



Subject card

Subject name and code	General and inorganic chemistry , PG_00063390						
Field of study	Technologies for Industry 5.0						
Date of commencement of studies	October 2026	Academic year of realisation of subject			2026/2027		
Education level	first-cycle studies	Subject group			Obligatory subject group in the field of study		
Mode of study	Full-time studies	Mode of delivery			at the university		
Year of study	1	Language of instruction			Polish		
Semester of study	1	ECTS credits			4.0		
Learning profile	general academic profile	Assessment form			exam		
Conducting unit	Department of Chemistry and Technology of Functional Materials -> Faculty of Chemistry -> Faculties of Gdańsk University of Technology						
Name and surname of lecturer (lecturers)	Subject supervisor	prof. dr hab. inż. Ewa Wagner-Wysiecka					
	Teachers						
Lesson types	Lesson type	Lecture	Tutorial	Laboratory	Project	Seminar	SUM
	Number of study hours	30.0	15.0	0.0	0.0	0.0	45
	E-learning hours included: 0.0						
Learning activity and number of study hours	Learning activity	Participation in didactic classes included in study plan		Participation in consultation hours		Self-study	SUM
	Number of study hours	45		5.0		50.0	100
Subject objectives	The aim of the course is to provide knowledge of general and inorganic chemistry in connection with technologies used in industry, in particular those important for information technology. Selected issues of experimental planning and optimisation of modern technological processes will also be presented using selected examples.						
Learning outcomes	Course outcome	Subject outcome			Method of verification		
	[K6_U01] applies knowledge of mathematics, physics, chemistry, IT tools and other engineering disciplines to solve theoretical, engineering and technological problems	Student can write and balance chemical equations, can perform basic chemical calculations, can critically evaluate the obtained results and estimate their uncertainty			[SU1] Assessment of task fulfilment [SU2] Assessment of ability to analyse information [SU3] Assessment of ability to use knowledge gained from the subject		
	[K6_W01] demonstrates knowledge and understanding of mathematics, physics, chemistry and IT tools at the level necessary to formulate and solve typical engineering and technological problems	Student knows the basic groups of chemical compounds, understands the relationship between the structure and properties of chemical compounds, describes their properties, knows the basic types of chemical reactions, lists the basic technologies crucial for the development of information technology and other branches of technology			[SW1] Assessment of factual knowledge		

Subject contents	<p>Course content – lecture</p> <p>Lecture</p> <p>1) Periodic table of elements (2h). 2) Basic chemical laws and concepts (1h). 3) Chemical bonding in classical description and quantum mechanics (3h). 4) Types of chemical reactions, redox reactions, coordination compounds formation (2h). 5) Theories of acids and bases (1h). 6) Fundamentals of thermodynamics and chemical kinetics; industrial aspects (3h). 7) Physicochemical properties of solutions. Water in industry. (2h). 8) Elements of electrochemistry (cells, electrolysis, corrosion, electroplating) (2h). 9) Introduction to industrial technologies, issues of process optimisation, contemporary technologies and sustainable development; computerisation and automation, intelligent infrastructure in the chemical industry (2h). 10) Selected s block elements. Alkali and alkaline earth metals - their properties, compounds, production and application in energy storage and conversion technologies (2h). 11) Selected p block elements: 13 group of elements - their properties, compounds, obtaining and application including: aluminium, production, application, recycling (2h). 12) Selected p block elements: 14 group of elements - their properties, compounds, obtaining and application in semiconductor technologies (2h). 13) Selected p block elements: 15 group of elements - their properties, compounds, obtaining and application of selected technologies, with special emphasis on bismuth (2h). 14) Selected p block elements: 16 group of elements - their properties, compounds, obtaining and application in technology and industry (1h). 15) Selected p block elements: 17 group of elements (halogens) - their properties, compounds, obtaining and application in technology and industry (1h). 16) Selected d block elements - their properties, compounds, obtaining and application in high-tech technologies (2h).</p> <p>Exercises</p> <p>1) Planning an experiment (2h). 2) Measurement and estimation of experimental uncertainty (1h). 3) Statistical methods in the analysis of experimental results (2h). 4) Basic chemical laws, abundance of matter (1h). 5) Solutions - methods of expressing concentrations (1h). 6) Colloquium (1h). 7) Equilibria in electrolyte solutions (dissociation, strong and weak electrolytes, degree and constant of dissociation) (2h). 8) Aqueous solutions, pH. Buffer solutions (2h). 9) Electrochemical calculations (cells, electrolysis, Faraday's law) (1h). 10) Energy effects of physical processes and chemical reactions. Thermochemical calculations (1h). 11) Colloquium (1h)</p>									
Prerequisites and co-requisites	Basic knowledge of mathematics, physics and chemistry required.									
Assessment methods and criteria	<table border="1"> <thead> <tr> <th data-bbox="456 1043 794 1070">Subject passing criteria</th> <th data-bbox="799 1043 1137 1070">Passing threshold</th> <th data-bbox="1142 1043 1481 1070">Percentage of the final grade</th> </tr> </thead> <tbody> <tr> <td data-bbox="456 1077 794 1104">Two colloquia from exercises</td> <td data-bbox="799 1077 1137 1104">51.0%</td> <td data-bbox="1142 1077 1481 1104">40.0%</td> </tr> <tr> <td data-bbox="456 1111 794 1137">Written exam</td> <td data-bbox="799 1111 1137 1137">51.0%</td> <td data-bbox="1142 1111 1481 1137">60.0%</td> </tr> </tbody> </table>	Subject passing criteria	Passing threshold	Percentage of the final grade	Two colloquia from exercises	51.0%	40.0%	Written exam	51.0%	60.0%
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Written exam	51.0%	60.0%								

