



## Subject card

Subject name and code	, PG_00057504						
Field of study	Nanotechnology						
Date of commencement of studies	October 2023		Academic year of realisation of subject		2025/2026		
Education level	first-cycle studies		Subject group		Optional subject group Subject group related to scientific research in the field of study		
Mode of study	Full-time studies		Mode of delivery		at the university		
Year of study	3		Language of instruction		Polish		
Semester of study	6		ECTS credits		3.0		
Learning profile	general academic profile		Assessment form		assessment		
Conducting unit	Division of Electrochemistry and Surface Physical Chemistry -> Institute of Nanotechnology and Materials Engineering -> Faculty of Applied Physics and Mathematics -> Faculties of Gdańsk University of Technology						
Name and surname of lecturer (lecturers)	Subject supervisor		prof. dr hab. inż. Jacek Ryl				
	Teachers		prof. dr hab. inż. Jacek Ryl				
Lesson types	Lesson type	Lecture	Tutorial	Laboratory	Project	Seminar	SUM
	Number of study hours	30.0	0.0	15.0	0.0	0.0	45
	E-learning hours included: 0.0						
Learning activity and number of study hours	Learning activity	Participation in didactic classes included in study plan		Participation in consultation hours		Self-study	SUM
	Number of study hours	45		3.0		27.0	75
Subject objectives	The aim of the course is to familiarize students with the role of electrochemical processes in the world of science and industry, including in particular the possibilities of using electrode phenomena in practice, e.g. in electricity storage technologies, mechanisms of catalyzing chemical processes, mechanisms of electrochemical sensors operation, anti-corrosion technologies, water and wastewater treatment technologies, synthesizing thin-film systems, etc. Electrochemical measurement techniques will be presented and discussed as part of the course.						
Learning outcomes	Course outcome		Subject outcome		Method of verification		
	K6_W07		The student has knowledge of the electrical and electrochemical properties of selected materials and nanomaterials, resulting from the methods of synthesis, modification and environmental and other conditions		[SW3] Assessment of knowledge contained in written work and projects [SW1] Assessment of factual knowledge		
	K6_W06		The student has knowledge about the influence of the structural and chemistry properties of solids on the electrochemical properties and the process of electric charge transfer.		[SW1] Assessment of factual knowledge		
	K6_K05		The student is able to analyze the conducted experiments and present a report summarizing the obtained state of knowledge about the tested materials.		[SK2] Assessment of progress of work [SK4] Assessment of communication skills, including language correctness		
	K6_U06		The student understands the limitations and advantages resulting from the structure of nanomaterials and their applications in electrochemical processes.		[SU2] Assessment of ability to analyse information [SU3] Assessment of ability to use knowledge gained from the subject		

Subject contents	Course content – lecture <ul style="list-style-type: none"><li>• Fundamentals of electrochemistry</li><li>• DC measurements</li><li>• AC measurements</li><li>• Electroanalysis, electrochemical sensors</li><li>• Electrochemical energy storage devices</li><li>• Fuel cells</li><li>• Photo and electrocatalysis</li><li>• Electrochemical water treatment</li><li>• Corrosion and protection against corrosion</li><li>• Electrochemical techniques for applying thin layers</li></ul>		
Prerequisites and co-requisites	Knowledge of the structural properties of materials, solid state physics, chemistry, surface physico-chemistry. Basics in electrical engineering and physical chemistry are valuable.		
Assessment methods and criteria	Subject passing criteria	Passing threshold	Percentage of the final grade
	Laboratory reports	60.0%	50.0%
	Lecture exam	60.0%	50.0%
Recommended reading	Basic literature	P. Atkins - Chemia Fizyczna  K. Pigoń, Z. Ruziewicz - Chemia Fizyczna  A. Czerwiński - Akumulatory, baterie, ogniwa	
	Supplementary literature	Publications in journals from the ISI list, presented during lectures	
	eResources addresses		
	Example issues/ example questions/ tasks being completed	What is the role of each electrode in the measuring system? The role of diffusion in electrochemical processes Describe the mechanisms of selected forms of anti-corrosion protection Why are lithium ion batteries the most widely used today, what are the alternatives? Diversify anodic and cathodic electrochemical coating technologies	
Practical activities within the subject	Not applicable		

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