



## Subject card

Subject name and code	Corrosion measurement, PG_00060767						
Field of study	Chemical Technology						
Date of commencement of studies	October 2026	Academic year of realisation of subject				2029/2030	
Education level	first-cycle studies	Subject group				Optional subject group Subject group related to scientific research in the field of study	
Mode of study	Full-time studies	Mode of delivery				at the university	
Year of study	4	Language of instruction				Polish	
Semester of study	7	ECTS credits				2.0	
Learning profile	general academic profile	Assessment form				assessment	
Conducting unit	Department of Corrosion and Electrochemistry -> Faculty of Chemistry -> Faculties of Gdańsk University of Technology						
Name and surname of lecturer (lecturers)	Subject supervisor	dr hab. inż. Paweł Ślepski					
	Teachers						
Lesson types	Lesson type	Lecture	Tutorial	Laboratory	Project	Seminar	SUM
	Number of study hours	15.0	0.0	15.0	0.0	0.0	30
	E-learning hours included: 0.0						
Learning activity and number of study hours	Learning activity	Participation in didactic classes included in study plan		Participation in consultation hours		Self-study	SUM
	Number of study hours	30		2.0		18.0	50
Subject objectives	The aim of this course is to introduce students to the topic of corrosion and familiarize them with the principles and methods of measuring corrosion processes in various environments, as well as with techniques for assessing the technical condition of materials and structures exposed to corrosion. Students will acquire the ability to select appropriate testing methods (electrochemical, gravimetric, potentiometric, etc.), interpret measurement results, and formulate conclusions necessary for designing anticorrosion protection and assessing the operational durability of equipment.						

Learning outcomes	Course outcome	Subject outcome	Method of verification
	[K6_W07] Has knowledge of raw materials and technologies in the chemical and polymer industries, also covering issues of corrosion and material protection.	has broad and structured knowledge of raw materials used in the chemical and polymer industries, knows their properties, applications and processing methods, and understands the mechanisms of corrosion, its effects on various materials and available techniques and technologies for corrosion protection used in modern industrial processes.	[SW1] Assessment of factual knowledge [SW2] Assessment of knowledge contained in presentation
	[K6_W05] Has knowledge of electrical engineering, automation and computer science, including the operation of measurement and control systems	has broad, structured knowledge in the field of electrical engineering, automation and computer science, covering both the principles of operation and cooperation of elements of measurement and control systems, as well as the structure, functions and method of data processing in modern process control and supervision systems.	[SW1] Assessment of factual knowledge [SW2] Assessment of knowledge contained in presentation
	[K6_U08] Is able to select elements of automatic control systems for simple technological processes and use computer programs to control and optimize chemical processes.	is able to select control system components for a simple technological process and use software to simulate, control and optimize chemical processes.	[SU4] Assessment of ability to use methods and tools
	[K6_K01] Is aware of the social role of a technical university graduate and understands the need to provide information about technical achievements and engineering activities to society, including through the media.	is aware of the social role of an engineer, understands the importance of responsible transfer of knowledge about technical achievements and engineering activities, is able to communicate them in a way that is understandable to various recipients, including through the media, and actively promotes technical knowledge and innovation in social life.	[SK1] Assessment of group work skills [SK5] Assessment of ability to solve problems that arise in practice [SK4] Assessment of communication skills, including language correctness
Subject contents	<p>Course content – lecture</p> <p>Basic concepts of corrosion and corrosion protection. Chemical and electrochemical corrosion. Environmental factors affecting corrosion intensity. Examples of corrosion in industrial installations. Principles of anticorrosion protection. Types of corrosion protective coatings. Methods for testing coating quality.</p> <p>Course content – laboratory</p> <p>Determining Corrosion Rate and Corrosion Potential Gravimetric Techniques Electrochemical Methods Measuring Protective Coating Thickness Basic LabView Programming</p>		
Prerequisites and co-requisites	Students should have a basic understanding of general and physical chemistry, particularly electrochemical processes, as well as the fundamentals of materials science and materials mechanics. Knowledge of basic concepts in thermodynamics and chemical kinetics, as well as the ability to perform simple physicochemical calculations, is recommended. Students should be familiar with basic laboratory equipment and be familiar with laboratory safety regulations.		
Assessment methods and criteria	Subject passing criteria	Passing threshold	Percentage of the final grade
	Laboratory (reports)	51.0%	30.0%
	Lecture (exam)	51.0%	70.0%
Recommended reading	Basic literature	Technika przeciwkorozyjna, praca zbiorowa pod redakcją Romulada Juchniewicza, PWSZ 1973 Techniki przeciwkorozyjne, Józef Mikula, Politechnika Rzeszowska 1992 Podstawy teoretyczne i aspekty praktyczne zjawiska korozji, T. Hryniewicz, K. Rokosz, Politechnika Koszalińska 2010	
	Supplementary literature	Eksperymentalna chemia fizyczna, praca zbiorowa, PWN 1982	
	eResources addresses		
Example issues/ example questions/ tasks being completed	Types and mechanisms of corrosion: chemical, electrochemical, galvanic, crevice, and stress corrosion. Thermodynamic and kinetic determinants of corrosion processes. Electrochemical potential, corrosion current, and polarization resistance: their significance and interpretation.		
Practical activities within the subject	Not applicable		

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