.</p> <p>Topics covered in the lecture:</p> <p>1. Coordination compounds:</p> <ul style="list-style-type: none"> • Basic concepts • Nomenclature of complex compounds • Equilibrium in solutions of complex compounds • Isomerism of complex compounds • Crystal field theory • Color and magnetic properties of d-block metal complexes <p>2. Organometallic compounds:</p> <ul style="list-style-type: none"> • Phosphines as ligands • N-heterocyclic carbenes as ligands • Metal carbonyls • Alkenes as ligands • Aromatic hydrocarbons as ligands • Organolithium and organomagnesium compounds • Organoboron compounds • Organosilicon compounds • Basics of preparation of organometallic compounds <hr/> <p>Course content – laboratory</p> <p>LABORATORY:</p> <p>Classes include the synthesis of transition metal compounds containing selected classes of ligands, such as cyclopentadienyl, olefin, phosphine and carbene ligands, as well as their identification by spectroscopic methods (NMR, UV-Vis, IR). They are practical in nature and are aimed at independent planning and implementation of synthetic experiments. The selection of specific syntheses is made in consultation with students, taking into account their interests and current technical possibilities. The summary of the series of meetings is a prepared concise report describing the experiments and analyses performed.</p> <hr/> <p>Course content – seminar</p> <p>SEMINAR:</p> <p>The classes aim to develop skills in interpreting scientific literature in the field of organometallic chemistry. Students learn to critically evaluate publications, analyze the data contained in them, and present research results in a clear and objective manner. The work is based on independent selection and discussion of a scientific article from a renowned journal from the JCR or Scopus list that falls within the topic indicated by the instructor, and then on a joint discussion.</p> <p>Topics discussed during the seminar:</p> <ul style="list-style-type: none"> • Metal complexes with phosphines (synthesis and catalytic properties) • Organophosphorus and organoarsenic compounds (synthesis and application as plant protection agents or in medicine) • Metal carbonyls (synthesis and catalytic properties; application in medicine) • Metal complexes with unstable carbenes (synthesis and catalytic properties) • Metal and nonmetal complexes with N-heterocyclic carbenes (synthesis and catalytic properties, application in medicine) • Metal cyclopentadienyl complexes (synthesis and catalytic properties; application in medicine) • Organolithium compounds (synthesis) • Organogranium compounds (synthesis) • Organoboron compounds (synthesis and catalytic properties) • Organaluminium compounds (synthesis and catalytic properties) • Organosilicon compounds (synthesis and application) • Nobel Prizes in Chemistry organometallic • Metal clusters - discussion of selected examples • Organometallic polymers - discussion of selected examples 														
Prerequisites and co-requisites															
Assessment methods and criteria	<table border="1" style="width: 100%; border-collapse: collapse;"> <thead> <tr> <th style="width: 33%;">Subject passing criteria</th> <th style="width: 33%;">Passing threshold</th> <th style="width: 33%;">Percentage of the final grade</th> </tr> </thead> <tbody> <tr> <td>Seminar - presentation</td> <td>50.0%</td> <td>20.0%</td> </tr> <tr> <td>Laboratory – colloquium, report</td> <td>50.0%</td> <td>50.0%</td> </tr> <tr> <td>Lecture - colloquium</td> <td>50.0%</td> <td>30.0%</td> </tr> </tbody> </table>			Subject passing criteria	Passing threshold	Percentage of the final grade	Seminar - presentation	50.0%	20.0%	Laboratory – colloquium, report	50.0%	50.0%	Lecture - colloquium	50.0%	30.0%
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<p>Example issues/ example questions/ tasks being completed</p>	<ul style="list-style-type: none"> Phosphines and amines are important ligands in coordination chemistry. Give one example each for a phosphine and an amine. Give their names, Lewis structure, molecular shape, and nitrogen-phosphorus hybridization. Determine whether these compounds oxidize in the presence of air. Write the Lewis structure of the oxidation products. Describe how phosphines and amines bind to transition metal ions (describe similarities and differences in the nature of the bonds). The following pairs of complexes are given. Determine which of the complexes is more stable in a given pair using the theory of soft and hard Lewis acids. Justify your answer. Give names for the complexes listed. <p>[FeF₆]³⁻ i [FeCl₆]³⁻</p> <p>[HgI₄]²⁻ i [HgCl₄]²⁻</p> <p>[Pt(NEt₃)₄] i [Pt(PEt₃)₄]</p> <ul style="list-style-type: none"> Describe the structure and methods for obtaining NHC-carbenes using a selected carbene as an example. A student, while synthesizing a compound sensitive to oxidation and hydrolysis, needs to transfer a large amount of solution from one reaction flask to another. Help the student: write what equipment is needed and describe the procedure for transferring a large amount of solution under the required conditions. You can make an illustrative drawing.
<p>Practical activities within the subject</p>	<p>Not applicable</p>

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