GDAŃSK UNIVERSITY OF TECHNOLOGY

Subject card

Subject name and code	Thermodynamics II, PG_00040185						
Field of study	Mechanical Engineering						
Date of commencement of studies	October 2026		Academic year of realisation of subject		2027/2028		
Education level	first-cycle studies		Subject group		Obligatory subject group in the field of study Subject group related to scientific research in the field of study		
Mode of study	Full-time studies		Mode of delivery		at the university		
Year of study	2		Language of instruction		English		
Semester of study	4		ECTS credits		3.0		
Learning profile	general academic profile		Assessment form		assessment		
Conducting unit	Department of Energy and Industrial Apparatus -> Faculty of Mechanical Engineering and Ship Technology -> Faculties of Gdańsk University of Technology						
Name and surname of lecturer (lecturers)	Subject supervisor		prof. dr hab. inż. Dariusz Mikielewicz				
	Teachers						

Lesson types

Lesson type	Lecture	Tutorial	Laboratory	Project	Seminar	SUM
Number of study hours	15.0	0.0	15.0	0.0	0.0	30

E-learning hours included: 0.0

Learning activity and number of study hours

Learning activity	Participation in didactic classes included in study plan	Participation in consultation hours	Self-study	SUM
Number of study hours	30	6.0	39.0	75

Subject objectives

Familiarisation with advanced topics of thermodynamics

Learning outcomes

Course outcome	Subject outcome	Method of verification
K6_U06	Knows the mechanisms of combustion, condensation, moisture migration, basics of heat exchangers	[SU3] Assessment of ability to use knowledge gained from the subject
K6_W09	Knows the mechanisms of combustion, condensation, moisture migration, basics of heat exchangers	[SW1] Assessment of factual knowledge

Subject contents

Course content – lecture
LECTURE: Gas mixtures and moist gases. Mollier diagram and the basic moist air processes. Maxwell"s thermodynamic equations. Elements of combustion thermodynamics. Fundamentals of refrigeration. Fundamentals of heat transfer.
LABORATORIES: Gas analysis. Determination of calorific value of solid fuels and gases. The energy balance of the water boiler and heat exchanger (recuperator). Testing of the refrigerating unit. Testing of the air conditioning central unit. Testing of the fan.

Prerequisites and co-requisites

Thermodynamics 1

Assessment methods and criteria

Subject passing criteria	Passing threshold	Percentage of the final grade
exam	56.0%	67.0%
laboratory	56.0%	33.0%

Recommended reading

Basic literature	1. M.J. Moran, H.N. Shapiro, D.D. Boettner, M.B. Bailey, Fundamentals of Engineering Thermodynamics 8th Ed., Wiley, 2014 2. Y. Cengel, M. Boles, Thermodynamics An Engineering Approach, 8th Edition, Wiley, 2014
Supplementary literature	Any textbook in thermodynamics
eResources addresses	

Example issues/ example questions/ tasks being completed	1. Present and discuss known mechanisms of heat transfer on the example of overall heat transfer through a multilayer wall separating two fluids with different temperatures. 2. Define the thermal resistance due to conduction, convection and overall heat transfer. 3. Discuss how to include the effect of fouling on overall thermal resistance. 4. Definition of logarithmic mean temperature difference and temperature distribution in the parallel and counter-current heat exchangers. 5. Define specific humidity and relative humidity. What is a difference? 6. What is saturation temperature? 7. Construct sample of psychrometric chart. What the lines represent? 8. Describe graphically on a psychrometric chart all changes in the properties of air 9. The dry-bulb and wet-bulb temperatures in a classroom are 24degC and 16 degC, respectively. Determine (at psychrometric chart) the humidity ratio, relative humidity and dew point at atmospheric pressure. 10. Construction of Psychrometric Chart 11. Design and operation of Linde-Hampson liquifier with representation of the process on a thermodynamic diagram. 12. Definition of inversion point and inversion curve. 13. What is the Joule-Thomson effect? The purpose and the coefficient of this effect. 14. Definition of combustion process 15. The stages of the solid fuel combustion 16. The main characteristics of the flames 17. Describe what is air excess number and how we can calculate it 18. What is the difference between adiabatic flame temperature and real flame temperature
Practical activites within the subject	Not applicable

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