Recommended reading	Basic literature	<p>Białański A. , Podstawy chemii nieorganicznej (t.1 oraz t.2), Wydawnictwo Naukowe PWN, Warszawa 2024</p> <p>Cox P.A., Krótkie wykłady: chemia nieorganiczna. Wydawnictwo Naukowe PWN, Warszawa 2024</p> <p>Jones Loretta, Atkins Peter, Leroy Laverman, Chemia ogólna. Wydawnictwo Naukowe PWN, Warszawa 2020.</p> <p>Bobryk Ewa, Schmidt-Szałowski Krzysztof, Sentek Jan, Szafran Mikołaj, Technologia chemiczna. Przemysł nieorganiczny. Wydawnictwo Naukowe PWN, Warszawa 2023</p> <p>Bortel Edward, Konieczny Henryk, Zarys technologii chemicznej, Wydawnictwo Naukowe PWN, Warszawa 1992.</p> <p>Piotrowski Jerzy, Szarawara Józef, Podstawy teoretyczne technologii chemicznej. Wydawnictwa Naukowo-Techniczne, Warszawa, 2020.</p> <p>M. Korzyński, "Metodyka eksperymentu ,Planowanie, realizacja i statystyczne opracowanie wyników eksperymentów technologicznych", Wydawnictwo Naukowe PWN, Warszawa 2021</p> <p>Rola-Noworyta Anna, Pazdro Krzysztof, Akademicki zbiór zadań z chemii ogólnej. Oficyna Wydawnicza Krzysztof Pazdro, 2015</p> <p>Okuniewski Andrzej, Chemia Ogólna i Nieorganiczna Ćwiczenia Rachunkowe. Wydawnictwo Politechniki Gdańskiej, 2021.</p> <p>Konieczka Piotr, Namieśnik Jacek, Ocena i kontrola jakości wyników pomiarów analitycznych. Wydawnictwo Naukowe PWN, Warszawa 2017.</p>
	Supplementary literature	<p>D. C. Montgomery, "Design and Analysis of Experiment", John Wiley & Sons Inc., 2013</p> <p>Krawczyk Krzysztof, Petryk Jan, Schmidt-Szałowski Krzysztof, Technologia chemiczna. Ćwiczenia rachunkowe. Wydawnictwo Naukowe PWN, Warszawa 2021</p> <p>Grzywa Edward, Molenda Jacek, Technologia podstawowych syntez organicznych (t.1 oraz t.2). Wydawnictwa Naukowo-Techniczne, Warszawa, 2016.</p> <p>Jess Andreas, Wasserscheid Peter, Chemical Technology - from principles to products. Wiley-VCH Verlag GmbH, 2020.</p>
	eResources addresses	
Example issues/ example questions/ tasks being completed	<p>Explain the relationship between the properties of chemical elements and their position in the periodic table of elements (periodicity law)</p> <p>Give examples of elements and their industrial technologies related to the development of information technology</p> <p>Calculate the pH of an aqueous solution (strong and weak acids and bases)</p> <p>Planning an experiment: stages and their verification</p>	
Practical activities within the subject	Not applicable	

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