



Subject card

Subject name and code	Environmental chemistry, PG_00057712						
Field of study	Green Technologies						
Date of commencement of studies	October 2026	Academic year of realisation of subject			2027/2028		
Education level	first-cycle studies	Subject group			Optional subject group Subject group related to scientific research in the field of study		
Mode of study	Full-time studies	Mode of delivery			at the university		
Year of study	2	Language of instruction			Polish		
Semester of study	3	ECTS credits			5.0		
Learning profile	general academic profile	Assessment form			exam		
Conducting unit	Department of Inorganic Chemistry -> Faculty of Chemistry -> Faculties of Gdańsk University of Technology						
Name and surname of lecturer (lecturers)	Subject supervisor	dr hab. Katarzyna Kazimierczuk					
	Teachers						
Lesson types	Lesson type	Lecture	Tutorial	Laboratory	Project	Seminar	SUM
	Number of study hours	30.0	0.0	30.0	0.0	0.0	60
	E-learning hours included: 0.0						
Learning activity and number of study hours	Learning activity	Participation in didactic classes included in study plan		Participation in consultation hours		Self-study	SUM
	Number of study hours	60		10.0		55.0	125
Subject objectives	Familiarize students with the basics of chemical processes occurring in the natural environment, physical chemistry of the atmosphere, water and soil. Presentation of geochemical cycles of the most important elements in the environment. Familiarization with the most important environmental pollutants, their sources and methods of detection.						

Learning outcomes	Course outcome	Subject outcome	Method of verification
	[K6_U04] capable of formulating and solving design tasks in the field of environmental technology to recognize their non-technical aspects, including environmental, economic and legal. Is capable of applying the principles of occupational health and safety. Is able to make initial assessment of engineering solutions and actions	The student notices nontechnical, including environmental, aspects of technologies used in environmental protection. Applies the principles of occupational health and safety.	[SU2] Assessment of ability to analyse information [SU4] Assessment of ability to use methods and tools
	[K6_W03] has a basic knowledge of soil, air and water pollutants, design and supervision of environmentally friendly technologies and technologies which do not produce waste, knows technology of cleaning and neutralization of industrial waste and wastewater management, has a basic understanding of the theoretical basis of methods and types of apparatus used in chemical analysis of environmental pollutants	The student has basic knowledge in the field of soil, air and water protection against pollution and the theoretical basis of methods and types of apparatus used in the analysis of environmental pollution.	[SW1] Assessment of factual knowledge
	[K6_W02] has a basic knowledge of chemistry including general chemistry, inorganic, organic, physical, analytical, including the knowledge necessary to describe and understand the phenomena and chemical processes occurring in the environment; measurement and the determination of the parameters of these processes.	The student has basic knowledge in the field of chemistry necessary to describe and understand phenomena and chemical processes occurring in the natural environment. Knows the basics of the methods used for measuring the level of environmental pollution.	[SW1] Assessment of factual knowledge
Subject contents	Course content – lecture Atmospheric chemistry. Aquatic chemistry. Soil chemistry. Persistent organic pollutants in the environment. Carbon cycle. Nitrogen cycle. Phosphorus cycle. Oxygen and sulfur cycle. The role of the chemical elements in living organisms. Heavy metals and micronutrients. Environmental analytics. Methods of measuring the degree of pollution. Remote pollution measurement methods.		
Prerequisites and co-requisites	Passed course of Inorganic Chemistry		
Assessment methods and criteria	Subject passing criteria	Passing threshold	Percentage of the final grade
	written exam	60.0%	70.0%
	laboratory reports	60.0%	30.0%
Recommended reading	Basic literature	Gary W van Loon and Stephen J Duffy, Environmental Chemistry, Oxford University Press	
	Supplementary literature	S. Manahan, Environmental Chemistry, CRC Press, 2009	
	eResources addresses		
Example issues/ example questions/ tasks being completed	<ul style="list-style-type: none"> • Characterize photochemical smog. • List alternative sources of phosphorus. • Discuss the carbon cycle in nature. 		
Practical activities within the subject	Not applicable		

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